炭素ドープホウ化水素シートの合成と評価

Synthesis and evaluation of carbon-doped HB sheet

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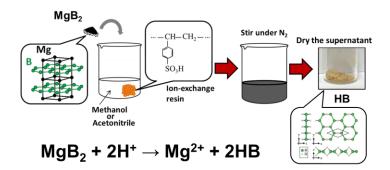
Recently, we have revealed that the hydrogen boride (HB) sheet can be experimentally prepared by exfoliation and complete ion-exchange between protons and magnesium cations in magnesium diboride (MgB₂) at room temperature (Fig. 1) [1]. Our extensive analysis revealed that the HB sheets did not show any long range order, but have a local structure of a hexagonal boron network with bridge hydrogens. HB sheets release their hydrogen as H₂ molecules by UV irradiation even at room temperature [2], while hydrogen can also be released by heating in a wide temperature range from 423 to 1473 K [1]. Theoretically, the release temperature of hydrogen by heating is expected to be lower by electron doping to HB, because of the electron occupation of anti-bonding orbital of H-B bonds by electron doping [3].

In this work, we have prepared carbon doped (C-doped) HB sheets by using C-doped MgB₂ as a starting material for ion-exchange reaction (Fig. 1) and analyze its geometric and electronic structures as well as hydrogen release property. By controlling the dopant amount in the C-doped MgB₂ as 2%, 4% and 10%, we have controlled the dopant amount in C-doped HB.

In the presentation, we will report on the results of X-ray photoelectron spectroscopy, work function

measurements, thermal desorption spectroscopy, and transmission electron microscopy as well as the theoretical calculation results about geometric and electronic structures of C-doped HB.

- [1] H. Nishino, et al., J. Am. Chem. Soc.139, 13761 (2017).
- [2] R. Kawamura, et al., Nat. Commun. 10, 4880 (2019).
- [3] T. A. Abtew, P. Zhang, Phys. Rev. B, 84, 094303 (2011).



 $\label{eq:Fig. 1} \textbf{Fig. 1} \ \text{Schematic image of the synthesis procedure of hydrogen} \\ \text{boride (HB) sheets. The same procedure was applied to} \\ \text{synthesize the C-doped HB by using C-doped MgB}_2$