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Distribution of 129 I in the environment released from the FDNPP accident and estimation of 131 I/ 129 I ratio

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Radioiodine is one of the most important radionuclides released from the Fukushima-Daiichi Nuclear Power Plant (FDNPP) accident. 131 I (half-life: 8 d) has a short half life time. Because of the difficulty of measuring 131 I at this time, it is expected to estimate 131 I precipitation from 129 I (half-life: 1.57×10^7 y) with the long half-life in the surface soil. We have measured 129 I concentrations in the surface soil at Fukushima. 129 I/ 127 I ratios were measured by accelerator mass spectrometry (AMS) at the MALT, the University of Tokyo (Matsuzaki et al., 2007). Stable iodine of 127 I was determined by inductively coupled plasma mass spectrometry (ICP-MS). We already got a result that the average 129 I concentration was $(2.74 \pm 1.35) \times 10^8$ atoms/g prior to the FDNPP accident as 129 I background at Fukushima. After the accident, average isotopic ratio of 131 I/ 129 I at Fukushima is estimated to $(4.02 \pm 0.81) \times 10^{-2}$ as at March 11, 2011. The results of calculation about 131 I/ 129 I ratio made by the ORIGEN2 code are 3.18×10^{-2} for the Unit 1, 4.57×10^{-2} for the Unit 2 and 4.81×10^{-2} for the Unit 3 (Nishihara et al., 2012). In this presentation, we report the distribution of 129 I in terrestrial environment at Fukushima and 131 I/ 129 I ratios by region.

Keywords: FDNPP accident, Radioiodine, 131 I/129 I, AMS

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