

## Plasmon assisted switching of emission in $\beta$ -In<sub>2</sub>S<sub>3</sub>/Ag nanohybrid Structures

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### 1. INTRODUCTION

Photonic devices require material whose optical properties are tunable over a wide range of wavelength with faster switching rate. Semiconductor quantum dots have wide range applications in such devices. To enhance their optical properties, semiconductors are coupled with plasmonic nanostructures. Such hybrid structures exhibits tunable absorption, fluorescence and scattering with fast radiative and non radiative decay rates. In this work we demonstrate quenching and enhancement of emission of  $\beta$ -In<sub>2</sub>S<sub>3</sub> quantum dots in the proximity of Ag nanoparticles. Thus making these nanohybrids highly suitable for photonic device fabrication.

### 2. EXPERIMENTAL DETAILS

In<sub>2</sub>S<sub>3</sub> quantum dots were synthesized by homogenous precipitation method using Indium chloride (0.1 M) and TAA (0.1M) as precursors. Ag nanoparticles were synthesized using silver nitrate (0.45 mg) and Polyvinyl pyrrolidene (100 ml) (as surfactant) by heating the mixture 60 °C for 3 hours. The In<sub>2</sub>S<sub>3</sub>/ Ag nanohybrids were synthesized by growing Ag nanoparticles over In<sub>2</sub>S<sub>3</sub> quantum dots dispersed in PVP solution.

### 3 RESULTS AND DISCUSSIONS

The XRD pattern indicates the formation of  $\beta$ -In<sub>2</sub>S<sub>3</sub> with cubic crystalline structure having orientation along (611), (440), (400), (222), (311) and (211) respectively (JCPDS file no 32.0456). TEM Images of Ag nanoparticle and In<sub>2</sub>S<sub>3</sub>/Ag nanohybrid structures is shown in figure 1a and b. The average size of the Ag nanoparticle is ~ 30 nm. The Raman Spectra of pure In<sub>2</sub>S<sub>3</sub> and In<sub>2</sub>S<sub>3</sub>/Ag nanohybrid structures (Figure 2a) shows intense and broad band at ~150 cm<sup>-1</sup> (In-In stretching mode), 219cm<sup>-1</sup> (In-S bending mode ) and 300 cm<sup>-1</sup>. The peaks at 179 cm<sup>-1</sup>, 247 cm<sup>-1</sup>, 150/369 cm<sup>-1</sup> corresponds to F<sub>2g</sub>, E<sub>g</sub> and A<sub>g1</sub> vibrational modes and peaks at 438 cm<sup>-1</sup> and 474 cm<sup>-1</sup> is due to S-S bond of elemental S .

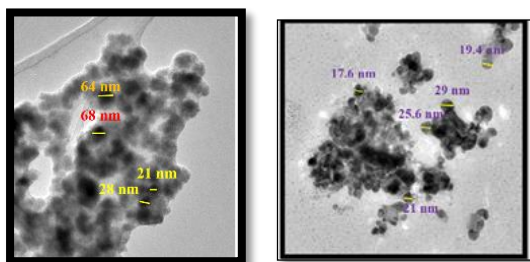


Figure 1 (a) TEM images of Ag nanoparticle and  $\beta$ -In<sub>2</sub>S<sub>3</sub> and (b)  $\beta$ -In<sub>2</sub>S<sub>3</sub>/ Ag nanohybrid structures

The excitonic absorption and emission rates of pure In<sub>2</sub>S<sub>3</sub> nanoclusters can be modified by the presence of the strong field produced by the plasmons. The extinction spectra of pure In<sub>2</sub>S<sub>3</sub> nanoclusters and hybrid nanostructure shown in figure 2b clearly depict a blue shift ~ 500 meV. The bandgap of the  $\beta$ -In<sub>2</sub>S<sub>3</sub> microflowers is ~ 3.1 eV which exhibits blue shift when compared to the bulk  $\beta$ -In<sub>2</sub>S<sub>3</sub> due to the quantum confinement effect. With incorporation of Ag nanostructures with plasmonic peak at around 463 nm to the  $\beta$ -In<sub>2</sub>S<sub>3</sub> microflower structures, bandgap shift by ~ 500 meV, which may be due to the strong interfacial electronic coupling between the neighbouring  $\beta$ -In<sub>2</sub>S<sub>3</sub> and Ag nanoparticles. The coupling of the electronic states of  $\beta$ -In<sub>2</sub>S<sub>3</sub> also leads to shift in the emission wavelength from red (670 nm) to blue region (494 nm) of the electromagnetic spectrum (figure 3).

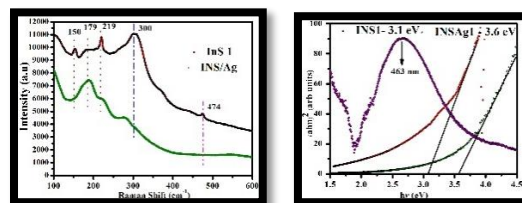


Figure 2 (a) Raman Spectra of Pure  $\beta$ -In<sub>2</sub>S<sub>3</sub> and  $\beta$ -In<sub>2</sub>S<sub>3</sub>/ Ag . (b) Excitation Spectra for  $\beta$ -In<sub>2</sub>S<sub>3</sub> and  $\beta$ -In<sub>2</sub>S<sub>3</sub>/ Ag

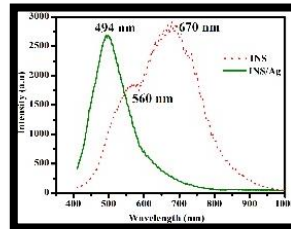


Figure 3. Emission spectra for Pure  $\beta$ -In<sub>2</sub>S<sub>3</sub> and  $\beta$ -In<sub>2</sub>S<sub>3</sub>/ Ag

### 4 Conclusion

$\beta$ -In<sub>2</sub>S<sub>3</sub>/ Ag nanohybrid structures were synthesized by simple chemical method. Incorporation of Ag nanostructures ~ 30 nm exhibited strong coupling of defect levels in In<sub>2</sub>S<sub>3</sub> microflowers leading to the blue shift of ~ 500 meV is in the extinction spectrum and switching of emission from red to blue region of electromagnetic spectrum.

### Reference

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