

Nanostructure control of 2D van der Waals Materials  $\alpha$ -MoO<sub>3</sub>

The University of Tokyo,<sup>o</sup>(M1) Jiaqi Yang, Hitoshi Tabata and Hiroaki Matsui\*

E-mail: yang@bioxide.t.u-tokyo.ac.jp, \*hiroaki@ee.t.u-tokyo.ac.jp

Molybdenum oxides (MoO<sub>x</sub>) are one of interesting oxides in various applications such as chemical catalysis, bio-sensing, and photovoltaic devices [1]. In particular,  $\alpha$ -type MoO<sub>3</sub> ( $\alpha$ -MoO<sub>3</sub>) is known as a two-dimensional (2D) van der Waals material, which provides low-dimensional electrical and optical characteristics. Additionally, control of the reduction of Mo elements causes change in electronic states from transparent insulating to semi-metals, which produces highly tunable and versatile applications. So far, growth related to  $\alpha$ -MoO<sub>3</sub> films and nanostructures are reported some papers [2]. However, studies of the controlled synthesis of  $\alpha$ -MoO<sub>3</sub> films and nanostructures are still limited. In this presentation, we report on growth of  $\alpha$ -MoO<sub>3</sub> nanostructured films and their related materials.

$\alpha$ -MoO<sub>3</sub> nanostructured films were grown on TiO<sub>2</sub>-terminated SrTiO<sub>3</sub> (100) substrates due to small lattice mismatch between  $\alpha$ -MoO<sub>3</sub> (010) and SrTiO<sub>3</sub> (100) by pulsed laser deposition (PLD). The x-ray diffraction (XRD) revealed that the film sample grown under high O<sub>2</sub> pressure of 10 Pa showed a single crystal of  $\alpha$ -MoO<sub>3</sub> along the [010] direction from the result of 2D reciprocal space mapping, which was also confirmed by  $\mu$ -Raman scattering. Firstly, the effect of O<sub>2</sub> pressure on the chemical composition of MoO<sub>x</sub> was investigated using x-ray photo-emission spectroscopy (XPS). The reduction of O<sub>2</sub> pressure from 15 Pa to 0.1 Pa provided remarkable change in valence states of Mo elements (Mo<sup>6+</sup>, Mo<sup>5+</sup> and Mo<sup>4+</sup>) [Table 1]. Chemical ratios of the Mo<sup>5+</sup> and Mo<sup>4+</sup> states gradually increased with reducing O<sub>2</sub> pressure as a result of the decrease in the Mo<sup>6+</sup> state. Besides, we observed morphologies on the film surfaces using scanning electron microscopy (SEM). The high O<sub>2</sub> pressures at 10 Pa clearly showed nano-needle structures. However, the decrease in O<sub>2</sub> pressure caused change in morphology from nano-needle structures to elongated nanowire structures. The sample grown under a low O<sub>2</sub> pressure of 0.1 Pa showed a smooth surface with a roughness of 5 nm. We succeeded to fabricate nanostructured films of  $\alpha$ -MoO<sub>3</sub> under the high O<sub>2</sub> pressures by PLD. Hereafter, we will progress plasmonic biosensing applications based on MoO<sub>x</sub>/MoO<sub>3</sub> nanostructures. MoO<sub>x</sub> with low valence states of Mo elements has metallic properties in the visible range, which are expected for surface-enhanced Raman scattering (SERS).

Pressure (O <sub>2</sub> )	Mo oxidation states (%)			Average formula
	Mo <sup>+VI</sup>	Mo <sup>+V</sup>	Mo <sup>+IV</sup>	
15 Pa	100%	—	—	MoO <sub>3</sub>
12.5 Pa	90.1%	9.9%	—	MoO <sub>2.95</sub>
10 Pa	89%	11%	—	MoO <sub>2.94</sub>
7.5 Pa	82.1%	17.9%	—	MoO <sub>2.91</sub>
5 Pa	74.3%	25.7%	—	MoO <sub>2.87</sub>
2.5 Pa	27%	59.1%	13.9%	MoO <sub>2.56</sub>
1 Pa	19.2%	38.8%	41%	MoO <sub>2.37</sub>
0.1 Pa	10.8%	42.3%	46.9%	MoO <sub>2.32</sub>

Table 1. XPS results for MoO<sub>x</sub> nanostructured films on the stepped STO substrates.

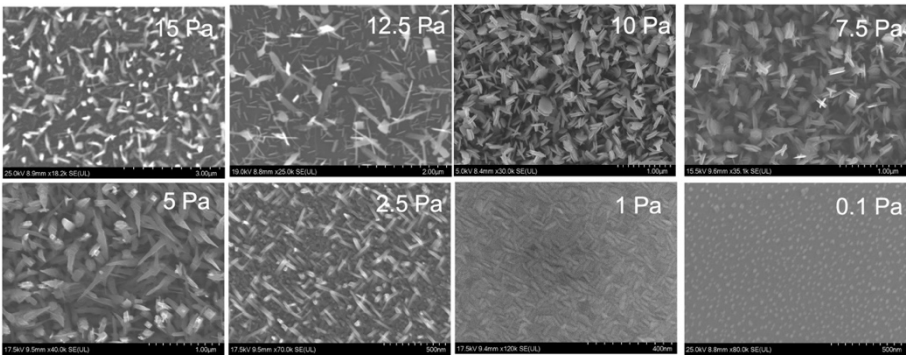


Figure 1. SEM results for MoO<sub>x</sub> nanostructured films on the stepped STO substrates.

Reference:  
[1] de Castro I A, Datta R S, Ou J Z, et al. Molybdenum oxides—from fundamentals to functionality[J]. Advanced Materials, 2017, 29(40): 1701619.  
[2] Gao Q, Wang S, Fang H, et al. One-dimensional growth of MoO<sub>x</sub>-based organic–inorganic hybrid nanowires with tunable photochromic properties[J]. Journal of Materials Chemistry, 2012, 22(11): 4709-4715.