## Synthesis and chiral separation of bis (bipyridyl) manganese-

## tartrate complex

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## [Introduction]

Even metal complexes with no asymmetric carbon can have chirality by limiting the coordination direction of ligands. In this study, cis-bis-bipyridylmanganese(II) chloride (1) was selectively prepared, then, chiral separation of (+)- and (-)-bis-bipyridylmanganese(II) tartrate complexes was accomplished through the reaction of (1) with (+)-disodium tartrate. Structural analysis of the obtained complex was carried out by e.g. powder X-ray diffraction measurement. Chiral separation of organic reagent using the synthesized complex is also tested.

## [Experiments and results]

substance is in progress.

cis-[Mn(bpy)<sub>2</sub>]Cl<sub>2</sub>: After mixing MeOH solutions of MnCl<sub>2</sub>.H<sub>2</sub>O and 2,2'-bpy in molar ratio of 1:2, the solvent was slowly evaporated by letting solution for several days. By recrystallizing the precipitates by H<sub>2</sub>O/MeOH (1:1 v/v) mixed solvent, yellowish columnar crystals of cis-[Mn(bpy)<sub>2</sub>]Cl<sub>2</sub>(1) were obtained.(yield:43%)

(+)-[Mn(bpy)<sub>2</sub>(tart)]: aqueous solutions of (1) and (+)-Na<sub>2</sub>(tart).H<sub>2</sub>O were mixed in molar ratio of 1:2. Pale yellow needle-like crystals were filterd after standing for overnight. The presence of tartrate was confirmed by FT-IR spectrum, and the water molecule content was determined by TGA. The chemical

fig. chemical structure of (+)-[Mn(bpy)(+)-tart]

formula should be (+)-[Mn(bpy)<sub>2</sub>(tart)][Mn(tart)].10H<sub>2</sub>O(2). Powder X-ray diffraction study revealed that the space group of the crystal is  $P \ n \ a \ 2_1$ , with no inversion symmetry, in accordance with chiral material. Specific rotation measurement gave the value of +24.9, which means that this compound is (+)- enantiomer.

- (-)-[Mn(bpy)<sub>2</sub>(tart)]: To the filtrate, excess amount of 30 wt% NaI solution was added, then brownish-yellow crystals of (-)-[Mn(bpy)<sub>2</sub>(tart)](3) appeared by cooling the solution down to 0°C. After filtration, the substance was washed by EtOH and dried in air. (yield:11%) chiral separation of (+/-)-alanine: To the aqueous solution of (1), (+/-)-alanine solved in water was added. After night, pale yellow precipitate was obtained. Chemical analysis of this
- 1) Y. Kobayashi,2021 Synthesis and structural analysis of Mn(II) complex with 2,2'-bipyridine(graduation thesis).