Study on Zr-Silicate Interfacial Layer of ZrO₂-MIS Structure Fabricated by Pulsed Laser Ablation Deposition Method

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1. Introduction

The ZrO_2 high-k dielectric has been investigated as a view to its application as a replacement for SiO₂ in sub-100 nm CMOS technology; however the unavoidable appearance of the interfacial layers between the high-k gate dielectrics and Si substrate have been reported [1]. Though the dielectric properties of interfacial layer are important for the sub-2 nm equivalent oxide thickness (Teq), required for the advanced CMOS technology, these properties have not been clarified yet.

In this study, the properties of the interfacial layer of ZrO_2 -MIS structures fabricated by the Pulsed Laser ablation Deposition (PLD) method [2], which is a method controllable independently of deposition parameters, i.e., substrate temperature and gas pressure, were investigated. For the first time, we successfully evaluated the dielectric properties of the interfacial layers directly, which were confirmed to be Zr-silicate structures by physical analysis.

2. Experimental

The ZrO₂ films were deposited on the Si substrates with diluted HF treatments by the PLD method at various substrate temperatures between 200 $^{\circ}$ C and 800 $^{\circ}$ C. O₂ gas pressure was 0.2 Torr. ZrO₂-MIS structures with the various ZrO₂ thicknesses with the same thermal budgets are also fabricated, in order to evaluate the dielectric properties of interfacial layers and ZrO₂ layers separately.

3. Results and discussion

A cross-sectional TEM image of the interface of ZrO2-MIS capacitor is shown in Fig.1. This image clearly shows the formation between ZrO₂ and Si substrate of the amorphous interfacial layer having a thinness of as little as 1nm. The C-V and J-V characteristics of this capacitor are shown in Fig. 2. Teq is about 1.3 nm estimated from the accumulation capacitance measured by the two-frequency C-V method [3]. This ZrO₂ MIS exhibits the negligible frequency dispersion between 50 kHz and 600 kHz and negligible hysteresis (Fig. 3). The leakage current density is about 10^{-2} A/cm² at the gate voltage of -2V, which is much smaller than that of SiO₂ film of the same Teq $(J_{SiO2} \sim 10^2 \text{A/cm}^2)$. However, since the physical thickness of the interfacial layer occupies more than half of the equivalent thickness, it is unquestionable that dielectric properties of these interfacial layers are so critical for the realization of the thinner dielectrics.

Firstly we investigated the structural properties of interfacial layer by using the EDX and XPS analysis. Zr atoms were detected in the region close to the Si surface (Fig. 3). The average composition of Zr in the interfacial layer of $4\sim5$ at% was obtained from both of the EDX and XPS analysis. Figure 4 shows the binding energy profiles of the

interfacial layer that was remained by the etching of ZrO_2 layer. For the comparison, the results of ZrO_2 and SiO_2 are also shown. Whereas the Zr-Si bonds were not detected, Zr-O and Si-O bonds of the interfacial layer were detected at the same binding energy with ZrO_2 and SiO_2 , respectively. Thus, we concluded that the interfacial layer of ZrO_2 -MIS capacitor is Zr-silicate layer.

Next, we evaluated the dielectric constants (ɛ) of the silicate layer and ZrO₂ layer separately, by using the MIS capacitors with various ZrO2 thicknesses, for a given silicate thickness that were confirmed by TEM observations. The schematic image of this evaluation method is shown in Fig. 5. The slope shows the $1/\epsilon_{ZrO2}$, and $\epsilon_{Silicate}$ is evaluated from the intercept value of 1/C. Then, using these methods, the effects of deposition temperatures on each dielectric constant have been investigated. Figure 6 shows the dielectric constants of the MIS capacitors fabricated at the different deposition temperature and with the equal thermal budgets. The ϵ_{ZrO2} are almost the same, however, $\epsilon_{silicate}$ of MIS capacitor fabricated at 200 °C (~5) is much lower than that of MIS capacitor fabricated at 600 °C (~9), even if the thermal budget is identical. This fact means that the dielectric constant of the silicate layer of MIS capacitor is sensitive to the ZrO₂ deposition temperature.

Finally, we report the direct evaluation of the dielectric properties of the interfacial silicate layer of MIS capacitor. The ZrO₂ layer of MIS capacitor was removed by the HF etching after the fabrication of ZrO2-MIS capacitor at 500 °C. Due to the difference in etching rate between ZrO_2 and the silicate layer (Fig. 7), only the silicate layer remained. In order to recover the etching damage, the remained silicate layer was annealed at 600 °C for 5 min. in dry O2 atmosphere. Figure 8 shows that the remained silicate film of as thin as 1.9nm after annealing. The C-V characteristic of the Zr-silicate MIS capacitor is shown in Fig. 9. This silicate MIS capacitor also exhibits the negligible frequency dispersion and hysteresis. Teq is 0.9 nm obtained from the accumulation capacitance of the two-frequency C-V method. The value of $\varepsilon_{\text{Silicate}}$ is estimated to be 8 that is slightly smaller than the Esilicate evaluated from the above method (Fig. 6), which may be due to the recovering annealing. Thus, we could evaluate only the ultra-thin interfacial silicate layer of ZrO2-MIS capacitor directly.

4. Conclusion

. We successfully investigated both the physical and the dielectric properties of only the interfacial Zr-silicate layer of ZrO₂-MIS capacitor. The ultra-thin interfacial Zr-silicate MIS capacitor with Teq of 0.9 nm and ϵ of 8 was successfully evaluated. It was also clarified that the dielectric

constant of the interfacial silicate layer of ZrO2-MIS capacitor depends on the deposition temperature even with keeping the thermal budgets to be equal.

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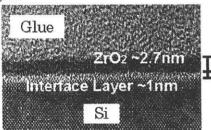


Fig.1 Cross-sectional TEM image of ZrO2-MIS capacitor fabricated at 500 °C, 5 min., in O2.

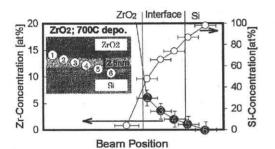


Fig. 3 Composition profiles in the interfacial layer of ZrO2-MIS capacitor fabricated at 700°C detected by EDX. Inset numbers correspond to the scanning beam position of EDX.

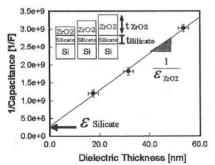


Fig. 5 Schematic image of the method of evaluating & ZrO2 and & Silicate, separately.

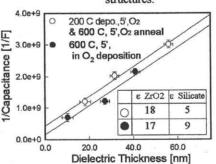
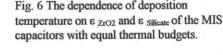


Fig. 6 The dependence of deposition



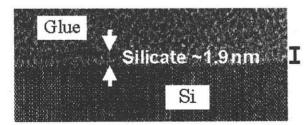
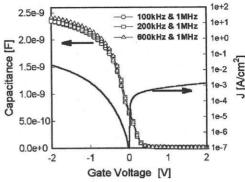


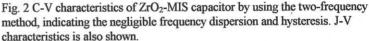
Fig. 8 Cross-sectional TEM image of the remained interfacial silicate after etching of ZrO2 layer that was deposited at 500 °C, which are annealed at 600 °C, 5 min., in O2.

Reference

[1] W.J.Qi et al., IEDM Tech. Dig., p145 (1999).

[2] Y.Oishi et.al., Jpn. J. Appl. Phys. Vol. 35, p1242 (1996). [3] K. J. Yang et.al., IEEE Trans. Electron Devices, ED-46, p1500 (1999).





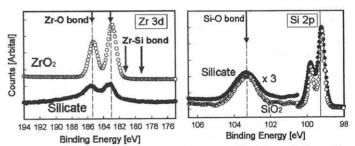


Fig. 4 XPS spectra of the interfacial layer remained through the etching of ZrO2 layer, indicating the suitable bonding structures to the silicate structures.

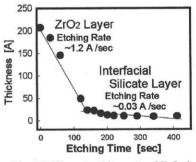


Fig. 7 Different etching rate of ZrO₂ layer and the silicate layer to enable only the interfacial silicate layer to be remained.

