## [2+2+1]-Cycloaddition Reaction of Azobenzene and Diarylacetylenes Catalyzed by Early Transition Metal Imido Complexes

(<sup>1</sup>Graduate School of Engineering Science, Osaka University, <sup>2</sup>University of Minnesota, <sup>3</sup>Graduate School of Pharmaceutical Science, Osaka University) ○Takuya Akiyama,<sup>1</sup> Yuya Kakiuchi,<sup>1</sup> Ian A. Tonks,<sup>2</sup> Hayato Tsurugi,<sup>1</sup> Kazushi Mashima<sup>3</sup>

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Reactivity of an early transition metal-nitrogen double bond involved in imidometal complexes with unsaturated organic compounds has been extensively investigated as a key step for synthesizing a variety of N-containing organic molecules. Catalytic [2+2+1]-cycloaddition reaction of the metal-nitrogen double bond with two equiv. of alkynes is a straightforward method in preparing synthetically valuable multi-substituted pyrrole derivatives. We previously reported that imido complexes of titanium and vanadium served as efficient catalysts to obtain alkyl-substituted pyrroles in one-pot manner; however, these catalysts were ineffective for incorporating diarylacetylenes.<sup>1</sup> Herein, we report that an in situ-generated bis(imido)tungsten species from WCl<sub>6</sub> and organosilicon reductant, Si-Me<sub>2</sub>-DHP, exhibits superior catalytic activity among group 4-6 transition metal imido complexes for the [2+2+1]-cycloaddition of diarylacetylene 1a with azobenzene (2), giving pyrrole **3a** in 90% yield (eq. 1). DFT calculations on the reaction sequences revealed that bis(imido)tungsten species possesses the lowest activation barrier for the rate-determining initial incorporation of 1a to one of two imido moieties of in situ-generated bis(imido)tungsten among the structurally related Mo, Nb, and Ta derivatives. In addition, low catalytic activity of bis(imido)vanadium species for the formation of 3a is ascribed to

the generation of energetically accessible and unreactive (a-diimine)vanadium species; fact. formation of the in  $\alpha$ -diimine 5a from the bis(imido)vanadium complex 4 and 1a was confirmed by the control experiment (eq. 2). Further details on the difference of W and Ti-catalyzed multiarylated pyrrole formation are disclosed in this presentation.



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