## Thermal degradation of a poly (ethylene-co-methacrylic acid) ionomer investigated by solid-state <sup>13</sup>C NMR and ESR

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Poly (ethylene-*co*-methacrylic acid) (EMAA) ionomers, which carboxy groups are neutralized with Na<sup>+</sup> cations, are well known to show the excellent physical and chemical properties, such as strong toughness and good fabricability. EMAA ionomers have been used as engineering and commercial materials for a long time and their structure and dynamics have been investigated by NMR and ESR measurements. It is also very important and interesting how the thermal degradation is affected by the existence of the ionic clusters. In this study, we examined EMAA and EMAA ionomers (EMAA-30Na and -54Na: the number represents the degree of neutralization) heat-treated at 200°C to reveal the structural change occurring during the thermal degradation using the solid-state <sup>13</sup>C NMR and ESR.

The radicals created after heat treatment were measured as a spin quantity with heating period by ESR (Fig. 1). It is obvious that the amount of radicals increases in two stages, and the quantity for EMAA-54Na became about twice and seven folds as compared to that for EMAA-30Na and for EMAA, respectively at 3000 min. However, the newly observed NMR peaks, which are attributed to the cross-linking structures produced with the radical residues, were much intense for EMAA rather than those for EMAA ionomers. The main product was a ~CH<sub>2</sub>-COO-CH<sub>2</sub>~ bond created from -COOH and ·OCH<sub>2</sub>- functional groups which are resonated around 175 ppm (Fig. 2). The other reaction processes during thermal degradation will be discussed in detail.

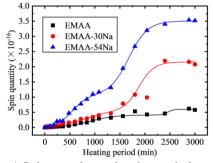


Fig. 1 Spin quantity vs. heating period.

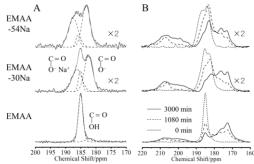


Fig. 2 Expanded <sup>13</sup>C CPMAS NMR spectra for C=O region; (A) before and (B) after heat treatment.

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