# Moisture Absorption-Induced Permittivity Deterioration and Surface Roughness Enhancement of Lanthanum Oxide Films on Silicon

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# 1. Introduction

Lanthanum oxide (La2O3) is a promising material to substitute SiO<sub>2</sub> as the gate dielectric film because of its relatively high permittivity and a large band gap of  $6eV^{1}$ . However, concerning the permittivity of La<sub>2</sub>O<sub>3</sub> there is not a widely accepted value. Both of very high<sup>2</sup> and low<sup>3</sup> permittivity La2O3 films were reported. Two possible reasons for low permittivity have been considered. One is the low density of amorphous La<sub>2</sub>O<sub>3</sub> films, and the other is silicate formation at the interface of La<sub>2</sub>O<sub>3</sub> film and silicon. On the other hand, it is well known that La<sub>2</sub>O<sub>3</sub> films are hygroscopic, forming hydroxide<sup>4)</sup> which creates charges in the gate dielectric and results in a large flat-band voltage shift<sup>5)</sup>. However, effects of moisture absorption on the permittivity and surface morphology change of La<sub>2</sub>O<sub>3</sub> film have not been investigated intensively. In this work, effects of the moisture absorption on the permittivity of La<sub>2</sub>O<sub>3</sub> film to elucidate the reason for the unstable k-value of La<sub>2</sub>O<sub>3</sub> and the surface roughness of La2O3 film have been investigated.

#### 2. Experiments

La<sub>2</sub>O<sub>3</sub> films were deposited on HF-last Si with RF sputtering method in argon ambient at room temperature and were annealed at 600 °C in pure N<sub>2</sub> ambient for 30 seconds. **Fig.1** shows the sample preparation procedures. The physical thicknesses of La<sub>2</sub>O<sub>3</sub> films were determined with the grazing incident x-ray reflectivity (GIXR) measurements.



**Fig.1.** Illustration of two kinds of sample preparation procedures. One is the water desorbed  $La_2O_3$  film (Type-A) which was annealed at 400°C in sputter chamber to make lanthanum hydroxide decompose into lanthanum oxide and  $H_2O^{6}$ . The other is the moisture absorbed sample (Type-B) which was exposed to the air with different time. Both samples were followed by 6nm SiO<sub>2</sub> layer deposition without any exposure to the air to prevent the moisture absorption.

### 3. Results and Discussion

The permittivity of Type-A  $La_2O_3$  film in Fig.1 was investigated. Capacitance-Voltage (C-V) measurements were performed for Au/SiO<sub>2</sub>/La<sub>2</sub>O<sub>3</sub>/Si/Al MIS capacitors with the frequency of 100 kHz. The CET has a good linear relationship with La<sub>2</sub>O<sub>3</sub> film thickness as shown in **Fig.2**. The permittivity calculated from the slope is about 24.

**Figure 3** shows the CET versus  $La_2O_3$  film thickness plot for Type-B samples in Fig.1. Zero hour exposed to the air means that the sample was put in the sputter chamber for SiO<sub>2</sub> layer deposition as quickly as possible after annealing. Also the permittivity ( $k_{exp}$ ) of the La<sub>2</sub>O<sub>3</sub> film exposed to the air can be calculated from slopes of linear fitting curves. It can be observed that the permittivity of the film changes with the time exposure to the air. Therefore moisture absorption should be a very possible reason for the unstable k-value of La<sub>2</sub>O<sub>3</sub> films in previous literatures.



Fig.2. The relationship of CET to the  $La_2O_3$  physical thickness for Au/SiO<sub>2</sub>/  $La_2O_3$ /Si/Al MIS capacitors.



**Fig.3.** The relationship of CET to Type-B  $La_2O_3$  physical thickness for  $Au/SiO_2/$   $La_2O_3/Si$  MIS capacitors. The sample was exposed to the air for (a) zero hour, (b) 6 hours and (c) 12 hours before SiO<sub>2</sub> layer deposition.



**Fig.4.** XRD patterns of  $La_2O_3$  films on silicon after they were exposed to the air for (a) zero hour, (b) 6 hours, and (c) 12 hours.

From the XRD patterns of La<sub>2</sub>O<sub>3</sub> films (Fig.4), zero hour exposed to the air La<sub>2</sub>O<sub>3</sub> film is poly-crystallized in the hexagonal phase. After the exposure to the air for 6 hours, peaks attributed to La(OH)3 appears, whereas the intensity of peaks attributed to the hexagonal La<sub>2</sub>O<sub>3</sub> decreases. After the exposure to the air for 12 hours, strong La(OH)<sub>3</sub> phase peaks are found and peaks of hexagonal La<sub>2</sub>O<sub>3</sub> disappear completely. Therefore we can conclude that the amount of hexagonal La(OH)<sub>3</sub> in the La<sub>2</sub>O<sub>3</sub> film increased with the time exposed to the air. Although there is no report about the permittivity of hexagonal La(OH)3, from Shannon's consideration<sup>7</sup> we can estimate the permittivity of hexagonal La(OH)<sub>3</sub> (k~10) on the basis of additivity rule of the polarizability. In fact, with the time exposed to the air Fig. 3 shows the degradation of  $k_{exp}$  (k-value obtained experimentally from the slope), though it is necessary to take account of an inhomogeneity of the film due to the partial reaction of the La<sub>2</sub>O<sub>3</sub> with the moisture. We can also point the effect of moisture absorption on the permittivity of La<sub>2</sub>O<sub>3</sub> film from the relationship of CET change to the time exposed the air. As shown in Fig.3, the CET of  $La_2O_3$ film increases with the time exposed to the air, while only a slight change of the physical film thickness is observed. This fact implies that the CET increase should principally be caused by the permittivity deterioration of La<sub>2</sub>O<sub>3</sub> film with the moisture absorption.

In addition, **Fig.5** shows the AFM images of  $La_2O_3$  films which indicate the surface roughness increase in  $La_2O_3$  films exposed to the air. One very possible cause of surface roughness enhancement is the nonuniform reaction of  $La_2O_3$  and  $H_2O$  followed by the nonuniform volume expansion of the film due to moisture absorption. This observation should be another concern of hygroscopic  $La_2O_3$  film properties.

Fig. 6 summarizes that all of CET, RMS and the amount of  $La(OH)_3$  in  $La_2O_3$  film increase with the time exposed to the air.



**Fig.5.** AFM images  $(1\mu m \times 1\mu m)$  of La<sub>2</sub>O<sub>3</sub> film surfaces after the exposure to the air for (a) zero hour, (b) 6 hours, and (c) 12 hours.



**Fig.6**. CET, RMS and hexagonal  $La(OH)_3$  (100) XRD peak intensity as a function of the time exposed to the air of  $La_2O_3$  film on silicon.

## 4. Conclusions

We found that the moisture absorption can degrade the permittivity of  $La_2O_3$  film after the film was exposed to the air for several hours because of the formation of  $La(OH)_3$  with a low permittivity. Thus it is concluded that the moisture absorption should be a possible reason for unstable *k*-value of  $La_2O_3$  films. Furthermore AFM results indicate that the moisture absorption also increases the surface roughness of  $La_2O_3$  films on silicon.

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