Low Temperature Solution Processed SiO₂ Insulator Thin Films for Organic FET

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1. Introduction

Research on organic field effect transistor (OFET) is rapidly progressive over the past few years because of their possibility to applications in low cost and large area electronic devices prepared by low temperature solution process. Recently, in order to take such advantage of OFET, approach into the use of the gate insulator prepared by low temperature solution process has been performed by using polymeric dielectric materials such as polyimide [1], polyvinyl phenol [2], polyvinyl alcohol [3]. On the other hand, though it was well known that the thermal oxidation thin film of silicon dioxide (SiO₂) was clearly superior for the gate insulator, it was difficult to prepare by low temperature solution process.

In this work, silazane compound was used as a precursor material for preparing a SiO₂ thin film by solution process. The electronic property of insulator thin film and OFET were investigated.

2. Experiments

Highly n-doped silicon wafers, whose resistivity was 1-10 Ωcm, was used as a substrate and electrode. Natural oxidation membrane on the silicon wafer was removed by the semiconductor cleaning solution SemiCoClean23 purchased from Furuuti chemical company. Silazane compound diluted to 20% g/g with dibutylether was coated on the silicon wafer by the spin-coating technique. The stepwise thermal treatments under UV irradiation (max. 373K) induced the conversion of the silazane compound into SiO₂. The thickness of obtained thin films was about 200-300 nm. Their XPS spectra were taken by Perkin Elmer PHI model 5600ci, in order to confirm the conversion of the coating films into SiO₂. The surface morphology of the thin films was estimated by SII SPA300 atomic force microscopy.

Gold round electrodes were deposited on the top of the SiO₂ thin film through a metal mask. The area of gold electrode was 7.85X10⁻³ cm². The I-V measurements were performed with a Keithley 4200-SCS parameter analyzer. The C-V measurement was performed by HP 4284A precision LCR meter at 1kHz. OFET with the SiO₂ thin film as a gate dielectric was fabricated using pentacene or P3HT as the semiconductive layer and the gold as the source and drain electrodes. The channel length was 20 μm and the channel width was 2 mm.

3. Results and discussion

Fig. 1 shows XPS spectra of the SiO₂ thin film on a Si substrate. Upper profile shows a XPS spectrum for a thermally oxidized SiO₂ and lower shows solution processed SiO₂ thin film fabricated in this work. Three typical peaks assigned to the photoelectron from O1s, Si2s and Si2p were observed at 535, 155 and 105 eV, respectively. The chemical composition was estimated to Si:O=1:2.09 for the solution processed SiO₂ thin film. There were no differences on their binding energy and peak intensity, indicating that the solution processed SiO₂ thin film prepared in this work was stoichiometrically equivalent to the thermal oxidized SiO₂ thin film.

The resistivity of the solution processed SiO₂ thin film was estimated to the order of 10¹³ Ωcm. The breakdown was not found below the electric field strength 5 MV/cm. A similar measurement was performed for a thermally oxidized SiO₂ thin film. The resistivity was estimated to the order of 10¹⁴ Ωcm and the breakdown was observed at 8MV/cm. It was found that the resistivity of the solution processed SiO₂ thin film was 10 times lower than that of thermally oxidized SiO₂ thin films.

Table 1 shows the electronic parameters of the SiO₂ thin films fabricated by various techniques. In order to employ the SiO₂ thin film as a gate dielectric for the OFET, it is required to have extreme smooth surface. The surface of thermal oxidized SiO₂ thin film is extreme smooth. However, SiO₂ thin films obtained by CVD and sol-gel technique have a poor smoothness. On the other hand, it was found that the surface of solution processed SiO₂ thin film prepared in this work had the smooth surface as same order with the thermal oxidized SiO₂.

Fig. 2 shows Iₓ–V₁ₓ characteristics for pentacene OFET fabricated using solution processed SiO₂ as a gate dielectric. The field effect mobility was estimated as 0.364 cm²/Vs. The field effect mobility of pentacene OFET using thermally oxidized SiO₂ as a gate dielectric was estimated to 0.296 cm²/Vs from the similar measurement. These results indicated that the electronic performance of OFET using solution processed SiO₂ thin film as a gate dielectric was comparable to that using thermally oxidized SiO₂ thin film, though the resistivity of the solution process SiO₂ thin film was 10 times lower than that of thermally oxidized SiO₂ thin films.
4. Conclusions

We were succeeded to prepare SiO₂ thin film from the silazane compound by low temperature solution process. Obtained SiO₂ thin film had a high resistivity in the order of $10^{13}$ Ωcm was 10 times lower than the thermally oxidized SiO₂ thin film. The OFET using solution process SiO₂ thin film was comparable to that using thermally oxidized SiO₂ thin film, indicating that the solution processed SiO₂ thin film was such useful for the gate insulator of OFET as the thermally oxidized SiO₂ thin film.

References


![Fig. 1 XPS spectra of SiO₂ prepared by thermally oxidation (upper) and low temperature solution process (lower).](image1)

![Fig. 2 Idrain-Vsource-drain characteristics of pentacene OFET fabricated by using solution processed SiO₂ gate insulator.](image2)

Table 1. Preparation temperature, surface roughness and electronic parameters of several SiO₂ thin films prepared by various techniques.

<table>
<thead>
<tr>
<th>Preparation Method</th>
<th>Temperature / K</th>
<th>RMS / nm</th>
<th>Field Effect Mobility / cm²V⁻¹s⁻¹</th>
<th>Resistivity / Ωcm</th>
<th>Dielectric Strength / MVcm⁻¹</th>
</tr>
</thead>
<tbody>
<tr>
<td>Thermal Oxidation</td>
<td>&gt;1223</td>
<td>0.139</td>
<td>0.296</td>
<td>~$10^{14}$</td>
<td>~8.0</td>
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<tr>
<td>Sputter</td>
<td>573</td>
<td>0.172</td>
<td>-</td>
<td>-</td>
<td>-</td>
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<tr>
<td>Thermal CVD</td>
<td>1173</td>
<td>0.916</td>
<td>-</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>Sol-gel</td>
<td>773</td>
<td>0.921</td>
<td>-</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>Solution Process (This Work)</td>
<td>&lt;373</td>
<td>0.346</td>
<td>0.364</td>
<td>~$10^{13}$</td>
<td>&gt;5.0</td>
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