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Improved electrical characteristics of atomic-layer-deposited Al₂O₃/GaAs MOS capacitors with (NH₄)₂S-C₄H₉OH sulfide treatment

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1. Introduction

GaAs metal-oxide-semiconductor (MOS) capacitors with different high-*k* gate dielectrics have been of more interest in recent years. Fermi-level-pinning is the most challenging issue, which arises from the remarkably high interfacial state density (*D*_{it}) at the interface. This problem casts some doubts on the anticipated superior electrical performance of NFETs with GaAs channel. In addition, relatively poor thermal stability needs to be cautiously tackled. As a result, additional surface passivation, such as sulfur chemical treatment, Si or Ge cap layer seems unavoidably required in order to effectively reduce native oxide and *D*_{it}. In this study, we found that the employed (NH₄)₂S-C₄H₉OH chemical treatment can significantly improve the high-*k*/GaAs MOS capacitor performance.

2. Experimental

MOSCAP structures were fabricated on high Si-doped (*n*-type, $\sim 1 \times 10^{18} \text{ cm}^{-3}$) GaAs(100) substrates. At first, the GaAs was rinsed in the diluted NH₄OH (1%) solution for 1min. Furthermore, we performed (NH₄)₂S treatments in combination with either H₂O or C₄H₉OH solvent on GaAs surface prior to high-*k* deposition. Surface chemistry was analyzed by employing x-ray photoelectron spectroscopy (XPS) that Al *K*α is used as an excitation source. The Al₂O₃ thin film was then deposited by atomic-layer-deposited (ALD) system at 300 °C as gate dielectric followed by post deposition annealing (PDA) at 600 °C for 30 s in an O₂ ambient. Sputtered Pt dots were patterned as circular gate electrodes through the specific shadow mask with the Al for backside contact. The capacitance-voltage (*C*-*V*) and gate leakage current (*I*-*V*) curves on Pt/Al₂O₃/GaAs MOS capacitors were measured using an HP4284 and Keithley 4200, respectively.

3. Results and Discussion

Figure 1 displays As2*p*3 and Ga2*p*3 photoemission spectra with different wet-chemical cleaning processes (WCPs); the composition ratios were summarized in Table I. It was observed that both kinds of native oxides, especially for As₂O_x (*x*=3,5), were obviously reduced with using (NH₄)₂S solutions; moreover, dissolving the sulfide in C₄H₉OH is more effective as compared to dissolving in (H₂O). The values of As₂O_x/As_{tot}(As-As/As_{tot}) were 97.8%(30.1%) for GaAs surface with deionized water (DIW) rinse only; its value can be reduced to 52.6%(22.6%) and 38.8%(20.1%) for the sulfide solution (H₂O) and the sulfide solution (C₄H₉OH) treatments, respectively. Increasing the concentration of the sulfide solution (C₄H₉OH)

can further eliminate the formation of As oxides. The effects of these surface treatments on the *C*-*V* characteristics of Al₂O₃/GaAs MOS capacitors are shown in Figs. 2(a) and 2(b), respectively. As can be seen, the *C*-*V* frequency dispersion was improved by the sulfide treatments; subsequent 600 °C PDA further improved this dispersion behavior. Here, we qualitatively examine the interface properties through the variation of ΔC and ΔV values (Fig. 3) that are defined: $\Delta C(@V_g=4V) = 1 - C(@100kHz) / C(@1kHz)$ and $\Delta V(@C_{FB}) = V_g(@1kHz) - V_g(@100kHz)$, respectively. It was observed that the sample subject to sulfide solution (C₄H₉OH) revealed the smaller $\Delta C/\Delta V$ and lower leakage current density (*J*_g) than that with sulfide solution (H₂O). The values of the ΔC and ΔV were 17-20% and 0.9-1.4 V for the as-deposited sample and we can improve these two values to 17-14% and 0.6-0.7 V by using 10% sulfide solution (C₄H₉OH). Fig. 4 shows the corresponding leakage current characteristics before and after 600 °C O₂-PDA. The result displayed that the GaAs capacitors through 10% sulfide solution (C₄H₉OH) treatment actually possessed the higher thermal stability due to the smaller amounts of the As-As species and As oxides. According to this chemical reaction: As₂O₃ + GaAs → Ga₂O₃ + 4As, we suggested that the lower As oxides existed close to the dielectric interface should accompany with the reduced formation of the metallic As by-product after thermal processing, thus, a lower leakage current was characterized. The fabricated ALD-Al₂O₃/GaAs MOS capacitors in this work after optimized surface treatment showed the comparable insulating properties with respect to HfSiO_x dielectrics reported on *n*-GaAs in Fig. 5.

4. Conclusions

In summary, we effectively improved the electrical properties of Pt/ALD-Al₂O₃/GaAs MOSCAP by dissolving (NH₄)₂S in C₄H₉OH solvent for the chemical treatment. We found this solution did result in superior electrical characteristics than the conventional (NH₄)₂S solution (H₂O). Further optimization of the sulfide solution (C₄H₉OH) conditions for the treatment of GaAs surface prior to high-*k* deposition is seems indispensable in pursuit of superior high-performance high-*k*/GaAs MOSFETs.

References

- [1] D. Shahrjerdi *et al.*, Appl. Phys. Lett. **89**, (2006) 043501.
- [2] H. S. Kim *et al.*, Appl. Phys. Lett. **89**, (2006) 222903.
- [3] H. S. Kim, *et al.*, Appl. Phys. Lett. **91**, (2007) 042904.
- [4] M. Zhu *et al.*, Appl. Phys. Lett. **89**, (2006) 202903.
- [5] M. H. Zhang, *et al.*, Appl. Phys. Lett. **89**, (2006) 042902.

- [6] S. I. Park, *et al.*, Appl. Phys. Lett. **91**, (2007) 082908.
 [7] M. K. Lee *et al.*, J. Electrochem. Soc. **153**, (2006) F266.

Wet Clean	As-As/ As_{Total}	As-S/ As_{Total}	Ga-S/ Ga_{Total}	As ₂ O ₃ / As_{Total}	Ga ₂ O ₃ / Ga_{Total}
D.I water only	30.1 %	—	—	97.8 %	54.3 %
NH ₄ OH only	23.5 %	—	—	73.4 %	25.8 %
1% Sulf.(H ₂ O)	22.6 %	18.9 %	7.1 %	52.6 %	26.4 %
1% Sulf.(C ₄ H ₉ OH)	20.1 %	17.8 %	7.3 %	38.8 %	22.7 %
10% Sulf.(C ₄ H ₉ OH)	13.9 %	16.3 %	10.9 %	23.2 %	24.6 %

Table. 1 Chemical ratios calculated by XPS fitting results according to the different WCPs in Fig. 1.

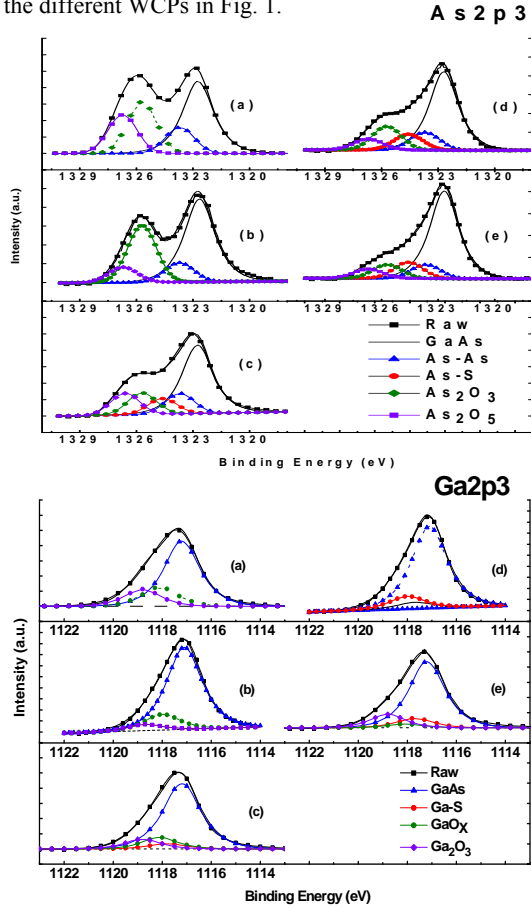


Fig. 1 As $2p_{3/2}$ and Ga $2p_{3/2}$ XPS spectra of GaAs surface receiving various WCPs: (a) DIW only, (b) NH₄OH only, (c) 1%-Sulf.(H₂O), (d) 1%-Sulf.(C₄H₉OH), and (e) 10%-Sulf.(C₄H₉OH). Note that the cleaned surface was exposed to air for 30 min.

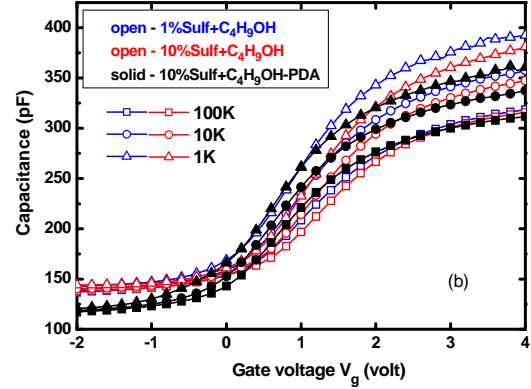
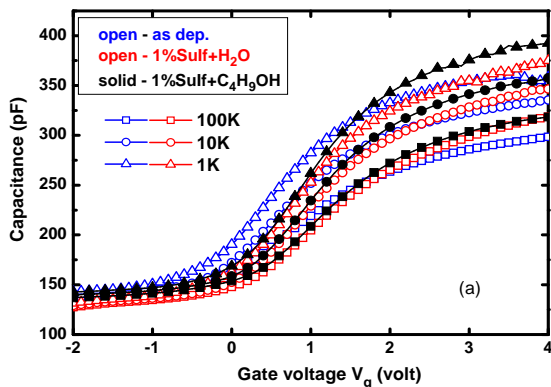


Fig. 2 Multi frequency C-V curves of Pt/Al₂O₃/n-GaAs capacitors with (a) different WCPs and (b) different WCPs and O₂ annealing.

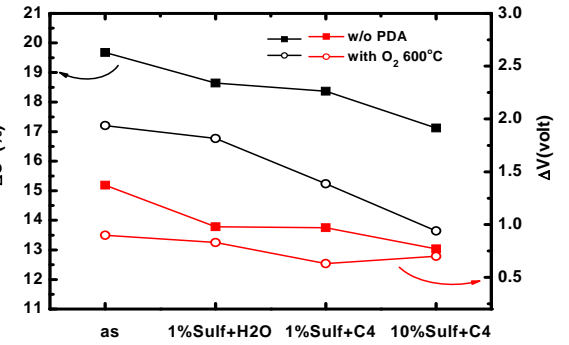


Fig. 3 variations of ΔC and ΔV values of Pt/Al₂O₃/n-GaAs capacitors after different WCPs and post annealing.

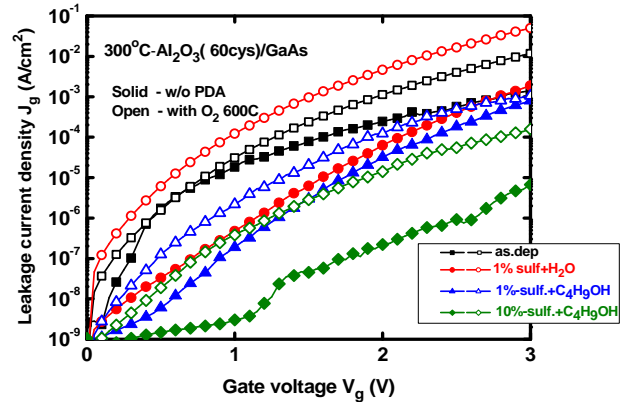


Fig. 4 Gate leakage current characteristics of Pt/Al₂O₃/n-GaAs capacitors with different WCPs. The Al₂O₃ thin films were deposited at 300 °C for 60 cycles.

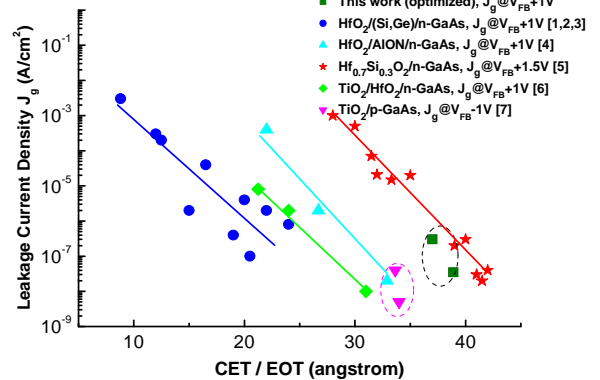


Fig. 5 Comparison of J_g versus CET or EOT characteristics in this work with other's published data.