Cu^{II}-dependent regulation of a split-DNAzyme having consecutive ethenoadenine nucleobases as metal recognition sites

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Metal-mediated base pairs consisting of opposing ligand-type nucleobases and a bridging metal ion are often applied towards the development of stimuli-responsive DNA architectures. Previously we have developed a Cu^{II} -dependent allosteric DNAzyme by incorporating artificial ligand-type nucleobases such as hydroxypyridone (**H**).^[1] However, the cumbersome synthesis and low yields associated with artificial nucleosides make the development of metal-responsive DNA systems especially challenging.

In this study, we employed an easily synthesized damaged nucleobase, $1,N^{6}$ ethenoadenine (ϵA), as a ligand-type nucleobase for the development of a metal-responsive DNAzyme. It was reported that the formation of a single ϵA -Cu^{II}- ϵA base pair stabilizes the DNA duplex ($\Delta T_m = +3 \ ^{\circ}C$),^[2] but such a low stabilization effect is insufficient for the metalmediated regulation of DNA structures. We found that incorporation of three consecutive ϵA -Cu^{II}- ϵA base pairs into DNA duplexes allows for Cu^{II}-dependent significant duplex stabilization ($\Delta T_m = +11.5 \ ^{\circ}C$). Motivated by this result, we modified a reported RNA-cleaving DNAzyme (NaA43)^[3] by splitting it into two strands and incorporating three ϵA - ϵA mismatch pairs into the stem duplex (Figure). In the absence of Cu^{II} ions, the activity of the ϵA -modified DNAzyme was significantly suppressed. Upon addition of Cu^{II} ions (3 equiv.), the RNAcleaving activity was enhanced by 5.3-fold. The activity of neither the unmodified NaA43 DNAzyme nor a control DNAzyme having A-A mismatches was altered by adding Cu^{II} ions. These results indicate that the split ϵA -DNAzyme can hybridize via ϵA -Cu^{II}- ϵA base pairing to reconstruct the catalytic core, thereby enhancing the activity in response to Cu^{II} ions.^[4]

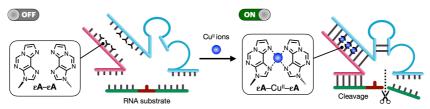


Figure. Schematic illustration of a Cu^{II}-dependent split DNAzyme regulated by the formation of consecutive ϵA -Cu^{II}- ϵA base pairs.

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