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Atomic-Scale Depth Profiling of Oxides/Si(111) and Oxynitrides/Si(100) Interface

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1. Introduction

insulator in MOSFET operated at room temperature[1], consists mostly of structural transition layer[2], the quality of which must be improved using atomic oxygen. Our studies on the oxidation mechanism of Si(111) surface using atomic oxygen revealed that the oxidation rate becomes extremely small if the amount of Si²⁺ and Si³⁴ localized at the interface takes their maximum value[3]. However, the mechanism for this phenomena is not clarified yet.

The silicon oxynitrides are also important as ultrathin gate dielectrics in future MOSFETs due to their improved reliability and ability to suppress boron penetration. wet oxidation followed by the annealing in NO ambient revealed that the most of nitrogen atoms localize near the interface and have bonding with two Si atoms at the interface and one Si atom, spectral intensities NO arising from SiO2, those NI arising from which has bonding with three oxygen atoms in the overlayer.[4] However, the effect of nitridation on the distribution of intermediate oxidation states of Si is not clarified yet.

It is the purpose of the present paper to demonstrate the atomic-scale depth profiling of the chemical structures of the SiO2/Si(111) interface and oxynitrides/Si(100) interface structures by applying the maximum entropy method [5,6] to the angle resolved Si 2p photoelectron spectra under the consideration of the elastic scattering of Si 2p photoelectrons in silicon oxide films.[7]

2. Experimental Details

The oxide films studied were prepared as follows. The atomically flat hydrogen terminated Si(111) surfaces were oxidized at 400°C using atomic oxygen produced in microwave (5.0 W/cm2 at 2.45 GHz) excited Kr/O2 mixture(Kr/O2 = 97/3, 1 Torr) gas plasma. Oxynitride films containing maximum nitrogen concentrations of 1, 2, 3, 5 and 6 at.% were grown on p-type Si(100) by the wet oxidation followed by the annealing in NO ambient at 900°C.

Angle resolved Si 2p spectra excited by AlKa radiation arising from these films were measured with photoelectron acceptance angle of 3.3° at the entrance of electron energy analyzer. using high resolution and highly sensitive ESCA-300[8].

3. Experimental Results and Discussions

Figure 1 shows the dependence of Si 2p photoelectron spectral intensities NI arising from intermediate oxidation states Si1+, Si2+, Si3+ and those NO arising from SiO2 in silicon oxide films formed using atomic oxygen normalized by Si 2p spectral intensities NS arising from Si substrate on photoelectron take-off angle for the interface containing maximum amount of Si³⁺, Si¹⁺ Figure 2 shows Gaussian distribution of the amount of and Si²¹ Si^{1+} , Si^{2+} , Si^{3+} and SiO_2 obtained by applying maximum entropy method to Fig. 1. The solid, dotted and dashed curves in Fig. 1 are calculated using the depth profile in Fig. 2. According to Fig. 2, the interface containing the maximum amount of Si¹⁺ is extremely abrupt than the interface containing the maximum amount of Si³⁺ or Si²⁺. This must be the reason for the

observation of extremely small oxidation rate at the interface 0.8-nm-thick silicon oxide, which was used as gate containing the maximum amount of Si³⁺ or Si²

Figure 3 shows the dependence of Si 2p photoelectron spectral intensities NI arising from Si¹⁺, Si²⁺, Si³⁺ and those NO arising from SiO2 in silicon oxide films formed using molecular oxygen normalized by Si 2p spectral intensities NS arising from Si substrate on photoelectron take-off angle for interface containing maximum amount of Si^{3+} and Si^{1+} . Figure 4 shows Gaussian distribution of the amount of Si^{1+} , Si^{2+} , Si^{3+} and SiO_2 obtained by applying maximum entropy method to Fig. 3. The solid, dotted and dashed curves in Fig. 3 are calculated using the Our studies on the depth profile in Fig. 4. It is found from the comparison between chemical structures of oxynitrides/Si(100) interface formed by the Figs. 2 and 4 that the interfaces formed using atomic oxygen are more abrupt than the interfaces formed using molecular oxygen.

> Figure 5 shows the dependence of Si 2p photoelectron intermediate oxidation states SiOx and those NN arising from oxynitrides SiOxNy normalized by Si 2p spectral intensities NS arising from Si substrate on photoelectron take-off angle for oxynitiride containing maximum nitrogen concentrations of 3 and 6 at.%. Figure 6 shows Gaussian distribution of the amount of $SiO_2,\,SiO_x$ and SiO_xN_y obtained by applying maximum entropy method to Fig. 5. The solid, dotted and dashed curves in Fig. 5 are calculated using the depth profile in Fig. 6. According to Fig. 6, the amount of intermediate oxidation states of Si is weakly affected by the incorporation of nitrogen atoms at SiO2/Si interface.

4. Conclusion

In conclusion, the atomic-scale depth profiling of SiO₂/Si(111) interface and silicon oxynitrides/Si(100) interface structures are demonstrated for the first time by applying maximum entropy method to the angle-resolved Si 2p spectral intensities arising from intermediate oxidation states of Si and those arising from silicon oxynitrides normalized by Si 2p spectral intensities arising from Si substrate under the consideration of the elastic scattering of Si 2p photoelectrons in silicon oxide films. The present depth profiling analysis must be effective for the study on the high-k dielectrics/Si interface structures.

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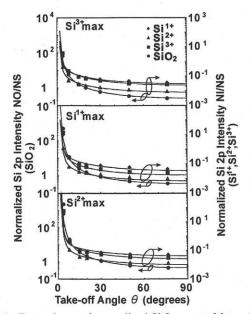
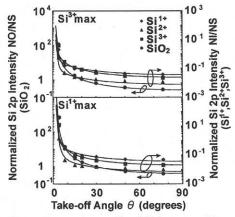
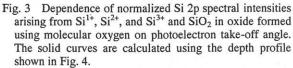


Fig. 1 Dependence of normalized Si 2p spectral intensities arising from Si¹⁺, Si²⁺, and Si³⁺ and SiO₂ in oxide formed using atomic oxygen on photoelectron take-off angle. The solid curves are calculated using the depth profile in shown in Fig. 2.





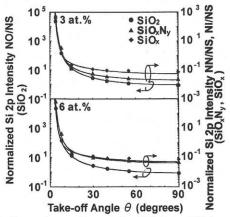


Fig. 5 Dependence of normalized Si 2p and N 1s spectral intensities arising from SiO_x , SiO_xN_y and SiO_2 in oxynitrides on photoelectron take-off angle. The solid curves are calculated using depth profile shown in Fig. 6.

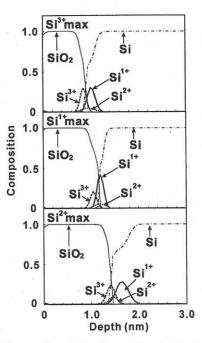


Fig. 2 Compositional depth profile obtained from the analysis of Fig. 1.

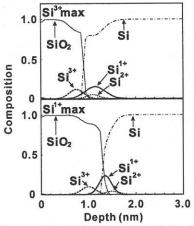


Fig. 4 Compositional depth profile obtained from the analysis of Fig. 3.

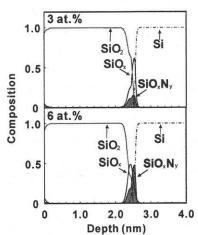


Fig. 6 Compositional depth profile obtained from the analysis of Fig. 5.