

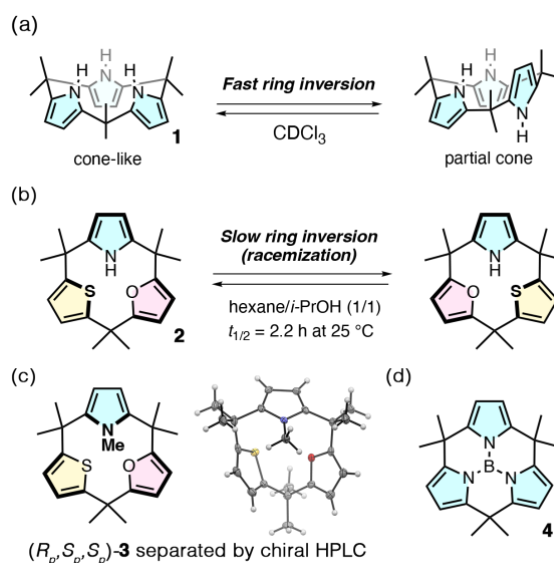
## Control of Aromatic Ring Inversion in Calix[3]pyrrole

(<sup>1</sup>Grad. Sch. Eng., Hokkaido Univ., <sup>2</sup>WPI-ICReDD, Hokkaido Univ.) ○Yuya Inaba,<sup>1</sup>Yu Kakibayashi,<sup>1</sup> Yasuhide Inokuma<sup>1,2</sup>**Keywords:** Calix[3]pyrrole; Ring inversion; Strained molecules; Boron complex; Planar chirality

Inversion of aromatic rings in macrocyclic compounds such as calixarenes or calixpyrroles can dynamically change their conformations and alter their reactivity, host-guest interactions, and chirality. In this work, we investigated aromatic ring inversion behaviors of calix[3]pyrrole<sup>1</sup>, a ring contracted analogue of calix[4]pyrrole, and its derivatives.

Pyrrole ring inversion in calix[3]pyrrole **1** was rapid in solution to be observed as a time averaged  $C_3$  symmetric molecule in <sup>1</sup>H NMR spectrum (**Fig. 1a**). Upon binding of fluoride anion, **1** adopts cone-like conformation to have hydrogen-bonds with fluoride efficiently. When two pyrrole rings of **1** were replaced with furan and thiophene rings to give **2**, its nonplanar conformations became chiral. Though two enantiomers of **2** were separated by chiral HPLC and exhibited mirror-image CD spectra, **2** underwent racemization via aromatic ring inversion in *n*-hexane/*i*-PrOH (v/v = 1/1) at room temperature with a half-life of 2.1 hours at 298 K (**Fig. 1b**). When **2** was methylated at NH site to give **3** in 50% yield, aromatic ring inversion was effectively suppressed. Enantiomers of **3** were successfully separated using chiral HPLC, and no racemization was observed for over a month at room temperature. The crystal structure of one of the enantiomers showed partial cone conformation that can be assigned as (*R<sub>p</sub>*, *S<sub>p</sub>*, *S<sub>p</sub>*) with respect to thiophene, furan, and pyrrole plane respectively (**Fig. 1c**). This difference in racemization behaviors between **2** and **3** were applied for chiral recognition during *N*-methylation using chiral ammonium salt.

Furthermore, when the ring inversion in **1** was inhibited in its boron complex **4** (**Fig. 1d**), a drastic change was observed in its reactivity towards acids. While **1** caused rapid ring cleavage upon treatment with TFA, boron complex **4** showed amphiphilic behavior with the macrocycle withstanding acidic conditions by reversible protonation at pyrrole ring.



**Fig. 1.** (a) Pyrrole ring inversion in **1**, (b) aromatic ring inversion-based racemization of **2**, (c) molecular and crystal structure of (*R<sub>p</sub>*, *S<sub>p</sub>*, *S<sub>p</sub>*)-**3** separated by chiral HPLC, and (d) molecular structure of **4**.

- 1) Y. Inaba, Y. Nomata, Y. Ide, J. Pirillo, Y. Hijikata, T. Yoneda, A. Osuka, J. L. Sessler, Y. Inokuma, *J. Am. Chem. Soc.*, **2021**, *143*, 12355–12360.