

Invited

Band Offsets at II-VI Heterojunctions

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Heterojunctions could become particularly useful for II VI compounds in which pn junctions are not obtainable, though better understanding of their properties will be needed. The standard Harrison theory is discussed vis a vis alternative possibilities in the empirical common anion rule. Experimentally important factors are the apparent process dependence of band offsets, and the abruptness attainable by various processes, particularly within the past two years. The merits of HgTe/CdTe in terms of theory, advantageous processing, and possible applications are cited.

In III V compounds, heterojunctions substantially extend the capabilities of p-n homojunctions, e.g. in various quantum-well devices. In II VI compounds, while such sophistication may eventually develop, the more immediate use for heterojunctions is to substitute for pn junctions which are not usually obtainable. Photovoltaic technology already does this and one hopes electroluminescence will in the future.

Fig. 1 sketches some ways in which band offsets occur at the interface of two semiconductors. Fig. 1 (a) shows both offsets, ΔE_c and ΔE_v , contributing in a positive way to make up the bandgap difference, $E_{g1} - E_{g2}$. This is the carrier confinement structure so widely used in III Vs. Fig. 1 (b) shows one offset, ΔE_v , making up for more than $E_{g1} - E_{g2}$, so that the other offset, ΔE_c , will be called negative. This type of structure is favored for solar cells, e.g. InP/CdS, where one does not want to block both carrier flows. A rather special case is shown in Fig. 1 (c), where both an offset and a bandgap are zero. It is similar to a Schottky barrier, should provide substantial minority carrier injection, and will be discussed later as a special example.

A Theory and an Empirical Rule.

How band offsets arise from the physics and chemistry of the interface is a long-standing question. In Swanks¹⁾ 1967 study of II VI photo-

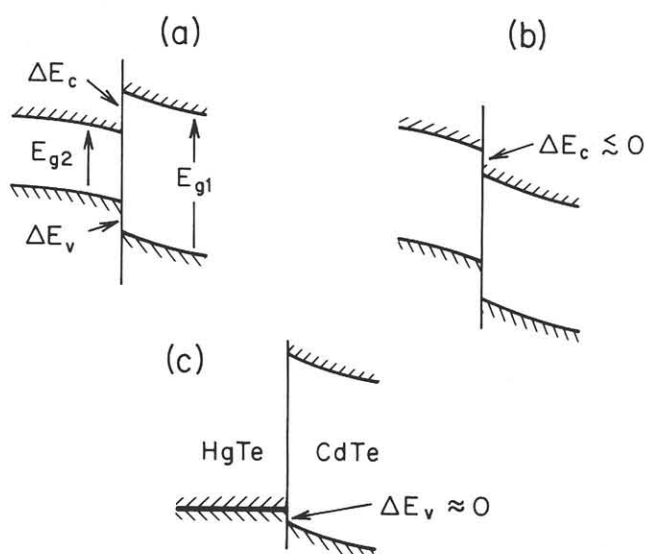


Fig. 1 Energy band lineups at heterojunctions. (a) Smaller bandgap, nested within a larger one. (b) Staggered lineup. (c) Lineup proposed for HgTe/CdTe.

electric threshold, $\Phi = \text{vacuum energy} - E_v$, he observed that Φ was "... determined primarily by the chalcogen ...". Combining his own II VI and earlier III V data of others, he further found a linear relation between Φ and Pauling electronegativity of the anion. In 1976 McCaldin, McGill, and Mead²⁾ using a different method for determining E_v analyzed Schottky barrier height, ϕ_p , for holes at semiconductor/Au interfaces obtaining results *qualitatively*

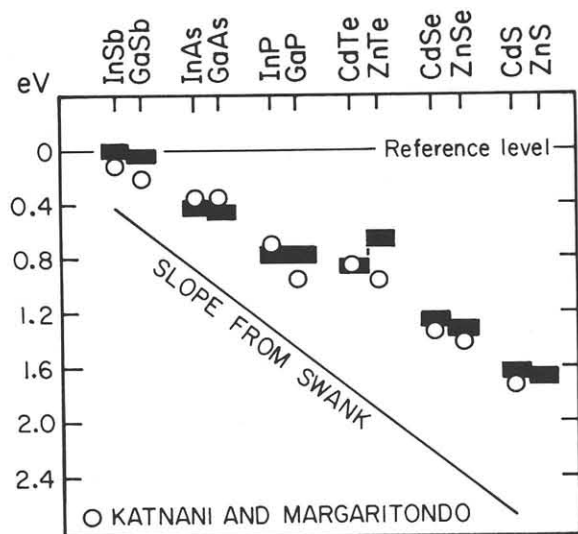


Fig. 2 E_v for III V and II VI compounds. ■ data from common anion rule with Au E_F as reference. O data sets Ge E_v as zero. For detail on Swank data, see ref. 3, Fig. 6

similar to Swanks, often alluded to as the "common anion" rule. However, their measurements of E_v showed a significantly *smaller* dependence on electronegativity than Swanks, as depicted in Fig. 2 and shown more explicitly in ref. 3. In 1977 Harrison⁴⁾ proposed a LCAO theory of heterojunctions, which has since become the standard in the field and which agrees with many recent band offset measurements to about ± 0.2 eV. It also agrees reasonably well with the Swank data, especially in the overall sense that E_v is predicted to decrease ~ 2.7 eV in going from the antimonides to the sulfides.

Recently Katnani and Margaritondo⁵⁾, using photoemission measurements, obtained E_v values more consistent with McCaldin, McGill, and Mead than with Swank, and hence Harrison. Their E_v values are also plotted in Fig. 2, where it should

Volts

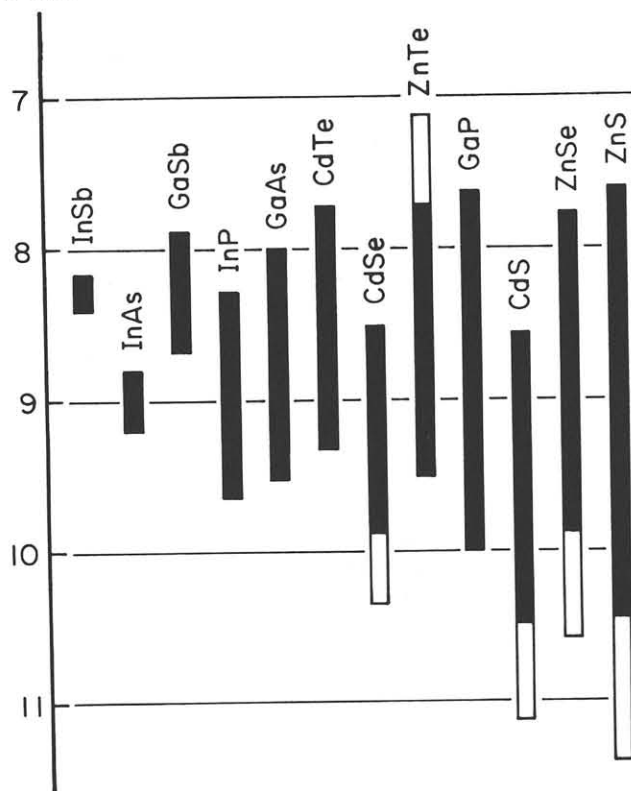


Fig. 3 Relative positions of bandgaps according to Harrison model (voltage measured from vacuum level). Solid black bars represent bandgaps at 0°K of compounds which can be made highly conducting n-type or p-type. Where one conductivity type is not accessible (at high conductivity), the appropriate end of the bar is left unshaded, e.g. n-type regime in ZnTe. Semiconductors are arranged in order of increasing bandgap from left to right.

Volts

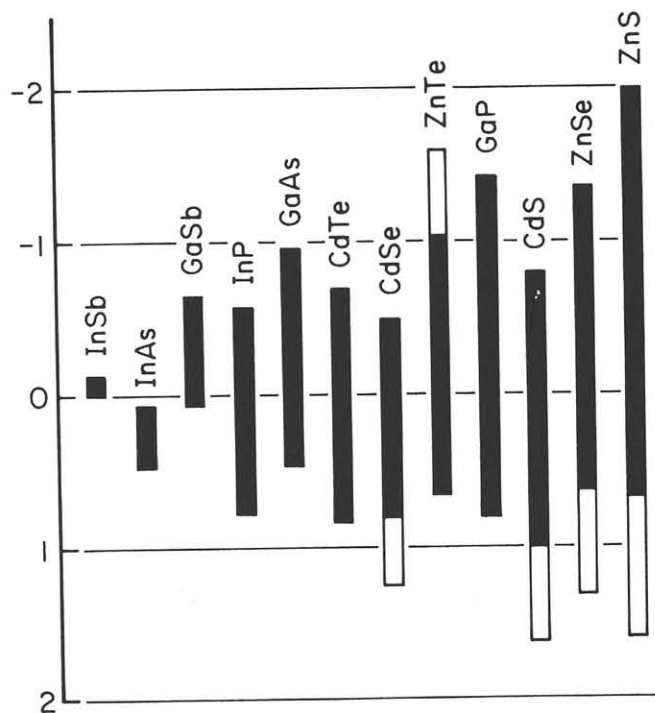


Fig. 4 Plot similar to Fig. 3, except using data from ref. 3 (common anion rule). Zero is Au E_F . Bandgaps at 300°K.

be noted that reference levels are arbitrary. In going from antimonides to sulfides, Katnani and Margaritondo, as well as McCaldin, McGill, and Mead, find that E_v changes only about 1.7 eV.

These comparisons can be seen in greater detail in Figs. 3 and 4, in which E_v appears as

the bottom of a bar representing bandgap. Overall the Harrison E_V values in Fig. 3 decline more steeply than those in Fig. 4. Also a few near-neighbor heterojunctions, e.g. GaAs/CdTe, differ substantially in the two figures.

How do experimental band offsets compare to Figs. 3 and 4? As mentioned earlier, Fig. 3 gives remarkably good agreement with some rather well-characterized heterojunctions. For example, the n-CdS/p-InP solar cell is predicted to have $\Delta E_C = -0.29$ eV by the Harrison model, which is almost halfway between the two reported experimental values, -0.56 eV⁶⁾ and -0.112 eV⁷⁾. On the other hand, Fig. 4 predicts $\Delta E_C = +0.23$ eV, in rather poorer agreement.

There are reasons, however, that experimentalists, especially those working with II VIs, should consider Fig. 4, besides checking against the standard Harrison theory. Of course, the Katnani-Margaritondo results suggest such a check. Also qualitative information on II VIs such as the well-known difficulty in making ohmic contact to ZnS, or intercomparisons of ZnSe and ZnS, suggest Fig. 4 may be relevant.

Lastly it is interesting to note that band offsets for several heterojunctions are roughly the same in the two figures. For example, ΔE_V for GaP/ZnSe is 0.58 eV in Fig. 3 and 0.51 eV in Fig. 4. This heterojunction, which has recently been formed at relatively low temperature^{8,9)}, $\sim 300^\circ\text{C}$, could, in principle, provide significant hole injection into ZnSe, since the difference in bandgap is only ~ 0.4 eV and the modest ΔE_V is no serious obstacle.

Experimental Considerations.

Treatments of band offsets just discussed are said to be "linear" because the only independent variables are parameters of each individual semiconductor. No terms for a specific interface, e.g. describing the specific bonds at the interface or the specific orientation of the interface, are included. Thus some variability is expected from specific interface to interface.

The observed experimental variability, however, appears to include additional effects. When the epitaxial growth of semiconductor A onto B is reversed (i.e. B grown onto A), band offsets

change, in one instance by 0.5 eV. Examples of this growth-sequence dependence, recently compiled by Bauer et al.¹⁰⁾, mostly are complicated by strong interdiffusion/doping effects. One example¹¹⁾, however, GaAs/AlAs versus AlAs/GaAs, which exhibits 0.25 eV change in offsets on reversing the growth sequence, is particularly worrisome. Two possible explanations¹²⁾ in this system are presently being considered: dependency of C incorporation or of Al/Ga interchange on growth sequence. Thus, even in this seemingly simple system, it may be a lack of the ideal compositional abruptness assumed in the simple interface models that causes the variability. Incidentally, the at-present poorly understood variability could become useful to obtain desired offsets for device purposes.

Experimentally the abruptness achievable has been improved substantially in recent years by use of lower growth temperatures. Growth of ZnSe and ZnS, usually on III V substrates, is now performed at $\sim 300^\circ\text{C}$ to 400°C by both MBE and CVD (usually metalorganic), a considerable achievement. Whether the abruptness obtained is sufficient to avoid some of the variabilities mentioned is an interesting question.

Lattice strain is also clearly a factor in junction quality, whether due to lattice mismatch or differences in thermal expansion. Also, deliberately grown-in lattice strain sufficient to change a bandgap substantially¹³⁾ is now achieved in III V systems.

HgTe/CdTe.

This particular system may be quite timely to help answer some of the questions just mentioned, one has reason to hope. In any case, the present discussion may benefit by focusing on a specific case for illustration.

From various guiding ideas, one expects the small offset, ΔE_V , sketched in Fig. 1 (c). The common anion rule, of course, predicts $\Delta E_V = 0$, since Te is common to both sides of the junction. The older Anderson electron-affinity rule¹⁴⁾, applied to the data of Shevchik et al.¹⁵⁾, gives a modest offset, $\Delta E_V = 0.3$ eV. Harrison's tabulated values of E_C and E_V , sketched in Fig. 3, do not include the Hg chalcogenides, possibly because of certain difficulties with the

conduction band on one side of the junction overlapping the valence band on the other side, as discussed in the appendix to his paper⁴⁾. However, if one neglects this difficulty, the formulations in his paper indicate $\Delta E_V < 0.1$ eV, as has been pointed out¹⁶⁾.

Experimentally this heterojunction is favored by a modest lattice mismatch, $\sim 0.3\%$, which, however, is still appreciably larger than for the selenide and sulfide analogs. Because of the substantial current interest in HgCdTe detectors, substrate material of good quality is often obtainable and considerable data on physical properties is being published. Probably the most important consideration, however, is processing temperature. Recent diffusion measurements^{17,18)} now enable good estimates of the concentration profiles to be expected at different temperatures.

The lowest reported junction formation temperature comes from MBE studies¹⁹⁾. Although even lower temperatures were investigated, most growths were made near 200°C. Extrapolating from 340°C - 280°C diffusivity measurements¹⁷⁾, $D \approx 10^{-17} \text{ cm}^2 \text{ s}^{-1}$ can be estimated so intermixing layer thicknesses near 20 Å or so are probable. From optical absorption measurements, an offset of $\Delta E = 0.04$ eV was deduced¹⁹⁾.

Temperatures near 350°C have been reported²⁰⁾ for CVD growth. Expected^{17,18)} diffusivity is $\approx 10^{-13}$ or $10^{-12} \text{ cm}^2 \text{ s}^{-1}$, so that intermixing layer should be $\sim 100\times$ thicker than for MBE growth. Schottky barrier heights up to 0.92 eV were measured¹⁶⁾, which is 0.58 eV short of the band-gap. The 0.58 eV difference arises from several factors, including the inversion layer effects originally discussed¹⁶⁾ and, in retrospect, from intermixing.

LPE, widely used in making HgCdTe structures, is generally practiced at even higher temperatures. However, lower temperatures should be feasible in Hg-rich solutions, because Hg is such a good solvent for many elements, including chalcogens²¹⁾. Using Te-saturated Hg solutions, growth of HgTe onto CdTe substrates has been obtained at MBE-type temperatures, i.e. 200°C and lower²²⁾.

If small ΔE_V values are confirmed for HgTe/CdTe, the proposed superlattice²³⁾ becomes an especially interesting goal. Furthermore, small

ΔE_V values imply substantial hole injection can be obtained in this heterojunction. Because of the long-standing interest in II VI electroluminescence, whether from electrolyte²⁴⁾ or solid-state injection, the selenide and sulfide analogs should then be investigated.

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