

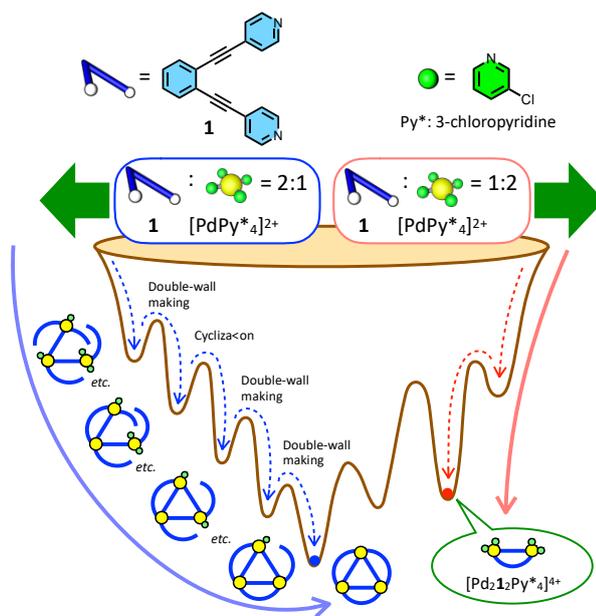
Reaction-network-based analysis and prediction on the coordination self-assembly of Pd₃L₆ double-walled triangle

(¹Department of Basic Science, The University of Tokyo, Japan, ²Department of Molecular Engineering, Kyoto University, Japan, ³ESICB, Kyoto University, Japan, ⁴Fukui Institute for Fundamental Chemistry, Kyoto University, Japan)

○Satoshi Takahashi,¹ Tomoki Tateishi,¹ Yuya Sasaki,¹ Hirofumi Sato,^{2,3,4} Shuichi Hiraoka¹

Keywords: Molecular self-assembly; Chemical master equation; Kinetic control

The self-assembly process of the [Pd₃1₆]⁶⁺ double-walled triangle (DWT)¹ was analyzed by our developed numerical analysis of self-assembly system (NASAP), which has already succeeded in obtaining detailed information of several coordination self-assemblies²⁻⁶. The self-assembly of the DWT, which takes place in a complicated reaction network consisting of multiple inter- and intramolecular reactions (oligomerization, cyclization, and double-wall making), is a good example not only to reveal general principles underlying complicated molecular self-assembly but also to demonstrate the power of numerical simulation to predict a suitable reaction condition under which a new molecular self-assembly is produced under kinetic control. The reaction network for the DWT composed of 161 chemical species that are connected by total 896 elementary reactions classified into 9 types of reversible reactions was established by the numerical search of 18-dimensional parameter space so that the time-development of the species would reproduce the corresponding experimental data obtained by quantitative analysis of self-assembly system (QASAP)⁷. NASAP led to the dominant reaction pathways, which are determined by the balance between inter- and intramolecular reactions. Besides, the numerical simulation predicted the formation of [Pd₂1₂Py*₄]⁴⁺ as a major product (#%) from **1** and [PdPy*₄]²⁺ in a 1:2 ratio under kinetic control, which was consistent with the experimental result (#%).



1) S. Takahashi *et al.*, *Phys. Chem. Chem. Phys.* **2020**, *22*, 26614. 2) Y. Matsumura *et al.*, *Phys. Chem. Chem. Phys.* **2017**, *19*, 20338. 3) S. Takahashi *et al.*, *Phys. Chem. Chem. Phys.* **2019**, *21*, 6341. 4) S. Komine *et al.*, *J. Am. Chem. Soc.* **2019**, *141*, 3178. 5) L. H. Foianesi-Takeshige *et al.*, *Commun. Chem.* **2019**, *2*, 128. 6) T. Tateishi *et al.*, *J. Am. Chem. Soc.* **2019**, *141*, 19669. 7) Reviews: a) S. Hiraoka, *Bull. Chem. Soc. Jpn.* **2018**, *91*, 957. b) S. Hiraoka, *Isr. J. Chem.* **2019**, *59*, 151. c) S. Hiraoka *et al.*, *Chem. Rec.* **2021**, *21*, DOI: 10.1002/tcr.202000124.