

A-2-6 Anodization of Silicon in RF induced Oxygen Plasma

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§1 Introduction

The anodization of silicon in oxygen plasma is very promising since the oxidation at low temperature suppresses the formation of stacking faults¹⁾ and variation of diffusion profiles. It was first reported by J.F. Ligenza²⁾ and was also investigated by other authors^{3,4)}. However, neither oxide films as thick as 1 μm nor defects in the oxide were studied in detail. Besides, the oxidation mechanism and masking against the oxide growth have remained as important problems for practical application. Here, the conditions for forming thick oxide films in RF oxygen plasma, oxidation mechanism, properties of the oxide, defects in the oxide and Si substrates, and selective oxidation are reported.

§2 Experiments and Discussions

2.1 Oxidation procedures and mechanism : The oxidation was performed with the equipment described elsewhere⁵⁾, at the temperature of 600°C and pressure of 0.2 Torr. oxide films as thicker than 1 μm were formed with the constant current of 100 mA/cm² within one hour. (Fig. 1)

The oxidation mechanism was studied by measuring the in-depth-profiles of O¹⁶ and O¹⁸ ions in the SiO₂ films which were anodized with O¹⁶ and subsequently done with O¹⁸. These profiles indicate that the oxidation has proceeded at both Si/SiO₂ and SiO₂/plasma interfaces through the motion of both silicon and oxygen ions across the oxide. Therefore, the oxidation process here can not be considered as diffusion limited process as discussed by some authors^{2,3)}. The high oxidation rate at low temperature is due to the electric field which drifts negative oxygen ions to silicon surface and positive silicon ions to the oxide-plasma interface.

2.2 Oxide properties : The oxide films were of amorphous structure as revealed by electron diffraction patterns, and were found to have the same bonding characters as thermally grown SiO₂ by measuring the Infrared Absorption spectra. The electric breakdown field was as high as about 7×10^6 V/cm.

2.3 Defects in the oxide and substrates

2.3.1 Transmission Electron Microscopy : TEM was used to investigate the Si/SiO₂ interface⁶⁾. The oxide was clean and no micro-structural defects such as silicon clusters were observed. Besides, the surface roughness was the same as the one of thermally grown Si/SiO₂ interface.

2.3.2 Electron Spin Resonances and C-V method : One ESR line was detected. It had the g-value of $2.000 \sim 2.007$ and density of $2 \sim 3 \times 10^{12}$ spins/cm².

This defect center was found to be in the region of about 100Å from the Si/SiO₂ interface and was attributed to the surface-states whose density was reduced to less than 10^{11} /cm² by annealing in forming gas at 450°C. (Fig. 2)

2.3.3 Defects in Si substrates : Sirtl etch was done but no oxidation induced stacking fault was observed.

2.4 Selective oxidation : Si₃N₄ film can not be used as oxidation mask since it was oxidized in oxygen plasma⁷⁾. Here, Al₂O₃ was used as oxidation mask.

It was confirmed that Al₂O₃ of 3000Å in thickness was effective in masking against the oxide growth on the silicon surface protected, by measuring the thickness of Al₂O₃ film after the anodization and by observing the bevelled surface with a photo microscope. (photo 1)

§3 Conclusion

The oxidation of silicon in an RF induced oxygen plasma was successfully used to produce oxide as thick as one micron within one hour at temperature of 600°C. It was shown that the oxide was comparable to the best thermally grown oxide. Moreover, in contrast to thermal oxidation, no oxidation induced stacking fault was generated in silicon substrate. Al₂O₃ was found to be useful as mask for selective oxidation. This process is very promising for LSI device fabrication.

References :

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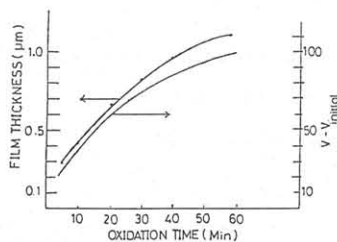


Fig.1 The growth rate of SiO₂ in the anodization with constant current.
(T=600°C, I=100mA/cm²)

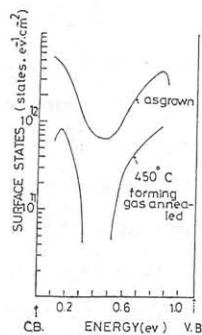


Fig.2 The density of surface states.

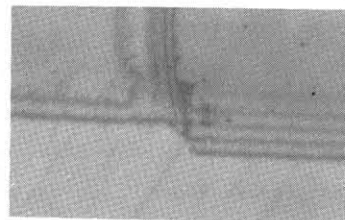


Photo 1 Silicon surface bevelled by 1 degree. (x - 300) Showing Al₂O₃ worked as oxidation mask.