

## Direct observation of H<sub>2</sub> dissociation and spillover process on Pd/Cu single atom alloy catalyst surface by spectroscopic method

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The dissociation of hydrogen molecules on the catalyst surface is an important elementary reaction in various industrial processes. For example, Cu-based catalysts in methanol synthesis reactions show low H<sub>2</sub> dissociation activity, and thus a high temperature is required. However, it is unfavorable for exothermic methanol generation from CO<sub>2</sub> and H<sub>2</sub>. More active catalysts for H<sub>2</sub> dissociation are indispensable for efficient hydrogenation reactions. For this purpose, Pd/Cu single atom alloy catalyst (SAAC) is one of the attractive solution, in which a few percent of Cu surface atoms are replaced with Pd atoms. It has been reported that H<sub>2</sub> is dissociated on the Pd/Cu SAAC surface even at low temperatures, and that H atoms spillover onto the Cu host surface<sup>1</sup>. In this study, the processes of hydrogen dissociation and spillover on Pd/Cu SAAC surface was investigated using high-resolution X-ray photoelectron spectroscopy (XPS) and infrared reflection absorption spectroscopy (IRAS).

For XPS experiments, the Pd/Cu(977) SAAC was prepared in which the coverage of Pd( $\theta_{\text{Pd}}$ ) was 0.07 ML. The peak of Pd3d<sub>5/2</sub> was observed at 335.3 eV on the clean SAAC surface. As a function of exposure to H<sub>2</sub> at 80 K, the Pd3d<sub>5/2</sub> peak was shifted to 335.6 eV and 336.5 eV in two steps (Fig. 1 (a)).

For IRAS experiments, the ML Pd/Cu(111) SAAC was prepared ( $\theta_{\text{Pd}}$  = 0.01 ML). Time-resolved IRAS measurements were performed during H<sub>2</sub> exposure at 83 K. The H-Cu vibrational peak appeared at 1143 cm<sup>-1</sup> about 300 s after starting H<sub>2</sub> exposure<sup>2</sup>, and it developed with elapsed time (Fig. 1 (b)).

These results directly indicated the processes of hydrogen dissociation, adsorption and spillover on the Pd/Cu SAAC surface. The Pd atoms act as a site for H<sub>2</sub> dissociation and initial H adsorption, and the spillover to Cu sites occurs only after Pd sites are saturated with H atoms.

1) G. Kyriakou *et al*, *Science* **2012**, 335, 1209. 2) K. Mudiyansele *et al*, *J. Chem. Phys.* **2013**, 139, 044712.

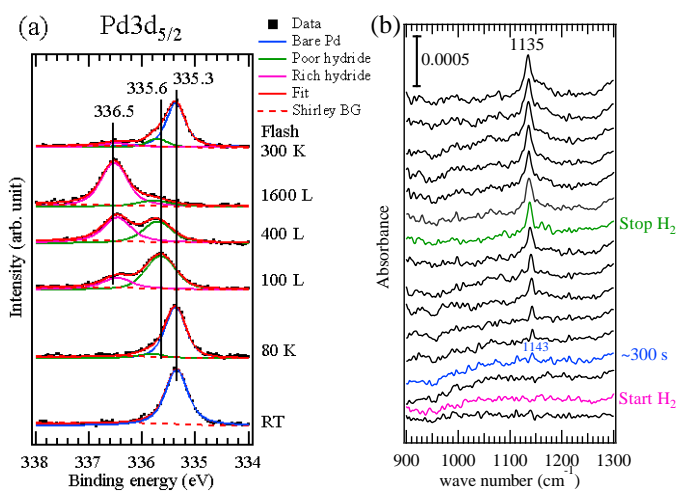


Fig. 1 Results of (a) XPS,  $h\nu = 680$  eV. (b) IRAS, 4 cm<sup>-1</sup> of resolution, 500 scans.