

Development of BaTaO₂N photocatalysts with distinctive carrier dynamics for one-step excitation overall water splitting

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BaTaO₂N (BTON) is an attractive perovskite oxynitride material and visible-light responsive photocatalyst, owing to its suitable band structures for hydrogen and oxygen evolution reactions. To date, several strategies have been employed to produce BTON with considerably less defects such as flux-assisted approach.¹⁻² However, the one-step excitation overall water splitting (OE-OWS) utilizing BTON as the photocatalysts remains elusive to achieve. Herein, we report BTON that demonstrates OE-OWS and unravel its characteristics suitable for simultaneous production of H₂ and O₂ in pure water condition and under visible light irradiation.

The BTON was synthesized using amorphous Ta₂O₅ as a Ta precursor and blended with BaCO₃ as a Ba precursor, followed by nitridation under NH₃ gas. With suitable hydrogen evolution reaction (HER) and oxygen evolution reaction (OER) cocatalysts, the BTON split water to H₂ and O₂ at the nearly stoichiometric ratio (2.3 and 1.1 μmol/h for H₂ and O₂, respectively), suggesting successful observation of OWS utilizing BTON as the photocatalysts. The carrier dynamics of BTON synthesized from amorphous Ta₂O₅ was investigated by transient absorption spectroscopy to reveal its distinctive characteristics compared to other BTONs nitrided using the commercially available Ta₂O₅ precursors and with the addition of fluxes. Under 470 nm laser pulse excitation, the mid-IR absorption spectra of photocarriers in BTON exhibited monotonic increase behavior towards lower probe energies, indicating notable free carrier absorption characteristics, as depicted in Figure 1. This absorption feature has not been observed previously for BTON synthesized using conventional Ta₂O₅ precursor.¹ Detailed results will be presented at the conference.

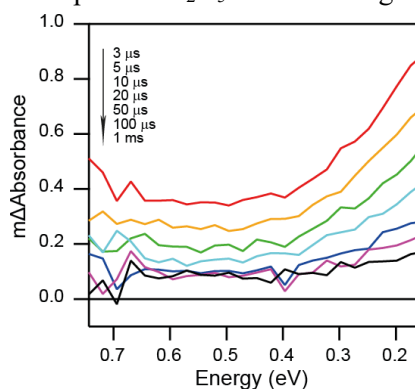


Figure 1. Transient absorption spectra of BTON BTON nitrided at 1173 K for 5 h under 470 nm laser pulses excitation. Measurement condition: room temperature and in N₂ (20 Torr).

References:

- 1) Z. Wang et al, Nature Communication, **12**, 1005 (2021),
- 2) S. Jadhav et al, *J. Mater. Chem. A* **8**, 1127 (2020).
- 3) M. Higashi et al, *Chem. Mater.* **21**, 1543 (2009).