

A Polyaromatic Capsule Solid as a Vapor Adsorbent for Benzene Derivatives

(Lab. for Chem. & Life Sci., Tokyo Tech) Ryuki Sumida, Michito Yoshizawa

Keywords: Porous solid; Volatile compound; Adsorption; Polyaromatic capsule; Reusability

Porous materials, such as metal-organic frameworks (MOFs) and covalent organic frameworks (COFs), are useful solids for adsorption, separation, and storage of molecules.^[1] Most of these materials possess infinite cavities and windows in subnanometer to nanometer size. In contrast, polyaromatic capsule **1**, composed of eight anthracene panels and two metal ions, provides a finite, large cavity with four small windows.^[2] Here we report that capsule solid (**1**)_n displays adsorption ability for benzene derivatives as a new vapor adsorbent.

Capsule solid (**1**)_n was put in a sealed vessel including benzene (BN), toluene (TL), and *o*-xylene (oXL) without direct host-guest and guest-guest contact (Figure 1a). After standing at r.t. for 1 h, the resultant solid (**1**)_n•(BN)_x•(TL)_y•(oXL)_z was subjected to reduced pressure and dissolved in CD₃CN. ¹H NMR analysis of the product solution revealed that 3.3 equivalent of the benzene derivatives (based on **1**) are bound by solid (**1**)_n, with >80% oXL selectivity (Figure 1b, left). In the same way, solid (**1**)_n bound oXL from a mixture of three xylene isomers with 53% selectivity (Figure 1b, right). The volatilization of the adsorbed molecules was suppressed by solid (**1**)_n even at 100 °C for 1 h (Figure 1c). Furthermore, solid (**1**)_n could be reused at least five times with the same adsorption ability for three xylene isomers (Figure 1d).

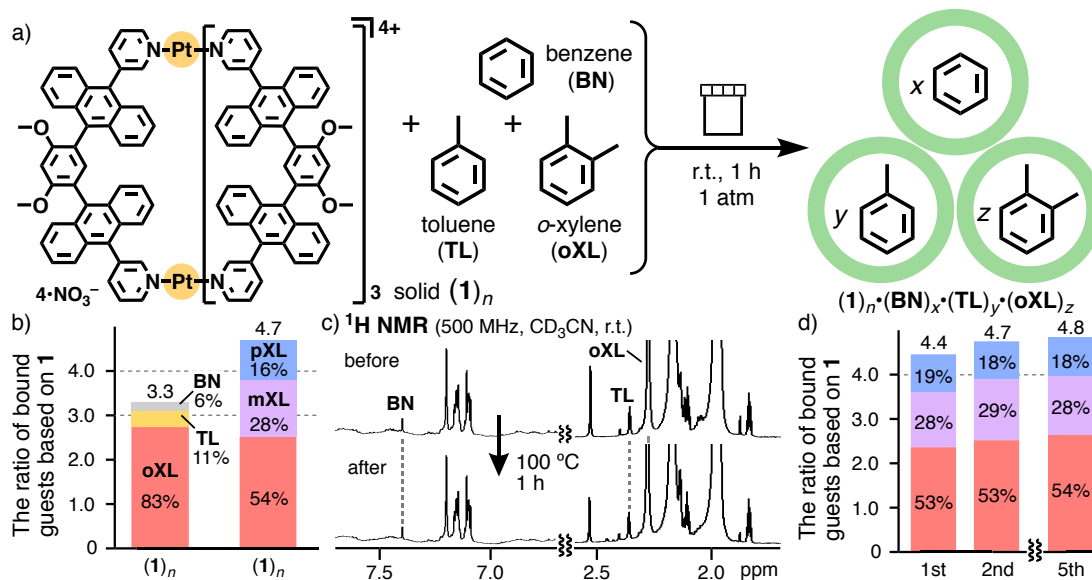


Figure 1. a) Selective vapor adsorption by solid (**1**)_n. b) Guest ratios and % after binding experiments. c) ¹H NMR spectra of (**1**)_n•(BN)_x•(TL)_y•(oXL)_z before/after heating. d) Reusability of solid (**1**)_n.

[1] G. Zhang, B. Hua, A. Dey, M. Ghosh, B. A. Moosa, N. M. Khashab, *Acc. Chem. Res.* **2021**, *54*, 155–168. [2] N. Kishi, Z. Li, K. Yoza, M. Akita, M. Yoshizawa, *J. Am. Chem. Soc.* **2011**, *133*, 11438–11441.