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Ultrathin Fluorinated Silicon Nitride Gate Dielectric Films Formed by Plasma Enhanced Chemical Vapor Deposition Employing NH₃ and SiF₄

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1.Introduction

The silicon nitride (SiN_x) film attracts much attention as scaled gate dielectric films in next generation's ULSI¹). However, the conventional SiN_x film has a poor interface with silicon and is leaky due to a high trap density in the film.

Recently, we have developed ultrathin fluorinated SiN_x films formed by ECR-PECVD employing NH₃/SiF₄. It is known that the average bond energy (5.73eV) of Si-F is higher than that of Si-H (3.18eV)². Therefore, it is expected that the Si-F bond in the film should have improved the quality of gate dielectric film.

In this study, we have investigated properties of ultrathin SiN_x films (4nm) formed at 350°C. This film (fluorinated SiN_x films) contains fewer hydrogen atoms than the conventional SiN_x films formed by ECR-PECVD employing NH_3/SiH_4 . This fluorinated SiN_x film reduced successfully the leakage current by several orders of magnitude than the thermal SiO_2 in the identical EOT. These film properties and the surface reactions for the SiN_x film formation with good quality are discussed on the basis of results of the *in-situ* XPS, *in-situ* FT-IR RAS, FT-IR, GDS and TDS.

2.Experimental

Figure 1 shows a schematic diagram of the experimental apparatus of typical ECR-PECVD system with a divergence magnetic field used in this study³). SiN_x films were formed on n-type (100) silicon substrates. Silicon substrates for *in-situ* XPS were cleaned by HF (HF:H₂O=1:10) solution at room temperature before deposition.

In this system, the process chamber is equipped with *in-situ* FT-IR RAS. The FT-IR RAS has been applied for *in-situ* observation of the growth process of SiN_x films under ECR-PECVD conditions employing NH₃/SiH₄ and NH₃/SiF₄. Furthermore, the XPS system is connected to the ECR-PECVD chamber through a transfer chamber in a vacuum.

The electrical properties of the SiN_x films were evaluated with making an Al/SiN_x/Si diode structures of electrode area of 3.98mm². In these samples, no post metal annealing treatments were performed. The leakage current density-voltage (J-V) characteristics were measured by using a semiconductor parameter analyzer (Hewlett-Packard : 4156B).

3.Results and Discussion

Figure 2 shows the leakage currents measured in SiN_x films formed employing NH_y/SiH_4 , NH_y/SiF_4 and the conventional thermal SiO_2 films whose equivalent oxide thickness (EOT) is about 2.8nm⁴). It was found that the leakage currents were drastically decreased by employing NH₃/SiF₄. Particularly, J-V characteristics exhibit the minimum leakage current in NH₃/SiF₄ (50/20sccm). The fluorinated SiN_x film formed by NH₃/SiF₄ reduced the leakage current by several orders of magnitude than the thermal SiO₂ in the identical EOT.

Figure 3 shows FT-IR RAS results of normalized absorption intensity of Si-N bonds and fluorine concentration determined from *in-situ* XPS, as a function of SiF₄ flow rate. With increasing the SiF₄ flow rate, the normalized absorption intensity of Si-N bonds decreased and the fluorine concentration had the tendency of saturation at flow rates above 20sccm. These results suggest that surplus fluorines contribute to etching of SiN_x films. The conditions of NH_y/SiH₄ PECVD were optimized for obtaining near stoichiometry. The typical condition in NH₃/SiF₄ PECVD was as follows; a total pressure : 0.5Pa, gas flow rate : NH₃/SiF₄=50/20sccm, a microwave power : 300W, a substrate bias : floating, a substrate temperature : 350°C.

Figure 4 show C-V curve of (a) SiN_x film formed employing NH₃/SiH₄ and (b) fluorinated SiN_x film formed employing NH₃/SiF₄. There were the hump and hysteresis (0.04V) attributed to charge traps in the SiN_x film observed in C-V data of SiN_x films formed employing NH₃/SiH₄. On the other hand, the hump was not observed in C-V data of fluorinated SiN_x films formed employing NH₃/SiF₄ and the excellent hysteresis (0.02V) was achieved with employing NH₃/SiF₄. These results suggest that the state trapping density of SiN_x films formed employing NH₃/SiF₄. The dielectric constant of both SiN_x films were almost the same value. We considered that these value were dependent on the composition of SiN_x films. N/Si ratio of fluorinated SiN_x films was about 0.95 (Si-rich).

Figure 5 shows the FT-IR spectra of (a) SiN_x film formed employing NH₃/SiH₄ and (b) fluorinated SiN_x film formed employing NH₃/SiF₄, which exhibit the Si-N bonds. The hydrogen concentration of as-deposited fluorinated SiN_x film, as evaluated by the N-H and Si-H signals, is low as compared to the SiN_x film formed employing NH₃/SiH₄. Furthermore, the hydrogen concentration was determined from glow discharge spectroscopy (GDS) (Rigaku : GDS3870) and indicated the similar tendency of FT-IR results.

Figure 6 shows the thermal desorption mass spectroscopy (TDS) of fluorinated SiN_x film formed employing NH_3/SiF_4 . The amount of desorption gas of H_2 was observed above 350°C. However, F and F_2

were not observed. This fact suggests that fluorine bonds in fluorinated SiN_x film are stable.

4.Conclusion

We have clarified effects of fluorines in the SiN_x films for improvement of MIS characteristics. As a result, the control of fluorine concentration in the SiN_x films was found to be a key factor for forming the fluorinated SiN_x films with high quality at low temperatures. No hump, excellent hysteresis and low leakage currents were achieved in MIS with fluorinated SiN_x films formed employing NH₃/SiF₄ (50/20sccm). The fluorinated SiN_x is very effective for ultrathin gate dielectric films in next generation's ULSI.



Fig.1 Schematic diagram of the experimental apparatus of typical ECR-PECVD system.



Fig.2 J-V characteristics with SiN_x films employing NH_3/SiH_4 , $NH_3/SiF_4(50/3, 50/20, 50/40sccm)$ and the conventional thermal SiO_2 .



Fig.3 FT-IR RAS results of normalized absorption intensity of Si-N bonds and fluorine concentration, as a function of SiF_4 flow rate.

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Fig.4 C-V curve of (a) SiN_x film formed employing NH_ySiH_4 and (b) fluorinated SiN_x film formed employing NH_ySiF_4 .



Fig.5 FT-IR spectra of (a) SiN_x film formed employing NH_3/SiH_4 and (b) fluorinated SiN_x film formed employing NH_3/SiF_4 .



Fig.6 Thermal desorption mass spectroscopy (TDS) results of fluorinated SiN_x film formed employing NH_ySiF_4 .