

# First-principles study of high temperature superconductivity of compressed $\text{YKH}_{12}$

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Recently, with the successful discovery of superconductivity in  $\text{LaH}_{10}$  (~260 K at ~200 GPa) [1] and  $\text{H}_3\text{S}$  (~200 K at ~200 GPa) [2] by experiment, metal hydrides have contributed more and more to the study of high temperature superconductivity (HTSC). The search for superconducting metal-hydride at very high pressures has long been viewed as a key problem in physics. Many researchers have conducted extensive research on binary hydrogen-rich compounds involving the binding of most elements of the periodic table to hydrogen. More than 60% of the binary metal hydrides over the Mendeleev's Periodic Table have been shown superconductivity. The ternary or multi-component hydrides have become the ideal playground to discover HSTC. Recently, the superconductivity in  $\text{Li}_2\text{MgH}_{12}$  [3] was found and shown an ultra-high critical temperature of 473 K (200 GPa) by structure prediction.

In this work, we extensively performed structure search for the high-pressure phase and superconductivity of  $\text{YKH}_{12}$  using the evolutionary algorithm structure prediction method and first-principles calculations. The results showed that  $\text{YKH}_{12}$  became stable at 150 GPa and had a  $\text{C2/m}$  structure. And its stability will increase with the increase of pressure, and the second stable phase  $\text{P2}_1/\text{m}$  appears above 270 GPa. Electron-phonon coupling calculations show that  $\text{YKH}_{12}\text{-C2/m}$  is a potential high-temperature superconductor, with a  $T_c$  of 145 K at 200 GPa.  $\text{YKH}_{12}$  can be regarded as a combination of  $\text{YH}_6$  and  $\text{KH}_6$ , so our current research provides the possibility to find new high temperature superconducting ternary hydrides.

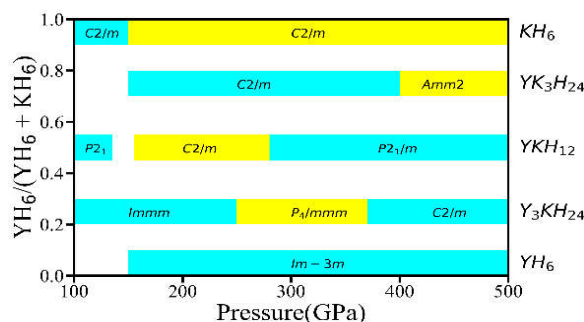


Figure 1. phase diagram of Y-K-H compound

## References:

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3. Sun, Y., et al. "Route to a Superconducting Phase above Room Temperature in Electron-Doped Hydride Compounds under High Pressure," Physical Review Letters, V. 123, No. 9, 2019