

The effect of domain interface on the photoelectrochemistry of an organic semiconductor heterojunction

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Introduction. Photocatalyst is an example of utilizing the energy that was converted from light, where the energy is used to accelerate a reaction. Organic semiconductor p/n junction such as 3,4,9,10-perylene -tertcarboxylic-bisbenzimidazole (PTCBI) of n-type semiconductor and phthalocyanine (H₂Pc) of p-type semiconductor behave as a photocatalyst in the water phase under wide visible light region. Heterojunction domain interface can be utilized as efficient site for photocatalysis because electron-hole separation is efficient at the heterojunction. In this study, the effect of heterojunction domain interface was investigated for the photoelectrochemical decomposition of thiol compound (organic donor) under the visible light region.

Experiment condition. Electrochemistry. As shown in the scheme of Figure 1, PTCBI and H₂Pc were fabricated into three types of bilayer ITO/PTCBI (50 nm)/H₂Pc (150 nm) electrode. Each of them contain a different boundary length of H₂Pc on PTCBI such as: 0 mm, 3mm and 40 mm. Electrolyte: 10mM 2-mercaptoethanol (pH 11). Light irradiation was from ITO side using halogen light source (70mW/cm²).

Scanning Kelvin probe microscopy (SKPM). A terraced-bilayer of ITO/PTCBI (50 nm)/H₂Pc (50 nm) was fabricated. The surface potential at boundary region that contain the heterojunction domain interface was scanned using ASYLUM CYPHER S atomic force microscopy (AFM) from ASYLUMRESEARCH.

Result and discussion. The results (Figure 1) of electrochemistry study showed that the increased the boundary length, the higher the photoanodic current density for thiol oxidation. In the SKPM measurement, the boundary region of the heterojunction has more positive surface potential, V_{sp} than the single layer and the bilayer regions. The more positive V_{sp} could be related to the accumulation of holes. The accumulation of holes at the heterojunction domain interface would promote the oxidation rate of thiol.

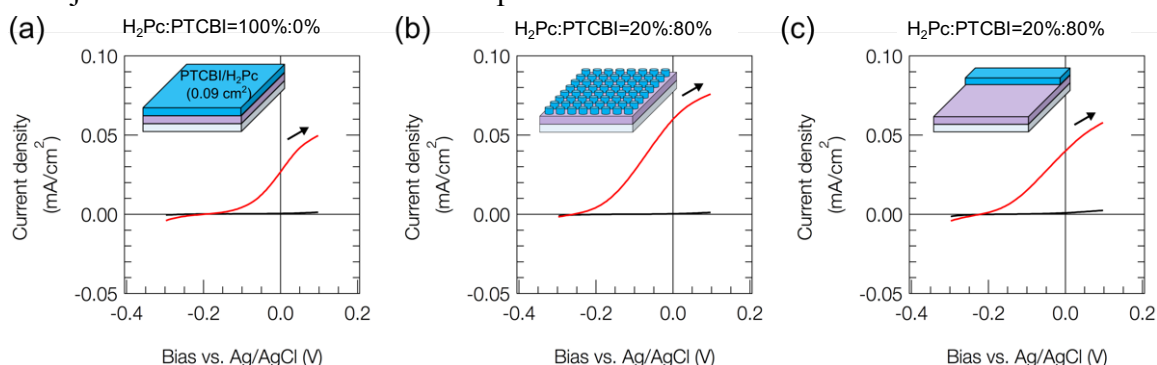


Figure 1 Voltammetry under dark (black line) and illumination (red line) conditions for electrode containing (a) 0 mm, (b) 40mm and (c) 3 mm of H₂Pc's boundary length. (a) ITO/PTCBI/H₂Pc (bilayer), (b) ITO/PTCBI/H₂Pc (8 x 8 circle patterns of H₂Pc layer on PTCBI layer, diameter of each H₂Pc circle = 200 μm, pitch = 200 μm; area ratio for H₂Pc: PTCBI = 20%: 80%) and (c) TB-ITO/PTCBI/H₂Pc (H₂Pc dimension = 0.3 cm x 0.058 cm, area ratio for H₂Pc: PTCBI = 20%: 80%). TB = Terraced bilayer. Effective electrode area = 0.3 cm x 0.3 cm. Thickness of PTCBI, 50 nm and H₂Pc, 150 nm.