

## Tungsten boride nanoparticle synthesis by pulsed discharge of compacted B powder

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Tungsten borides have been firstly reported by R. Kiessling et al. in 1947 [1]. Since then, many studies have been done to synthesize various kinds of tungsten borides, namely  $W_2B$ ,  $WB$ ,  $W_2B_5$ ,  $WB_4$ . Also, these tungsten borides have been found useful in industrial application including abrasive, corrosive-resistant, and electrode exposed to harsh environment because of their excellent mechanical properties such as high hardness and corrosion resistance[2]. Tungsten borides have been synthesized by chemical vapor deposition[3], solid state reaction[4], and thermal plasma synthesis methods[5]. Generally, these methods utilize high temperature to enable the reaction between tungsten and boron. However, these methods require complicated and costly equipment. In this study, we propose a simpler and low-cost method to synthesize various kinds of tungsten boride by a large pulsed current. Especially, with a high heating rate, tungsten (W) and boron (B) were heated simultaneously into vapor and reacted with each other in only a few hundred microseconds. Moreover, with a high cooling rate as a characteristic of this method, due to supersaturation, tungsten boride vapor was nucleated and formed tungsten boride nanoparticle. With this method, by adjusting the amount of starting materials and/or varying the discharging energy, various phases of tungsten boride were confirmed. In this study, experiment was conducted with 6.2 kV charging voltage and capacitor 30 $\mu$ F in Ar gas at 100 kPa. Figure 1 shows the experiment set-up. Figure 2 shows the X-ray diffraction patterns of tungsten boride nanoparticle synthesized by pulsed discharge of compacted B powder. For the case of 5 mg B crystalline powder, main phase was  $WB_4$  while for the case of 1.5 mg of similar powder, the amount of ( $\alpha$ - and  $\beta$ -)W and WB increased, however, the amount of  $WB_4$  decreased.

### References:

- [1] R. Kiessling et al., Acta Chem. Scand., 1, 893 (1947)
- [2] K. Ma et al., Ceram. Int., 43, 8551 (2017)
- [3] I. E. Campbell et al., J. Electrochem. Soc. 96, 318 (1949)
- [4] H. Itoh et al., J. Mater. Sci., 22, 2811 (1987)
- [5] M. Kim et al., J. Nanosci. Nanotechnol., 19, 10, 6264 (2019)

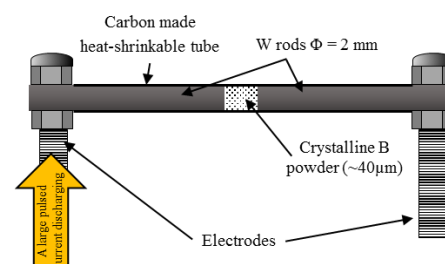


Fig. 1. Experiment set-up for tungsten boride nanoparticle synthesized by pulsed discharge of B powder.

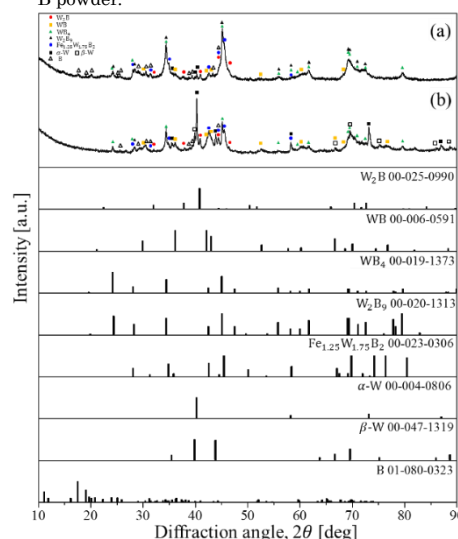


Fig. 2. X-ray diffraction pattern of tungsten boride nanoparticle synthesized by pulsed discharge of compacted (a) 5 mg B crystalline powder (b) 1.5 mg B crystalline powder.