Effect of cerium incorporation on the structure of hydrous rhyolitic melt

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Cerium (Ce) in silicate melts is considered to provide inportant information about the nature of silicate magmas in the Earth' s nad planetaly interiors, and especially their oxidation state. However, the behavior of Ce in a silicate melt structure may not be simple, paticularly in hydrous silicate melt where a strong depression of the Ce⁴⁺/Ce³⁺ ratio can occur (c.f. Smythe and Brenan, 2015). We investigated the effect of cerium incorporation into the structure of hydrous rhyolitic melt. hydrous ryholitic glasses quenched from melt at 1 GPa and 1300 °C show a significant decrease of H₂O solubility due to the incorporation of Ce. Pair distribution function measurement shows no distinct change in the local structures of the hydrous rhyolitic glasses with the incorporation of Ce, while we found a distict change in the position of T-OH (where T is Si or Al) Raman peak. The incorporation of Ce decreases the proportion of T-OH Species and increases that of Q⁴ and Q³ species. These results imply the occurrence of a direct charge transfer reaction between hydrogen (H) in melt and incorporated Ce as: $2CeO_2 + 2H$ (melt) $\mbox{(melt)} + H_2O$. This reaction causes a charge transfer from Ce⁴⁺ to Ce³⁺ in hydrous silicate melts may by due to the charge transfer reaction between Ce and H, which inplies that the Ce⁴⁺/Ce³⁺ ratio in hydrous silicate melts may not simply refrevt the oxidation state of magmas.

Keywords: Cerium, Hydrous rhyolitic melt, Melt structure