

Influences of Impurities on Oxidation Processes of Si(100) Substrates

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1. Introduction

With reduction of device dimensions in ultra large-scale integrated circuits (ULSIs), understanding and precise control of oxidation on Si surfaces becomes strikingly important since the electrical characteristics of metal-oxide-semiconductor field-effect transistors (MOSFETs) are seriously ruled by the roughness at the SiO₂/Si interfaces and SiO₂ bonding structures. Especially, the impurity diffusion from the gate poly-Si to the gate oxide film has been pointed as an important factor governing the ultra-thin gate-oxide reliability [1]. Therefore, it is essential to clarify the influences of impurity atoms on SiO₂ network structures. It has been also reported that the oxidation rate of Si substrates is strongly governed by the impurity concentration in the Si substrate [2]. It can be expected that impurities are made use of controlling the oxidation reaction of Si surfaces and the atomistic structures of the SiO₂ films and the SiO₂/Si interfaces.

We have studied the effect of H atoms on the oxidation processes and SiO₂ bonding structures on Si(100) surfaces by high-resolution electron energy loss spectroscopy (HREELS) [3-5]. In the present study, we have investigated the influence of the impurity atoms on the oxidation processes of D-terminated Si(100) surfaces and on the relaxation of SiO₂ structures.

2. Experimental

Experiments in the present study were carried out in an ultrahigh-vacuum (UHV) chamber, equipped with apparatuses of HREELS, low-energy electron diffraction (LEED) and Auger electron spectroscopy (AES). The base pressure of the UHV chamber was less than 3×10^{-10} Torr. Atomic D and O were produced using a W filament heated at 1500°C.

The B concentrations of p- and p⁺-Si(100) substrates used were 4×10^{15} and 1×10^{19} cm⁻³, respectively, and the P concentration of n-Si(100) substrates was 1×10^{15} cm⁻³. The D-saturated Si surfaces were prepared by atomic D exposure after thermally cleaning Si substrates at 1200°C. The oxidation was performed by exposing the Si surfaces to atomic O at room temperature. Oxide film thickness is defined as the ratio of the number of O atoms to that of surface Si atoms, determined by AES measurements. HREELS measurements were performed at room temperature under a specular reflection condition.

3. Results and discussion

Figure 1 shows HREELS spectra of D-terminated n-type Si(100) surfaces before and after exposure to atomic O at room temperature. Energy loss peaks for Si-2D species are observed in the spectrum of as-terminated

Si(100) surfaces at about 60, 82 and 190 meV, which correspond to wagging (vD₁), scissors (vD₂) and stretching (vD₃) modes, respectively [6]. The LEED observation confirms that this surface has a 1x1 structure. Therefore, it can be concluded that a dideutride structure is formed on the D-terminated Si(100) surface. Moreover, the atomic O exposure leads to the appearance of two loss peaks at about 50 and 130 meV, corresponding to a symmetric bending (vO₁) and an asymmetric stretching (vO₂) mode, respectively [7]. It is found from Fig. 1 that the peak of the asymmetric stretching mode (vO₂) shifts to higher energy losses with the progress of oxidation. This behavior is the same as the case of H-terminated p-type Si(100) surfaces [4].

The energy-loss changes of (a) the symmetric bending (vO₁) and (b) the asymmetric stretching (vO₂) modes for n-, p- and p⁺-Si(100) substrates are shown in Fig. 2, as a function of SiO₂ film thickness. From Fig. 2(a), the energy loss of the vO₁ peak hardly depends on the SiO₂ thickness above 1 ML and there are few differences among the types of Si substrates. On the other hand, in the case of vO₂, the energy loss keeps constant up to about 0.8 ML and then increases with an increase in SiO₂ thickness for the whole substrate types. This behavior in the vO₂ energy loss for the oxide thickness indicates that there is a two-step process in the initial oxidation [4]: O atoms preferentially adsorb on one of the two back-bond sites of a surface Si atom until the oxide thickness is 0.8 ML and in the sequential oxidation process, other back-bond sites become occupied by O atoms [4]. In the second oxidation step, the O adsorption on Si back bonds is accompanied with the structural relaxation of Si-O-Si bonds. Estimating the force constants of Si-O bonds and the Si-O-Si bond angles from the energy losses of vO₁ and vO₂ in Fig. 2 using a central-force network model [8], the Si-O-Si structural relaxation in the second step was found to originate mainly from the change in the force constant.

It is found from Fig. 2(b) that the Si-O-Si structural relaxation on the p⁺-Si surface is hard to occur compared with that on the p- and n-Si surfaces. The surface coverage of B atoms on the p⁺-Si substrate was evaluated to be 0.08 ML (4×10^{20} cm⁻³) by AES measurements, which results from the surface segregation of B atoms by annealing. The coverage of 0.08 ML means that about 8% of the adsorbed O atoms are bonding with B atoms. Taking account of the fact that the atomic radius of B atoms is smaller than that of Si atoms, compressive stress induced by the formation of Si-O-Si bonds may be diminished by the existence of B atoms. This explanation is consistent with the report that impurity

atoms from the gate poly-Si film relax the compressive stress near the SiO₂/Si interface [9].

Figure 3 shows the relationship between the oxide thickness and the amount of O₂ exposure, where the dissociative ratio of O₂ by the W filament heated at 1500°C was estimated to be 0.12%. The oxidation rate on D-terminated Si surfaces by O atoms is hardly influenced by the impurity. This finding is thought to be consistent with the reports that the dissociation of O₂ occurs through the electron transfer from the substrate to O₂ molecules [2] and the dissociation sites of O₂ such as dangling bonds vanish by the termination with H [3].

4. Conclusions

We have investigated the influences of substrate impurities on the initial oxidation processes and the relaxation of Si-O-Si structures on D-terminated Si(100) surfaces in the oxidation using O atoms at room temperature by HREELS. It is found that the impurities hardly influence the oxidation rate and the preferential adsorption of O atoms on one of two back bonds of a surface Si atom at oxide thicknesses below 0.8 ML. Moreover, B atoms influence the relaxation of Si-O-Si structures during oxidation, which may be related to the compressive stress in Si-O-Si structures.

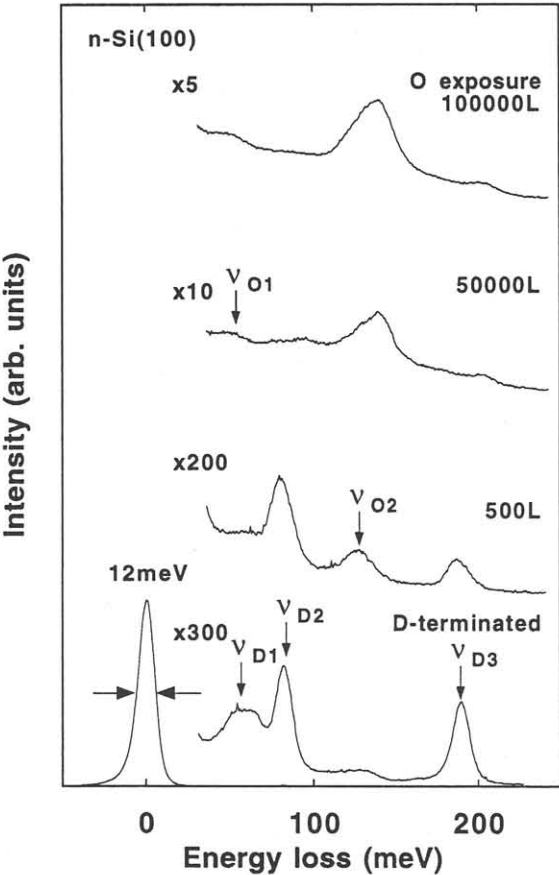


Fig. 1 HREELS spectra of D-terminated n-type Si(100) surfaces before and after O exposure at room temperature.

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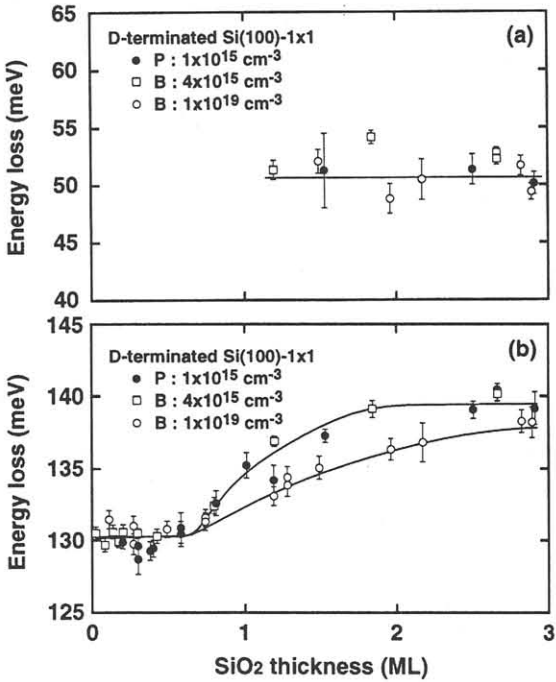


Fig. 2 Energy losses of (a) the symmetric bending and (b) the asymmetric stretching modes for n-, p- and p⁺-Si(100) substrates as a function of oxide thickness.

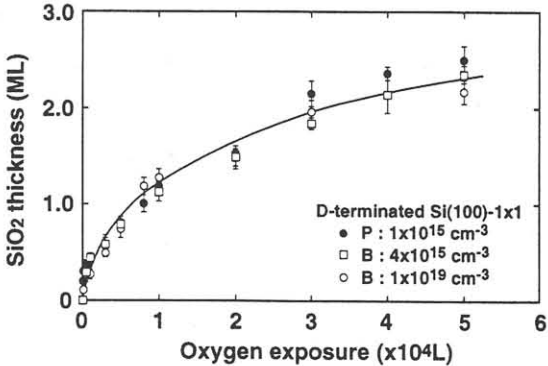


Fig. 3 Relationship between the oxide thickness and the amount of O₂ exposure.