# 高温 № (+02) アニールにおける SiC の表面酸化反応によって変化する 4H-SiC/SiO2 構 造へNの取り込みと除去の速度論的バランスの理解

Understandings of the kinetic balance between N incorporation and removal affected by SiC surface oxidation for 4H-SiC/SiO<sub>2</sub> structure in high-temperature N<sub>2</sub> (+O<sub>2</sub>) annealing

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# [Motivation]

Improvement of the performance of 4H-SiC MOSFET was demonstrated by nitridation process, which was related to the density of surface N atoms incorporated during the interface nitridation process [1]. However, for the conventional nitridation process by NO post-oxidation annealing (POA), there is a plenty of room to improve the surface N density, which would be determined by the kinetic balance between the simultaneous reactions of N incorporation and N removal from the SiC surface during annealing. For designing a nitridation process to achieve a higher surface N density (N(t)), where t is the annealing duration), the control of such balance is critical. In case of NO annealing the N removal is attributable to the surface oxidation [2]. To simplify the model, we assume the reaction rate of N-incorporation  $(N_+)$  to be constant during POA, and the removal rate  $(N_-)$  to be proportional to the surface oxidation rate (R) and N(t). Thus, the effective nitridation rate (dN(t)/dt) would be as follows:

$$\frac{dN(t)}{dt} = N_{+} - N_{-} \Longrightarrow N(t) = \frac{1}{c} \cdot \frac{N_{+}}{R} [1 - \exp(-cR \cdot t)], \text{ where } N_{-} = cR \cdot N(t)$$
(1)

where c is a constant. According to Eq. (1), N(t) would saturate at a certain level that is determined by  $N_{+}/R$ , which could be experimentally determined. This study aims to provide systematic understandings of the kinetic balance between N incorporation and N removal based on Eq. (1) for 4H-SiC(0001)/SiO<sub>2</sub> structures. For this purpose, the high-temperature N<sub>2</sub>-POA was used in this study to control  $N_+$  and  $N_-$  independently by tuning R with intentional O2 introduction instead of annealing in NO that decomposes to produce O2 with poor controllability. [Experimental]

4°-off 4H-SiC(0001) with epitaxial layer (~1×10<sup>16</sup> cm<sup>-3</sup> n-type doped) were used as substrates. After oxidizing in dry O<sub>2</sub> to grow around 6nm-thick SiO<sub>2</sub> layer, POA was done at 1370°C from 1 to 45min in N<sub>2</sub> ambient, with a series of small amount of O<sub>2</sub> whose partial pressure was denoted as P. The SiO<sub>2</sub> regrowth thickness ( $\Delta T_{OX}$ ) during N<sub>2</sub>-POA was determined by measuring X-ray reflectivity before and after POA. After removing the SiO<sub>2</sub> layer completely with HF, the relative ratio of N (N/Si) was estimated with the peak area ratios of N1s to Si2p by XPS.

## [Results and discussions]

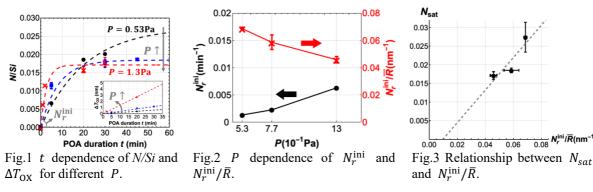
Fig. 1 shows the t dependence of N/Si as well as  $\Delta T_{OX}$  (inset) for a series of P. Saturation of N/Si is found for large t. The saturated surface N concentration  $(N_{sat})$  was determined by fitting the N/Si - t relationship with the exponential function shown by Eq. (1). The initial nitridation rate  $N_r^{\text{ini}}(=[N/Si(t)]/t)$  in the initial stage of POA was defined to approximately estimate  $N_+$ , while the oxide growth rate  $\overline{R}$  was determined from the slope of  $\Delta T_{\text{OX}} - t$  relationship. Fig. 2 summarized the *P* dependence of  $N_r^{\text{ini}}$  and  $N_r^{\text{ini}}/\overline{R}$ . Note that  $N_r^{\text{ini}}$  increases with *P*, which means a slight addition of O<sub>2</sub> enhances  $N_+$ . This is explainable considering that the incorporation reaction would be triggered by surface oxidation [3]. On the other hand,  $N_r^{\text{ini}}/\bar{R}$  decreases monotonously with P, indicating that the enhancement of  $N_{-}$  by oxidation is more significant than that of  $N_{+}$ . Fig. 3 shows the relationship between  $N_{sat}$  and  $N_{r}^{\text{ini}}/\bar{R}$ . A linear relationship was observed, which clearly shows that  $N_{r}^{\text{ini}}/\bar{R}$  is a good indicator for saturated surface N density rather than *P*. *P* would affect  $N_{r}^{\text{ini}}/\bar{R}$  since  $N_{+}$  and  $N_{-}$  show quantitatively different relationships with *P*, and therefore have an impact on  $N_{sat}$ . Besides, a higher  $N_{sat}$  was found for P = 0 cases (data not shown), which will be explainable by taking the active oxidation rate as  $N_{-}$ instead of  $\overline{R}$  in the passive oxidation cases.

## [Conclusions]

While a slight amount of O<sub>2</sub> inclusion in N<sub>2</sub>-POA ambient was found to enhance  $N_+$ , the parameter  $N_+/R$  was verified to be a good indicator of the incorporation-removal balance which determines  $N_{sat}$ .

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References [1] J. Rozen et al. J. Appl. Phys. 105, 124506 (2009). [2] K. McDonald et al., J. Appl. Phys. 93, 2257 (2003). [3] T. Yang et al. Extended Abstract of the 8th ADPS Conference (2021) IIA-9.



\_\_\_\_\_ *N\_r^ini/R*(nm<sup>-1</sup>)