Lanthanoid Metal Oxide MIM Capacitors for Precision Analog Circuits: Material Screening, Process Development, and Characterization

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I. Introduction

Metal-insulator-metal (MIM) capacitors occupy substantial areas in integrated circuits for radio-frequency (RF) and analog/mixed-signal applications. Increasing the capacitance density enables smaller die sizes. A high capacitance density should be achieved without compromising the voltage coefficient of capacitance (VCC) and leakage current. High-k dielectrics have been investigated to replace the conventional SiO₂ and SiON in MIM capacitors [1]-[2]. High-ĸ (κ ~20) lanthanoid metal oxides have attracted much attention [3]. However, a systematic study of lanthanoid metal oxides for the application in RF and analog/mixed-signal MIM capacitors has not been performed except for La₂O₃ [4]. In this paper, we report the first investigation of a series of oxides of lanthanoid elements (La, Gd, Tb, Dy, Er, and Yb) for MIM capacitor applications. Er₂O₃ was found to be very promising. We further discuss the optimization and electrical characterization of MIM capacitors with Er₂O₃ dielectric.

II. Experiments

MIM capacitors were fabricated on Si wafers covered with 400 nm thick SiO₂. A 200 nm thick TaN bottom electrode layer was sputter-deposited. Lanthanoid dielectrics (La2O3, Gd2O3, Dy2O3, Er2O3, and Yb₂O₃) were reactively sputtered from metal targets in Ar/O₂ ambient with various Ar/O2 ratios, while the total gas flow was fixed at 30 sccm. Post deposition annealing (PDA) was performed in 3 different ambient: (1) 100% N2 with no O2, (2) trace O2, or (3) 95% N2 mixed with 5% O2. For the process with "trace O2", the N2 gas flow contained 5% O₂ before temperature ramp-up, but there was no O₂ flow afterwards. Finally, a 150 nm thick TaN top electrode layer was formed by sputtering and patterning using dry etching. The MIM device structure and process flow are shown in Fig. 1.

III. Results and discussions

(a) Material Screening

Table I lists the lanthanoid oxides investigated. All these oxides have energy gaps close to or larger than 5 eV, except Tb₂O₃. Fig. 2 compares the quadratic VCC α of MIM capacitors with 6 different dielectrics. Data presented in Fig. 2 are the best ones for each oxide obtained from a first round of process optimization (Section II). The black line represents the highest capacitance density for a given α and is contributed by data points from Er₂O₃. MIM capacitors with Er₂O₃ dielectric have lowest α (positive) for a given capacitance density. Fig. 3 compares the leakage current density J at +3.3 V for all 6 materials. Er₂O₃ dielectric gives the lowest leakage current (black line) for a given capacitance density. Er₂O₃ outperforms all other 5 candidates in both quadratic VCC and leakage current; it is also better than other high-k materials reported in literature, which will be discussed later.

(b) Process Development of Er₂O₃ MIM capacitors

To optimize the performance of Er2O3 MIM capacitors, we varied the ambient oxygen concentration during deposition and PDA. Fig. 4 shows the capacitance obtained at the different process conditions. For the 12 nm split, the capacitance density drops significantly with increased O2 in PDA process; this is due to the oxidation of bottom TaN electrode. However, there is no significant capacitance change for 24 nm and 36 nm splits. Fig. 5 demonstrates typical C-V curves of MIM capacitors with 12 nm Er_2O_3 , where the quadratic α and linear VCC β are extracted. Fig. 6 summarizes α values for all thicknesses and process conditions. Thinner oxides show larger α . For each thickness, capacitors annealed in trace O2 ambient demonstrate smaller α than the other two PDA conditions with less or more oxygen. This implies that a proper amount of O_2 is helpful in maintaining a small α . The capacitor with Er₂O₃ sputtered using 2 sccm O₂ exhibits

significantly higher a comparing to one with Er₂O₃ sputtered in 3 sccm O₂, although their thicknesses and capacitance densities are almost the same. This indicates that the oxygen vacancies induced in the sputtering process has a large impact on α . Linear VCC β values are summarized in Fig. 7. β is related to asymmetric device structure. β reduces with increased oxygen concentration during PDA, and the effect is less pronounced for thicker oxides; this can be explained by a more uniform dielectric layer with less oxygen vacancies. The leakage current J of 24 nm Er_2O_3 is shown in Fig. 8. J is at the order of 10^{-8} at ± 3.3 V bias; the split with trace O₂ during PDA has a smaller J than the other two. Considering the results discussed above, we decided the optimized process condition for Er₂O₃ deposition to be: 3 sccm O₂ flow in sputter and trace O2 in PDA process. The following characterization was all performed on devices formed with this process condition.

(c) Electrical Characterization of Er₂O₃ MIM capacitors

Fig. 9 shows the capacitance densities and loss tangent of MIM capacitors with Er2O3 dielectric as a function of frequency. The loss of capacitance from 1 kHz to 200 kHz are 0.68%, 0.8% and 1.1%, for 12 nm, 24 nm, and 36 nm Er₂O₃ MIM capacitors, respectively. The loss tangent increases from $\sim 10^{-4}$ at 1 kHz to $\sim 10^{-2}$ at 200 kHz for all samples. The frequency dispersion of α and β are shown in Fig. 9 and Fig. 10, respectively. $Log(\alpha)$ decreases linearly with a logarithmic increase in frequency. The frequency dependence of α can be explained as the change of relaxation time with different carrier mobility in insulator [6]. β decreases with increasing frequency for 24 nm and 36 nm Er₂O₃ capacitors, but increases for the 12 nm split. This is due to different electrode/dielectric interface conditions. Fig. 11 shows the temperature dependence of capacitance (TCC). 24 nm and 36 nm samples both show acceptable TCC of ~180 ppm/°C. The smaller TCC for 12 nm sample might be due to the interfacial layer caused by the oxidation of bottom electrode. Literature data on high-k MIM capacitors are summarized and compared in Table II. In comparison with the data in literature, MIM capacitors with PVD Er₂O₃ dielectric show high capacitance density, low leakage current, low $\alpha,$ acceptable β and small TCC, suggesting its potential use in future RF and analog/mixed signal IC applications.

IV. Conclusion

We systematically studied 6 lanthanoid oxides as dielectrics in MIM capacitors for RF and analog/mixed-signal circuits. Among those oxides, Er2O3 outperforms others in quadratic VCC and leakage current. With process optimization, Er2O3 MIM capacitors achieve high capacitance densities ranging from 5.8 to 18.4 fF/µm², low VCC and TCC values, and low leakage current at around 1×10^{-8} A/cm².

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Fig. 1. MIM capacitor structure and process flow.



Fig. 3. Leakage current at +3.3V plotted versus capacitance density for all lanthanoid oxide MIM capacitors.



Fig. 6. Quadratic VCC α against different oxygen concentrations in PDA ambient.



Fig. 9. Frequency dispersion of capacitance density and loss tangent.

Table I: Comparison of physical properties of selected 'lanthanoid elements and their sesquioxides [5].

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	La	Gd	Tb	Dy	Er	Yb
Atomic Number	57	64	65	66	68	70
Oxide	La ₂ O ₃	Gd ₂ O ₃	Tb ₂ O ₃	Dy ₂ O ₃	Er ₂ O ₃	Yb ₂ O ₃
Band-gap (eV)	5.6	5.4	3.8	4.9	5.3	4.9



Fig. 4. Capacitance densities against different oxygen concentrations in PDA ambient.



Fig. 7. Linear VCC β against different oxygen concentrations in PDA ambient.



Fig. 10. Frequency dispersion of quadratic VCC α and linear VCC $\beta.$







Fig. 5. Normalized *C-V* curves of 12 nm Er₂O₃ MIM capacitors. The solid lines are polynomial fitted curves.



Fig. 8. *J-V* curves of 24 nm Er₂O₃ MIM capacitors with different oxygen concentrations in PDA ambient.



capacitance density. TCC values are small.

Гable II.	Comparison	of DC	performance	of reported	binary	high-K	MIM	capacitors
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	Dielectrics	Cap. Density (fF/µm ²)	$J_{\text{leak}} @ 1 V (A/cm2)$	$J_{\text{leak}} @ 3.3 V (A/cm2)$	α @ 100kHz (ppm/V ²)	β@ 100kHz (ppm/V)	TCC
SiO ₂ /Si ₃ N ₄ MIM	SiO ₂ , ONO, Si ₃ N ₄ [7-11]	1~2	1E-7~1E-9	N/A	10~70	10~160	40~130
High-κ MIM	$Ta_2O_5[12]$	4	N/A	6E-5	-9.9	N/A	106
	Al ₂ O ₃ [13]	5.2	4.3E-8	N/A	2051	1888	109~208
	ALD HfO ₂ [14]	8	~4E-8	~6E-7	~1800	~4000	N/A
	$La_2O_3[4]$	9.2	<1E-5	N/A	~3000	~3000	347
	PVD HfO ₂ [15]	13.7	N/A	4E-4@125°C	4631	-4843	135
Our results	Er ₂ O ₃ (36 nm)	5.8	3.5E-9	~6.4E-9	240	-430	178
	Er ₂ O ₃ (24 nm)	8.9	5.5E-9	~1.4E-8	580	-620	170
	Er_2O_3 (12 nm)	18.4	1.3E-8	~4.9E-7	2000	-360	126