

Large latent heat and barocaloric effects at charge transfer transitions of A-site ordered perovskite oxides

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An A-site ordered perovskite oxide NdCu₃Fe₄O₁₂ containing unusual high valence Fe ions shows an intersite charge transfer transition between Cu at the A' site and Fe at the B site represented by $3\text{Cu}^{2+} + 4\text{Fe}^{3.75+} \rightarrow 3\text{Cu}^{3+} + 4\text{Fe}^{3+}$ near room temperature.^{1,2} The transition is accompanied by a large volume change as well as metal-to-insulator and paramagnetic-to-antiferromagnetic transitions (Figure 1). We found that the present NdCu₃Fe₄O₁₂ released large latent heat 25.5 kJ/kg and the corresponding entropy change 84.2 J/K·kg by the intersite charge transfer transition (Figure 2). We also demonstrated that the latent heat could be utilized through a barocaloric effect by applying pressure.³

The observed large entropy change is contributed by the lattice and spin degrees of freedom. The large negative-thermal-expansion-like volume change and the unusual first-order magnetic transition at the charge transition temperature should play important roles in giving rise to the significant entropy change.

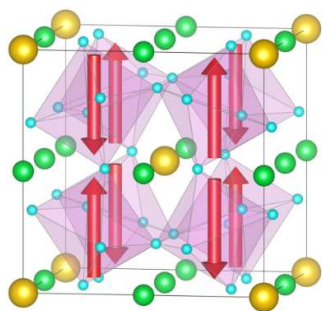


Figure 1 Crystal and magnetic structure of NdCu₃Fe₄O₁₂.

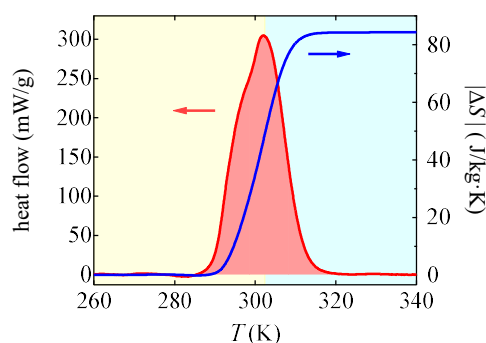


Figure 2 Heat flow and the corresponding entropy change of NdCu₃Fe₄O₁₂.

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