Extended Abstracts of the 1994 International Conference on Solid State Devices and Materials, Yokohama, 1994, pp. 556-558

Study on the Fundamental Electrical Properties of Ultra-Thin Oxides Grown by Low Temperature Microwave Plasma Afterglow Oxidation

P. C. Chen, Klaus Y. J. Hsu, J. Y. Lin* and H. L. Hwang Department of Electrical Engineeering, National Tsing Hua University, Hsin-chu, Taiwan 30043, R.O. C. *Department of Electrical Engineering, Chung Cheng Institute of Technology, Ta-shi, Tao-yuan, Taiwan, R. O. C.

Fundamental characteristics such as the oxide breakdown fields, oxide charges and interface state density of various ultra-thin oxides (≤ 8 nm) grown by microwave plasma afterglow oxidation at low temperatures were investigated. The effective oxide charge density of 600 °C as-grown oxide was as low as 6×10¹⁰ cm⁻². The breakdown fields of the oxides were further enhanced and the interface state densities were reduced by employing fluorination (HF soaked) and low temperature N₂O plasma annealing.

1. INTRODUCTION

When ultra-large-scale integrated circuits (ULSI) miniaturization enters the deep submicron domain, low temperature indispensable processes are to maintain the shallow dopant profiles and reduce the strain in oxide films during the thin oxide growth. The utilization of low-temperature plasma (induced by RF or microwave) oxidation process can satisfy these requirements^{1,2)}.

In this work, microwave plasma afterglow oxidation was utilized to grow ultra-thin oxides (7-8 nm) at 400 °C and 600 °C. The fluorine addition and the low temperature N₂O plasma annealing were performed for the first time in such system. The electrical properties of these ultrathin oxides were investigated. Some properties of the oxides in this study are superior to those reported previously by the same method³.

2. EXPERIMENTAL

The schematic diagram of the experimental apparatus is illustrated in Fig. 1. The forward microwave power was 100 W and the oxygen pressure was 1 torr. To fabricate fluorinated oxides, RCA



Fig. 1 The schematic diagram of the experimental apparatus.

cleaning was followed by soaking in HF solution (DI water:HF = 10:1), and without DI water rinse the samples were directly loaded into the oxidation tube to grow oxides. The N₂O plasma annealed samples were first grown at 600 °C in 1 torr O₂ plasma to a thickness of 6.5 nm. The growth were immediately followed by a 100 W microwave annealing in 3 torr N₂O plasma for 15, 30 and 60 min.

3. RESULTS AND DISCUSSION

The oxide thickness (d), $V_{\rm fb}$ and $Q_{\rm eff}$ of the as-grown and fluorinated oxides are listed in Table I. The addition of fluorine

led to more positive charges in those oxides. The breakdown field of the fluorinated oxides grown at 600 °C could be improved to 11 MV/cm.

Table I Thickness, flat-band voltage and oxide charges of asgrown and fluorinated oxides

Sample	thick-	Vfb	Qeff
prepa-	ness	(V)	$(\#/cm^2)$
ration	(d)		
600 °C	8	-0.35	5.8×10^{10}
400 °C	7	-0.36	8.8×10 ¹⁰
600 °	7.2	-0.37	1.9×10 ¹¹
C+F			
400 °	7	-0.39	2×10 ¹¹
C+F			,

Figure 2 shows the cumulative failure rate as a function of the electric fields applied to the asgrown and fluorinated oxides. Fluorinated oxides grown at 600 °C had only 20% failure rate at an electric field of 7 MV/cm while others were almost all brokedown.



Fig. 2 The cumulative failure rates of the as-grown and fluorinated oxides.

Figure 3 shows the interface state density (D_{it}) of the samples listed Table I. And the fluorinated in process can reduce D_{it} throughout the whole silicon bandgap. The D_{it} at the mid-bandgap (Ditm) of the fluorinated oxide grown at 400 °C was still high, but the Dit near the valence band edge was significantly reduced. The reduction of D_{it} became °C 600 efficient for the more 600°C (35%). When samples fluorinated oxides was put through the poly-gate fabrication process,



densities of the samples listed in Table I.

it was found that the interface state density was further reduced to 5×10^{10} cm⁻²eV⁻¹. Fig. 4 shows the interface state density of the 600 ° C fluorinated oxide with n⁺-poly electrode.



Fig. 4 The interface state density of the 600 °C fluorinated oxide with n⁺-poly electrode.

Table II Thickness, flat-band voltage and oxide charges of 600 °C 3 torr N₂O plasma annealing oxides

4 97		77	
N ₂ O anneal- ing time(mi ns)	thick- ness (d) (nm)	V _{fb} (V)	Q _{eff} (#/cm ²)
0 (as- grown)	8	-0.35	5.8×10 ¹⁰
15 30 60	6.7 7.7 8	-0.44 -0.44 -0.47	3.9×10 ¹¹ 3.3×10 ¹¹ 3.8×10 ¹¹

For the N₂O plasma annealed samples, the annealing condition, oxide thickness, V_{fb} and Q_{eff} are listed in Table II. N₂O plasma annealed oxides have higher Qeff than those of the as-grown oxides. The highest breakdown field of the 15 min N₂O plasma annealed oxide was the 12 MV/cm. Fig. 5 compares cumulative failure rate of the N20 As plasma annealed oxides. the



Fig. 5 The cumulative failure rates of the 600 °C as-grown and N_2O plasma annealed oxides.

electric field > 7 MV/cm, the 600 °C as-grown oxides almost all failed(> 90%) and 15 min annealed oxide only had a 20% failure, the 30 min and 60 min annealed oxides failure rates were about 28%-30%. Figure 6



Fig. 6 The interface state densities of the 600 °C as-grown and N₂O plasma annealed oxides.

interface illustrates the state N_2O plasma annealed density of oxides. The D_{it} near the valence band of the oxide annealed for 15 min was lower than that of the asgrown oxide, but the Ditm was higher. When the annealing time was prolonged to 60 min, the D_{it} became low again, but Ditm was still the same as that of the as-grown oxide.

4. CONCLUSIONS

This study is the first attempt inherent to investigate the of the aselectrical properties grown, fluorinated and N₂O plasma annealed ultra-thin oxides (7-8 nm) grown by 100 W microwave afterglow oxidation at low temperatures (400 ° C and 600 °C). The fluorine and nitrogen introduced further positive charges into the thin oxides. The fluorinated process at 600 °C could improve the oxide breakdown property and reduce the interface traps. This provides an appealing option for the oxidation in ULSI.

REFERENCES:

- J. Siejka and J. Perriere, Phys. of Thin Films, <u>14</u>(1989) 81.
- 2) T. Sugano, Mat. Res. Soc. Sym. Proc., 38(1985) 487
- 3) Y. Nishioka, E. F. da Silva Jr., Y. Wang and T. P. Ma, IEEE Electron Device Lett., 9(1988) 38.