

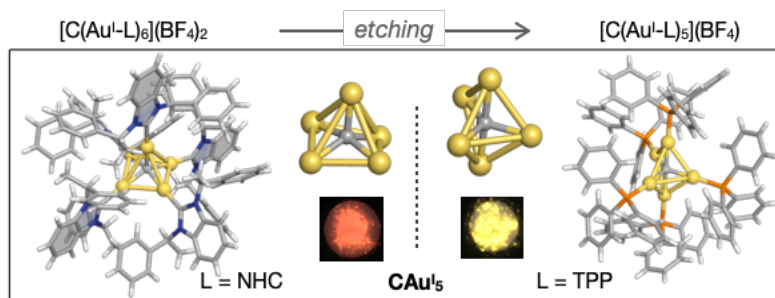
Post-Synthesis from Carbon-Centered Hexagold(I) Clusters to Pentagold(I) Clusters with Controlled Geometry and Luminescence Behaviors

(¹Graduate School of Science, The University of Tokyo) ○Xiao-Li Pei,¹ Hitoshi Ube,¹ Zhen Lei,¹ Mitsuhiro Shionoya¹

Keywords: Gold(I) Cluster; Hypercoordinated Carbon-Center; *N*-Heterocyclic Carbene Ligand; Luminescence; Structure-Property Relationship

In recent years, there has been a growing interest in the field of gold chemistry to tune the size and structure of gold clusters at the atomic level.¹ Post-synthetic reactions such as etching/degradation reactions using free ligands to obtain smaller gold nanoclusters have made significant progress,¹ but the development of the underlying chemistry has only just begun.² Our carbon-centered hexagold(I) (CAu^I₆) clusters with *N*-heterocyclic carbene (NHC) ligands³ are promising as starting materials for the synthesis of smaller CAu^I_{*n*} (*n* < 6) clusters, which are difficult to synthesize directly. Recently, we have reported the synthesis of [(C)(Au^I-NHC)₆](BF₄)₂ and the ligand effects on their chemical and physical properties.³ With the CAu^I₆ clusters in hand, we set out to precisely tune the smaller carbon-centered gold clusters by an etching process *via* degradation with the addition of phosphine ligands.

By adding two equivalents of 1,2-bis(diphenylphosphanyl)benzene to a solution of [(C)(Au^I-L)₆](BF₄)₂ (L = NHC, TPP: triphenylphosphine) in CH₂Cl₂, a series of pentagold(I) clusters [(C)(Au^I-L)₅](BF₄) were successfully obtained in 79%-93% yield. To our best knowledge, this work is the first example of phosphine-etching in NHC-protected gold(I) clusters. Single-crystal X-ray analyses revealed that the NHC and TPP ligands ensure the distorted trigonal-bipyramidal and square-pyramidal geometries of the pentagold(I) motifs, respectively. The NHC-CAu^I₅ cluster showed an orange-red luminescence in the solid state, while the TPP-CAu^I₅ cluster showed a bright yellow luminescence. This structure-dependent luminescence behaviors is a noteworthy result.



1) M. Zhu et al., *Chem. Mater.* **2019**, 31, 9939. 2) J. Xie et al., *Nat. Commun.* **2021**, 12, 3212. 3) a) M. Shionoya et al., *Organometallics* **2018**, 37, 2007. b) M. Shionoya et al., *J. Organomet. Chem.* **2020**, 917, 121271. c) M. Shionoya et al., **2021**, submitted.