Single Step Gap Filling Technology for Subhalf Micron Metal Spacings on Plasma Enhanced TEOS/O₂ Chemical Vapor Deposition System

Katsuyuki MUSAKA, Shinsuke MIZUNO, Kiyoaki HARA

Applied Materials Japan Inc. Technology Center 14-3 Shinizumi Narita, Chiba 286 Japan

A new inter-metal dielectric (IMD) film formation technology for subhalf micron metal to metal space filling has been developed by utilizing C_2F_6 gas on mixed frequency plasma enhanced TEOS/O₂ chemical vapor deposition (PE-TEOS/O₂ CVD) system. Gap filling capability for greater than 2.3:1 aspect ratio of metal to metal spacings was confirmed. Furthermore, it was obtained that those films contain 5~10 atomic % of F, which reduce the dielectric constant of IMD film and is expected to enhance the device performance.

1. Introduction

As the device geometry getting smaller, the requirement for IMD film is more severe, mechanically and electrically. As the result, the IMD process used in advanced semiconductor devices are very complex. Multi-layer structure (PE-CVD~ SOG~ PE-CVD^{1~4}), PE-CVD~TEOS/O₃~PE-CVD^{5~9})) is widely used with the sacrifice of productivity.

The characterization of PE-TEOS/O2/NF3 CVD process has been reported to reduce the void formation and to be simple, single step process.¹⁰⁻¹¹) Recently, we have developed PE-TEOS/O2/C2F6 CVD process. By adding C₂F₆ gas to TEOS/O₂ chemistry on PE-CVD system, it was confirmed that the gap filling capability was remarkably enhanced, which can be applied for single step gap filling of subhalf micron metal to metal spacings. So, the IMD process productivity will be greatly improved. In addition, it was obtained that the H2O content and absorption of the film was maintained as low as that of conventional PE-TEOS/O2 CVD film. Furthermore, those film contains 5~10 atomic % of F, which reduce the dielectric constant of IMD film and, is expected to enhance the device performance.

In this paper, physical and chemical properties (stress, wet etch rate, FT-IR,SIMS and ESCA) of the PE-TEOS/O₂/C₂F₆ CVD film will be presented and compared to those of conventional PE-TEOS/O₂ CVD film. Also, the result of electrical measurement (I-V, C-V) will be presented, as well as the SEM microphotographs for deposition profile.

2. Experimental

All the CVD experiments were carried out on AMAT Precision-5000CVD system. The system has the parallel plate mixed-frequency PE-CVD configuration. TEOS is fed into process chamber employing He gas as a carrier. In general, (TEOS+He)/ O_2 flow rate ratio, high frequency power, pressure, susceptor temperature and electrodes spacing were fixed throughout the experiments.

3. Results and Discussion

3.1 Optimization of PE-TEOS/O₂/C₂F₆ process

Table 1 shows that the gap filling capability strongly depends on the C_2F_6/O_2 flow rate ratio and low frequency/high frequency (LF/HF) power ratio.

LF/HF Ratio	0.75	1.13	1.75	2.75
0.14			VOID	VOID
0.29		VOID	VOID FREE	VOID FREE
0.57	VOID	VOID FREE	VOID FREE	
0.86	VOID	VOID FREE	·	-
1.14	VOID FREE	VOID		

Table 1 Gap filling capability. (Aspect ratio=2.0)

The film characteristics such as F concentration, stress and wet etch rate are also affected by the C_2F_6/O_2 flow rate ratio and LF/HF power ratio as shown in Figs.1-3.





Fig.1 FT-IR spectrum as a function of a) C₂F₆/O₂ flow rate ratio, b) LF/HF power ratio.



Fig.2 Film stress as the function of C₂F₆/O₂ flow rate ratio and LF/HF power ratio.



Fig.3 Wet etch rate as the function of C_2F_6/O_2 flow rate ratio and LF/HF power ratio.

There exists an optimal C_2F_6/O_2 flow rate ratio and LF/HF power ratio. In the following experiments, we consider $C_2F_6/O_2=0.57$ and LF/HF=0.89 as the optimal process condition.

3.2 Characteristics of PE-TEOS/O₂/C₂F₆ CVD film

Fig.4 shows the SIMS spectrum for F concentration and depth profile analysis in PE-TEOS/O₂/ C_2F_6 CVD film.



Fig.4 SIMS spectrum for PE-TEOS/O₂/C₂F₆ CVD film.

Fig.5 shows the XPS spectrum for chemical bond analysis in PE-TEOS/ O_2/C_2F_6 CVD film.



Fig.5 XPS spectrum for PE-TEOS/O₂/C₂F₆ CVD film

It is obtained that approximately 7.4% of F atoms are uniformly distributed, and those F atoms are bonded to Si atoms in PE-TEOS/ O_2/C_2F_6 CVD film.

TDS spectrum for H_2O desorption analysis is shown in fig.6, and also, the film properties are summarized in table 2 as a comparison to Thermal-TEOS/O₃ CVD and PE-TEOS/O₂ CVD films.



Fig.6 TDS spectrum for PE-TEOS/O₂/C₂F₆ CVD film

Table 2 Properties	of PE-TEOS/	O_2/C_2F_6	CVD film.
--------------------	-------------	--------------	-----------

	PE-TEOS/O2	Th-TEOS/O3	PE-TEOS/O2/C2F6
Wet Erch Rate (/Thermal Ox.: 6:1 BOE)	2.0	3.6	4.7
Smess (E9Dyne/cm2)	1.3 Compressive	4.0 Tensile	0.2 Compressive
Refractive Index	1.46	1.45	1.41
Dielecteic Constant	4.2	5.3	3.9
Leak Current (A:3MV/cm)	7.0E-12	1.0E-09	4.0E-08

It is obvious that the PE-TEOS/O₂/C₂F₆ CVD film has such an strong advantage that dielectric constant is lowest among the 3 films compared. On the other hand, it has some disadvantages such as weakness in dielectric leakage and H₂O desorption comparing to the PE-TEOS/O₂ CVD film. In the next section, we present one approach to supplement these disadvantages.

3.3 Application of PE-TEOS/O2/C2F6 CVD film

One approach to supplement the poor film properties of Thermal-TEOS/O₃ CVD film is to form the stacked structure, which is practically used for device application of the Thermal-TEOS/O₃ CVD film. We applied the same concept for PE-TEOS/ O_2/C_2F_6 CVD film. An advantage of PE-TEOS/ O_2/C_2F_6 CVD process is to be able to combine non

C₂F₆ process without any requirement for hardware change.

Fig.7 shows the TDS spectrum for water desorption analysis of the PE-TEOS/O2~PE-TEOS/ O₂/C₂F₆~PE-TEOS/O₂ CVD films, triple layer stacked structure.



It was obtained that the H₂O desorption was greatly reduced by the stacked structure. It seems that the H₂O desorption spectrum traces that of PE-TEOS/O₂ CVD film at low temperature regime (<600 C) and PE-TEOS/ O_2/C_2F_6 at high temperature regime (>600 C), and as the result, total H₂O desorption is even lower than that of PE-TEOS/O2 CVD film.

Other film characteristics are summarized in table 3.

Table 3 Characteristics of PE-TEOS/O2~PE-TEOS/ O2/C2F6~PE-TEOS/O2 CVD film.

	PE-TEOS/O2	PE-TEOS/O2 Th-TEOS/O3 PE-TEOS/O2	PE-TEOS/O2 PE-TEOS/O2/C2F6 PE-TEOS/O2
Stress (E9Dyne/cm2)	1.3 Compressive	1.5 Tensile	0.6 Compressive
Dielecteic Constant	4.2	4.3	3.6
Leak Current (3MV/cm)	7.0E-12	1.3E-11	1.1E-11

The weakness in film stress and dielectric leakage is also supplemented by the stacked structure. Further more, the dielectric constant is even lower comparing to that of single PE-TEOS/ O_2/C_2F_6 CVD film.

As shown in fig.8, even with this stacked structure, we could achieve the gap filling of 0.7um height/0.3um width (aspect ratio>2.3) of metal to metal spacings on flat oxide surface.



Fig.8 SEM microphotographs of PE-TEOS/O2~PE-TEOS/O₂/C₂F₆~PE-TEOS/O₂ CVD film over metal lines on flat oxide surface

4. Summary

By utilizing C₂F₆ gas in PE-TEOS/O₂ CVD system, single step gap filling of 0.7um height/0.3 um width metal to metal spacings has demonstrated on AMAT Precision-5000CVD system. Gap filling capability of PE-TEOS/O2/C2F6 CVD process strongly depend on the C2F6/O2 flow Rate ratio and LF/HF power ratio. PE-TEOS/O2/C2F6 CVD film contains approximately 7.4% of F atoms, and the F atoms are bonded to the Si atoms and uniformly distributed in the film. As the result, the film has such a strong advantage as lower dielectric constant. On the other hand, the film is weak in dielectric leakage and H₂O desorption. However, those weakness is supplemented by PE-TEOS/O2~PE-TEOS/O2/ C₂F₆~PE-TEOS/O₂ CVD stacked film structure. The stacked structure also further enhances lowering the dielectric constant, and still remain good gap filling capability.

5. References

- C.Chiang, N.Y.Lam, J.K.Chu, N.Cox, 1) D.Fraser, J.Bozarrth and B.Mumford: Proc. IEEE VMIC Conf., Santa Clara, 1987, p.404.
- J.D.Romero, M.Khan, H.Futemi and J.Turlo: 2) J. Mater. Res., 6, 1996 (1991). N.Lifshitz, W.Y.C.Lai and G.Smolinsky: IEEE
- 3) Electron Device Lett., 10, 562 (1989).
- Y.Numazawa: Semiconductor World, 3, 185 4) (1991).
- 6) H.Kotani, M.Mtsuura, A.Fujii, H.Genjou and S.Nagao: Dig. Int. Electron Devices Meet., Washington D.C., 1989, p.669.
- 7) Y.Nishimoto, N.Tokumasu, T.Fukuyama and K.Maeda: Ext. Abstr. 19th Conf. SSDM, Tokyo, 1987, p447.
- 8) K.Fujino, Y.Nishimoto, N.Tokumatsu and K.Maeda: J. Electrochem. Soc., 137, 2883 (1990).
- M.Matsuura, H.Kotani and H.Abe: Ext. Abstr. 9) 22nd Conf. SSDM, Sendai, 1990, p.239.
- M.Matsuura, Y.Hayashida, H.Kotani and 10)
- H.Abe: Jpn. J. Appl. Phys., 30, 1530 (1991). D.A.Webb, A.P.Lane and T.E.Tang: Proc. 2nd 11) Int'l ULSI Science and Tech. Symp., 1989, p.571.
- 12) D.Yu, D.Favreau, E.Martin and A.Manocha: Proc. IEEE VMIC Conf., Santa Clara, 1990, p.166.