Photo-generated extremely long-lived spin-polarized radicals in Metal-Organic Frameworks

(¹*Grad. Sci. Eng., Kyushu Univ.,* ²*CMS, Kyushu Univ.,* ³*FOREST, JST,* ⁴*RIKEN,* ⁵*Institute for Molecular Science.*) O Kana Orihashi,¹ Akio Yamauchi,¹ Saiya Fujiwara,⁴ Kenichiro Tateishi,⁴ Tomohiro Uesaka,⁴ Mizue Asada,⁵ Toshikazu Nakamura,⁵ Nobuo Kimizuka,^{1,2} Nobuhiro Yanai^{1,2,3}

Keywords: Polarized radical; Metal-organic framework; EPR; Photoexcited triplet

The generation and control of the electron spin polarization (ESP) is a crucial foundation not only for a wide range of quantum technologies but also for dynamic nuclear polarization (DNP), which is a technique to enhance the signal intensity of NMR and MRI. In addition to the generation of ESP, elongated spin-lattice relaxation times of the electron spin (T_{1e}) is desired for expanding the scope of future quantum technology and DNP. However, in conventional systems, T_{1e} has been limited to less than a few µs at room temperature. In this work, we have discovered photo-generated spin-polarized radicals with extremely long lifetime in acene-based metal-organic frameworks (MOFs).

To obtain large ESP at room temperature, we employed 5,12-diazatetracene (DAT) which can produce polarized triplets by photo-irradiation. We synthesized new ligand DPyDAT by modifying DAT with pyridine as a coordination site to the metal. Pillared-layer type of the MOF was synthesized with DPyDAT, Zn ion and carboxylic ligand. DAT moieties are aligned one-dimensionally with relatively close distance in the skeleton of the MOF. ESP formation in the MOF was evaluated by time-resolved ESR measurements at room temperature. The ESR spectrum of the MOF showed, in addition to the DAT triplet-derived pattern, a relatively sharp emissive peak of radical origin around 342 mT. Significantly, spin-lattice relaxation time T_{1e} was found to be as long as 275 μ s by pulsed ESR inversion recovery sequence at 342 mT at room temperature. We will also show the relationship between assembly structure of acenes and the T_{1e} value by comparing different MOF structures.

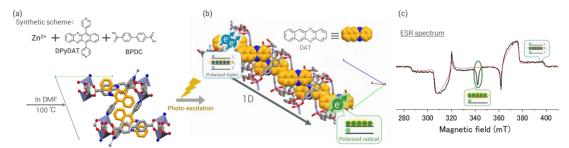


Fig. 1 (a) Synthetic scheme and structure of the MOF. (b) Schematic image of this work. (c) Tr-ESR spectrum of the MOF.

1) S. Fujiwara, N. Matsumoto, K. Nishimura, N. Kimizuka, K. Tateishi, T. Uesaka, N. Yanai, *Angew. Chem. Int. Ed.* **2022**, 61, e2021157.