

Study the effect of GO: TiO₂ proportions in GO/ TiO₂ composite for the fabrication of QCM Gas Sensor

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Graphene oxide (GO) contain a range of reactive oxygen functional groups on the sheet surface, which provide a good candidate for supporting metal oxide particles and also may become an ideal sensing platform because of potential hydrogen bonding and π - π stacking interactions [1]. Titania (TiO₂) has been explored for advanced chemical gas sensors based upon different parameters to detect different gas species [2]. This research deals with the construction of sensors based on quartz crystal microbalance (QCM) containing GO / TiO₂ composite sensor layers and to find the optimum composite parameters.

GO was prepared according to the modified Hummers method. TiO₂ was prepared using Ammonium hexafluoro titanate [AHFT]. TiO₂ particles were deposited in aqueous GO solution using liquid phase deposition technique. The composite was spin coated on the gold-plated QCM resonator (5 mm $\varnothing$, 9 MHz). Appropriate ethanol concentrations were obtained by injecting known volumes of ethanol gas and N₂ gas was passed through the chamber until the frequency difference became stable. All experiments were carried out at room temperature.

Fig. 1(a) shows the sensor response to 4167ppm EtOH gas of the functionalized QCM sensor with GO, TiO₂ and GO/TiO₂ composite of different GO ratios.

Optimum sensitivity was obtained for 30% GO in the precursor solution whereas individual composites show lower sensitivity. Further, sensing performance of functionalized QCM resonator (GO w/w=30%) was examined by changing EtOH concentration from 417 ppm – 42 ppm [Fig. 1(b)]. When EtOH gas was introduced into the testing chamber, all responses obtained were similar in shape and the response of QCM resonator decreased proportionally to the concentrations. This concludes that GO/TiO₂ composite has a significant response to EtOH gas and is anticipated to investigate for other gasses as well.

[1] M. Yang and J. He, Graphene oxide as quartz crystal microbalance sensing layers for detection of formaldehyde, *Sens. Actuators B: Chem.* 228 (2016) 486–490

[2] M. Procek, A. Stolarczyk, T. Pustelny and E. Maciak, Study of a QCM sensor based on TiO₂ nanostructures for the detection of NO₂ and explosives vapors in air, *Sensors* 15 (2015) 9563-9581

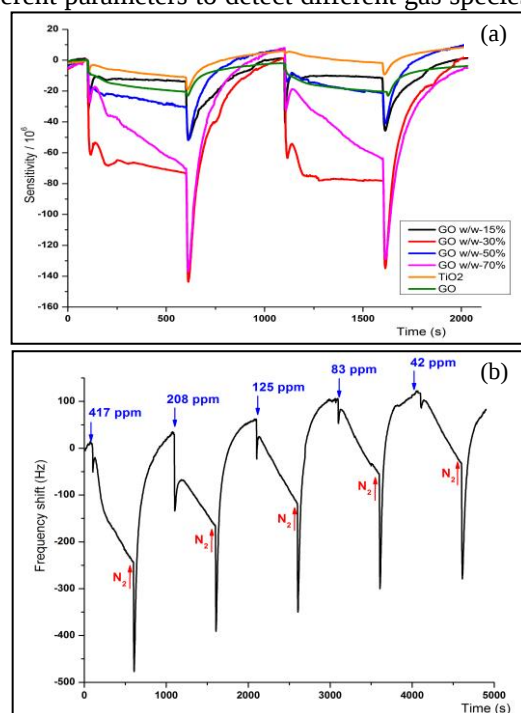


Fig 1. (a) Response to 4167ppm EtOH concentration of the functionalized QCM sensor with GO, TiO₂ and GO/TiO₂ of different GO proportions (b) Plot of frequency shift vs time of GO/TiO₂ (GO w/w=30%) functionalized QCM upon exposure to various low EtOH concentrations