ゲルマニウムの高圧熱酸化機構に関する研究

Thermal oxidation kinetics of Ge under high O₂ pressure 東大院工 王旭、西村知紀、矢嶋 赳彬、鳥海 明 Univ. of Tokyo, Xu Wang, Tomonori Nishimura, Takeaki Yajima, and Akira Toriumi

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1. Introduction

The control of the thermal oxidation is one of the most important processes in semiconductor device fabrication. Si oxidation is well described by the Deal-Grove model [1]. However, Ge oxidation kinetics has been proved to be different from Si experimentally [2, 3]. We have proposed a possible kinetic model based on oxygen vacancy (V_0) to describe relatively thick Ge oxidation at atmospheric pressure [3]. It is however, not clear in terms of anomalous oxidation rate under high *p*-O₂ [4].

In this paper, ¹⁸O₂ tracer is used to study Ge oxidation under high p-O₂ for the first time. Both oxygen vacancy and atomic O interstitial (O_i) diffusion are considered in Ge oxidation under high p-O₂.

2. Experiment

P-type Ge(100) wafers were thermally oxidized in wide ranges of temperatures (T) and p-O₂. The GeO₂ thickness was determined by the grazing incidence X-ray reflectivity (GIXR) measurement.

In isotope tracer experiments, Ge wafer was first oxidized in ${}^{16}O_2$ at 520°C to form 89-nm-thick GeO₂. Then, it was re-oxidized in ${}^{18}O_2$ at 520°C for 50 min, and the total oxide thickness was 93 nm. *p*-O₂ was fixed at 40 atm in both oxidations steps (HPO+HPO). Finally, 30-nm-thick Ge¹⁶O₂ was deposited on the top by radio-frequency (rf) sputtering to minimize the surface effect in SIMS measurement. The depth profiles were analyzed by the SIMS.

3. Results and discussions

Fig. 1 shows an inverse p-O₂ dependence of the Ge oxidation rate over atmospheric pressure at all temperatures (oxidation time was fixed at 30 min.), which has never been observed in Si. **Fig. 2** shows the SIMS profiles of sample by HPO+HPO. An accumulation of ¹⁸O at the interface indicates that some diffusion species may directly transport to the interface. Molecular O₂ interstitial and atomic O interstitial (O_i) are both considered. Since O₂ diffusion in GeO₂ is suggested much limited [5], O_i is more possible diffusion species in Ge oxidation process in HPO case.

We propose a possible model for the Ge oxidation by considering both V₀ generation at the interface and O_i diffusion from oxide surface (**Fig. 3**). In APO, V₀ diffusion is dominant with GeO desorption at the oxide surface. However, V₀ formation is considerably suppressed in HPO, which has been clarified by thermodynamic calculation [6]. Thus O_i diffusion becomes important in HPO, resulting in an ¹⁸O interfacial peak in **Fig. 2**. Considering that O_i contribution to the total Ge oxidation is limited, it is reasonable that the oxidation rate is lowered in HPO. This model can well explain why HPO is a good method to achieve high performance of GeO₂/Ge gate stacks [4]. High *p*-O₂ can improve the oxide quality by suppressing the formation of V₀, meanwhile O_i diffusion can terminate the interfacial dangling bonds to achieve a high quality Ge/GeO₂ interface.

4. Conclusions

The p-O₂ dependence of the Ge oxidation rate is well explained by considering both V₀ and O_i diffusion. In HPO, V₀ formation is suppressed, and O_i diffusion passivates dangling bonds at the interface.

Reference

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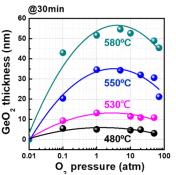


Fig. 1 GeO₂ thickness vs *p*-O₂ in a wide range of temperatures.

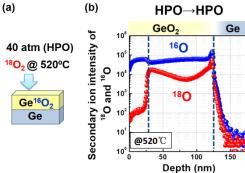


Fig. 2 Schematics of $Ge^{16}O_2/Ge$ oxidized in $^{18}O_2$ at 520°C under 40atm (HPO) and SIMS profile of GeO_2/Ge .

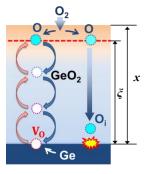


Fig. 3 Schematic of Ge oxidation model in HPO.