

## Dependence on the Film Thickness of the Characteristics of ZnO Thin Films Grown Using High-Energy H<sub>2</sub>O Generated by a Catalytic Reaction

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Zinc oxide (ZnO) is a highly useful material for various applications such as short-wavelength optoelectronics, photoconductive devices, and transparent conductive electrodes. A wide variety of growth techniques, including molecular beam epitaxy (MBE)<sup>1-3)</sup>, pulsed laser deposition (PLD)<sup>4-6)</sup>, magnetron sputtering, and metal-organic chemical vapor deposition (MOCVD)<sup>7-8)</sup>, have been used for preparing ZnO films. Although MOCVD has many advantages for industrial applications, ZnO films deposited by conventional MOCVD techniques consume a lot of electric power for the reaction of source gases and for raising the substrate temperature for the deposition of high-quality ZnO films. In order to save energy and resources, a more efficient reaction of the oxygen and metalorganic source gases during film growth is highly desired. In a previous paper, we reported a new growth method for preparing ZnO films by reacting dimethylzinc and high-energy H<sub>2</sub>O generated from the Pt-catalyzed exothermic H<sub>2</sub> and O<sub>2</sub> reaction<sup>9)</sup>. And we also reported that ZnO films with excellent crystallinity and optical properties<sup>10, 11)</sup> were grown on a-plane (11-20) sapphire (a-Al<sub>2</sub>O<sub>3</sub>) substrates. In this paper, the dependence of electrical properties of the ZnO films on the film thickness is reported.

ZnO films with various film thicknesses were grown directly on a-Al<sub>2</sub>O<sub>3</sub> substrates at 773K for 2-60 min without any buffer layer. Figure 1 shows the variation in electron mobility as a function of film thickness measured at room temperature (290 K). The electron mobility increased from 30 to 190 cm<sup>2</sup>V<sup>-1</sup>s<sup>-1</sup> with increasing film thickness from 0.1 μm to 2.8 μm. The electron mobility of the ZnO films 100 – 400 nm in thickness increases significantly with increasing film thickness. The increase in the mobility becomes slower when the film thickness becomes greater than 800 nm, and the mobility reaches the saturation point for thicknesses greater than 2000 nm. On the other hand, the electron concentration of the ZnO film 100 – 400 nm in thickness is on the order of 10<sup>18</sup> cm<sup>-3</sup>, and decreases to the order of 10<sup>17</sup> cm<sup>-3</sup> with increasing film thickness at greater than 500 nm. The mobility became less than 3×10<sup>17</sup> cm<sup>-3</sup> for the films greater than 1500 nm in thickness. Hydrogen and boron atoms were detected on the order of 10<sup>18</sup> cm<sup>-3</sup> and 10<sup>17</sup> cm<sup>-3</sup>, respectively, by secondary ion mass spectroscopy. These atoms are considered to be donor impurities in the ZnO films grown in this study.

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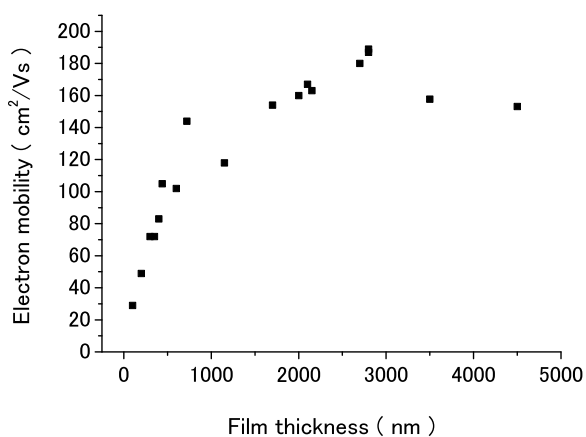


Fig. 1 Variation in electron mobility of ZnO films grown on a-Al<sub>2</sub>O<sub>3</sub> as a function of film thickness