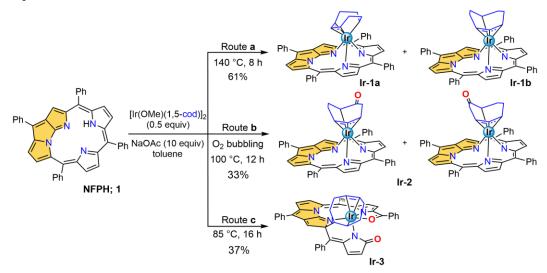
Synthesis and Characterization of N-Fused Porphyrin Iridium Complexes

(*Graduate School Engineering, Kyushu University*) OJibin Alex Abraham, Masatoshi Ishida, Hiroyuki Furuta

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N-fused porphyrin (NFpH) is a class of porphyrin analogs containing an inner fused tripentacyclic pyrrole unit, which serves as a unique monovalent N3 tridentate ligand (Scheme 1).¹ To date, several sitting atop metal complexes of NFp with ruthenium(II), manganese(I), iron(II), ruthenium(II), and tungsten(VI) ions at the inner core have been reported.² In this work, we have investigated the iridium(I) (1,5-cyclooctadiene: cod) complexation of NFp to understand the metal effect on the coordination mode, reactivity, and optical properties.

Complexation with an $[Ir(I)(cod)(OMe)]_2$ afforded a sitting atop iridium complex (Ir-1) supported by a transformed η^3 -allylic and σ -bonded cod ligand (Scheme 1). The resulting complex has two independent ancillary cod rotational isomers (i.e., Ir-1a and Ir-1b), and under aerobic conditions, a methylene-oxygenated cod analog (Ir-2) was formed. During the study, a unique octahedral Ir complex possessing a cod-fused and ring-broken dioxo-analog of NFp (Ir-3) was obtain. We have further studied their optical and electrochemical properties of these Ir complexes.



Scheme 1. Synthesis of N-Fused Porphyrin Iridium Complexes

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