In-situ observation of N₂ and H₂ adsorbed on Ru/MgO catalyst using modulation excitation-infrared spectroscopy

(¹WPI-I²CNER, Kyushu University, ²Graduate School of Advanced Science and Engineering, Hiroshima University, ³WPI-AIMR, Tohoku University, ⁴IMCE, Kyushu University) OTomohiro Noguchi¹, David S. Rivera Rocabado², Takayoshi Ishimoto², Miho Yamauchi^{1,3,4} **Keywords**: Ammonia synthesis catalysts; Infrared spectroscopy; Adsorption/desorption; Ru catalysts

Ru catalysts were reported to exhibit high catalytic activity under milder conditions than Fe-based catalysts used in Haber-Bosch process.¹ This is because both adsorption of N_2 molecules and desorption of products on Ru surfaces proceed smoothly due to the moderate binding energy between Ru and N species.² On the other hand, the strong binding energy between Ru and H species causes hydrogen poisoning, which diminishes the catalytic activity under high pressure conditions. To extend availability of Ru catalysts, the adsorption behavior of N_2 and H_2 under reaction conditions should be clarified. In this study, we examine the adsorption behavior of reactants on Ru nanoparticle catalysts surface under N_2 or H_2 atmosphere by employing modulation-excitation infrared spectroscopy (MEIRS), which enables highly sensitive and time-course observation of their adsorption states.

We prepared MgO supported Ru nanoparticle catalyst with 4-5wt% of Ru loading (Ru/MgO) by impregnation, thermal decomposition and hydrogen reduction according to the report.³ Fig. 1 shows the time-domain spectrum of MEIRS while alternately introducing N₂ and Ar at 250 °C under 0.1 MPa. The absorbance increased with time to exposure to N₂ and decreased after introduction of Ar. We observed two sharp peaks at 2071 and 1932 cm⁻¹ and broad peaks at 1631, 1427, 1211 and 980 cm⁻¹. From the density functional theory (DFT)

calculations, the sharp and broad peaks were assigned to vertical and horizontal adsorption of N_2 on Ru nanoparticles, respectively.⁴ In addition, each peaks were attributed to the adsorptions on different sites, which hold N_2 by different number of Ru atoms. The newly observed adsorption states of N_2 are possibly ascribed to hidden intermediate species formed on Ru/MgO and this finding would contribute to the elucidation of the



Fig. 1. Time-domain spectrum of MEIRS for Ru/MgO measured in the modulation of N $_2$ and Ar at 250 °C under 0.1 MPa.

reaction mechanism in catalytic ammonia synthesis.

K. Aika et al., J. Catal. 1972, 27, 424. 2) A. Vojvodic et al., Chem. Phys. Lett. 2014, 598, 108. 3) K.
K. Ghuman et al., Phys. Chem. Chem. Phys., 2019, 21, 5117. 4) D. R. Rocabado et al., ACS Nano, 2021, 15, 20079.