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B-6-3 Quantitative In-depth Profile of Passivated Oxide Layers of GaAs by AES-SIMS - a Comparison of Thermal, Anodic and Plasma Oxidation

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Introduction In order to establish the technology of GaAs devices, there have been a number of investigations recently, in which one of the most important problems may be suitable choice of passivation methods to form homogeneous nativeoxide on GaAs crystal. In the present study, we will describe a comparison of the in-depth profile of surface passivated (oxide) layers prepared by thermal¹⁾, anodic²⁾ and plasma³⁾ oxidation, by means of a simultaneous measurement of AES-SIMS with a proposed quantitative analysis for complicated oxide layers. Experimental An Auger system(Varian) with a CMA optics with an internal electron gun and a secondary ion mass spectroscopy(SIMS) system are equiped together to a conventional UHV chamber which can be evacuated down routinely to 5 x 10^{-10} Torr. Further aspects of the simultaneous measurement of AES-SIMS ha been described elsewhere^{4,5}

Quantitative Analysis of In-depth Profile Peak-to-peak height H, from an element a in a derivative AES spectrum is considered to be proportional to atomic concentration X_{\sim} as⁶⁾

.....(1) $H_{\alpha} = k_{\alpha} X_{\alpha}$

where \textbf{k}_{α} is denoted as AES sensitivity of the element $\alpha.$ In the present study, the AES sensitivities of Ga, As and O are determined following Pons et al⁷. When the AES sensitivities are known, the atomic concentration of each element is estimated as

 $X_{Ga} = H_{Ga}/k_{Ga}$, $X_{As} = \delta_{As}H_{As}/k_{Ga}$, $X_{O} = \delta_{O}H_{O}/k_{Ga}$ (2) where $\delta_{AS} = k_{Ga}/k_{AS}$ and $\delta_0 = k_{Ga}/k_0$, which are denoted as the relative sensitivity of As and O, respectively. Table 1. AES relative The relative sensitivities are determined on variously prepared oxide layers and summerized in Table 1. As seen in the Table, the relative sensitivities such as δ_0^{oxide} and δ_{As}^{bulk} agree well with those estimated from the literature⁸, while the others significantly differ

	δ _{As}	80°
GaAs(Bulk)	2.31±0.20	
Interface	3.42±1.09	0.19±0.07
Oxide Layer	425±0.45	0.36±0.02

from the literature. It has been also found that the relative sensitivity in such systems as Cu-Au, Ta-O, Ta-N and Ta-C changes with concentration9?

Comparison of In-depth Profile of Passivated Layers By applying the relative sensitivities summerized above, quantitative in-depth profiles of each sample were

* Samples by thermal and plasma oxidation were furnished by T.Sugano's Lab. (Univ. of Tokyo) and anodic oxidation by H.Hasegawa's Lab. (Hokkaido Univ.)

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determined. Fig. 1 shows a typical example of the profile of thermally oxidized GaAs (620°C, 5hr in dry air). It can be seen that As does not exist in the oxide layer and chemical composition is Ga202. the interface (which is defined 1nside as where the amount of oxygen decreased to the half value of that of the oxide layer) a large pile-up of As was observed. It was also found that the pile-up and the thickness of the transition layer between the oxide and the GaAs bulk increased with higher temperature of the oxidation. Figs. 2 and 3 show examples of the profile of passivated layers by anodic and plasma oxidation, respectively. Characteristic common to both passivated layers is the fact that nearly same amount of Ga and As exists in the oxide layers without any pileup. A distinct difference between the two is content of oxygen in the oxide, the chemical composition being estimated as $GaAsO_{1.7 \ v2.0}$ for anodic oxidation and GaAsO1.0 for plasma oxidation. SIMS Measurement Information from SIMS measurements will be described concerning especially with the chemical constituents

of the transition layers for each sample.







Fig.2. In-depth profile of passivated layer by anodic oxidation



Fig.3. In-depth profile of passivated layer by plasma oxidation

References 1) F.Koshiga and T.Sugano, Japanese J.Appl.Phys.Suppl.<u>16</u>-1(1977),465, 2) H.Hasegawa and H.Hartnagel, J.Electrochem.Soc.,<u>123</u>(1976),713, 3) T.Sugano and Y.Mori, J.Electrochem.Soc.,<u>121</u>(1974),113, 4) M.Mohri, M.Hashiba, K.Watanabe and T.Yamashina, submitted to J.Catalysis, 5) M.Mohri, H.Kakibayashi, K.Watanabe and T.Yamashina, submitted to Appl.Surface Sci., 6) R.E.Weber and A.L.Johnson, J.Appl. Phys.,<u>40</u>(1969),314, 7) F.Pons, J.Le Héricy and J.P.Langeron, Surface Sci.,<u>51</u> (1975),336, 8) P.W.Palmberg, G.E.Raich, R.E.Weber and N.C.MacDonald, "Handbook of Auger Electron Spectroscopy", Phys.Electr.Ind.Inc.(1972), P.M.Hall, J.M.Marabito and D.K.Conley, Surface Sci.,<u>62</u>(1977),1

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