

The CeO₂ Sensing Membrane with NH₃ Plasma Treatment in Biosensor Applications

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Abstract- In the study, the electrolyte-insulator-semiconductor (EIS) biosensor device using CeO₂ sensing membrane and combined with NH₃ plasma treatment has been investigated. It can be found that the better sensing characteristics such as higher sensitivity (48.62 mV/pH) after plasma treatment at 3min, including lower hysteresis voltage, higher biosensor sensitivity, and higher linearity about 99.61%.

I. Introduction

Various high dielectric constant materials such as Gd₂O₃, Ta₂O₅, HfO₂ and Er₂O₃ [1] have been reported in EIS devices to substitute SiO₂ as a sensing membrane because of low cost, compatible with silicon, and small size. However, since high-k based devices are strongly dependent on fabrication processes, different treatments and alternative material compositions, intensive studies have investigated to achieve better ion sensing performance, smaller hysteresis, lower drift and higher stability to meet the requirement of commercial bio-medical industrial applications.

In this study, we discuss CeO₂ (k=20~26)[2-3] applied in EIS type combined with NH₃ plasma at different time. The physical analysis includes XRD (X-ray diffraction), XPS (X-ray photoelectron Spectroscopy) and AFM (Atomic Force Microscope). The electrical analysis includes the C-V curve, hysteresis voltage, drift rate and biosensor sensitivity.

II. Experiment

The EIS structures of a high-k CeO₂ sensing membrane were fabricated on 4-inch n-type (100) Si wafers. A 50 nm CeO₂ film was deposited on the Si substrate by means of sputtering from a gadolinium target in the ratio of Ar/O₂ = 25/0. Subsequently, samples were plasma treatment in NH₃ ambient for 1min, 3min and 6min. Then samples were annealed in N₂ ambient for 30 s by rapid thermal annealing (RTA) at various temperatures from 600°C. After that, the back-side contact of the Si wafer was deposited by Al film with 300nm-thick. Then, the sensing area was defined by standard photolithography process using a photosensitive epoxy, SU8-2005 of Micro Chem Inc. Finally, the samples were fabricated on the copper lines of printed circuit board (PCB) by silver gel. Epoxy package was used to separate the EIS structure and the copper line. Fig.1

III. Results and Discussion

Fig.2(a)-(d) shows the C-V curves CeO₂ and sensing membrane by NH₃ plasma treatment (1min to 6min) in buffer solution with various pH values. The sample of CeO₂ and treated in the plasma of 3min possessed the highest sensitivity. It showed that the capacitance value of the EIS device with NH₃ plasma treated at 6min is smaller than that of treated at 3min indicating the formation of a thicker amorphous Ce silicate layer at the CeO₂/Si interface. Therefore, the pH sensing membrane with a proper NH₃ plasma treatment at 3min can obtain an excellent linearity and high sensitivity.

Fig.3(a)-(c) shows hysteresis voltage and drift rate under different NH₃ plasma treatment time conditions, and illustrates that the sample of CeO₂ film at 3min shows the lowest hysteresis voltage of 7.8 mV and the lowest drift rate of 0.51 mV/hr. Proper plasma treatment can repair bond connections, and repair the defects in the CeO₂ film by nitrogen incorporation.

Fig.4 shows the XRD spectra of high-k CeO₂ film before and after NH₃ plasma treatment samples. It can be seen that the NH₃ plasma treatment can induce crystallization in high-k sensing membrane, and the sample with NH₃ plasma treatment at 3min exhibited a stronger peak intensity in CeO₂ (400). Besides, in order to investigate the composition and chemical behaviors of high-k CeO₂ film, fig.5(a)-(b) shows the O 1s and Ce 3d XPS spectra of CeO₂ sensing membrane. In addition, the CeO₂ plasma treatment at 3min film can be seen the Ce 3d peak located at 881.6eV. Then O 1s spectra of CeO₂ film exhibited a smaller Ce-silicate peak intensity when the sensing membrane plasma treatment at 3min. This result indicates that the CeO₂ with NH₃ plasma at 3min could reduce SiO₂ and silicate formation.

Fig.6(a)-(b) shows AFM images of the CeO₂ sensing membrane for the as-deposited and NH₃ plasma treatment at 3min samples. The root mean square (rms) values of the above samples were 0.42 nm, and 0.56 nm, respectively. Because of interior grain size and structure, which became stronger and larger as NH₃ plasma treatment time increased.

Fig.7 shows the urea concentration controlled in a range between 5mM, and 40mM. The sensitivity values of the CeO₂ film above as-deposited and NH₃ plasma treatment at 3min samples were 2.42 and 3.81 mV/mM, respectively. Therefore, the CeO₂ sensing membrane NH₃ plasma treatment at 3min has better sensitivity and linearity for urea detection.

Fig.8 shows the glucose sensing properties of the CeO₂ sensing membrane on EIS structure. From these figures, CeO₂ film shows the sensitivity is 3.50 and 4.42 mV/mM in the concentration range between 2mM to 7mM. The CeO₂ sensing membrane with NH₃ plasma treatment at 3min has better sensitivity and linearity than as-deposited sample for glucose detection.

IV. Conclusions

In this study, the sensing membrane with NH₃ plasma treatment at 3min and annealed at 600°C can improve biosensor device performance, including excellent C-V curve, better hysteresis voltage, and smaller drift rate about 0.51 mV/hr. Therefore, the high-k CeO₂ sensing membrane with proper NH₃ plasma treatment is very promising for EIS biosensor applications.

V. References

- [1] L. Bousse, S. Mostarshed, B. Van der Schoot and N. F. de Rooij, *Sensors and Actuators B*, 17 (1994) 157-164.
- [2] Chun-Heng Chen, Ingram Yin-Ku, Joseph Ya-Min Lee, and Fu-Chien Chiu, *Appl. Phys. Lett.* 92(2008), 043507.
- [3] Yi Wang, Feng Wei, Shoujing Yue, Zhimin Yang, and Jun Du, *Appl. Phys. Lett.* 92(2008), 012915.

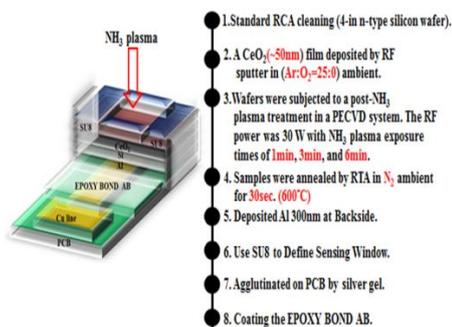


Fig. 1 The process flow of CeO₂ sensing membrane with NH₃ plasma treatment.

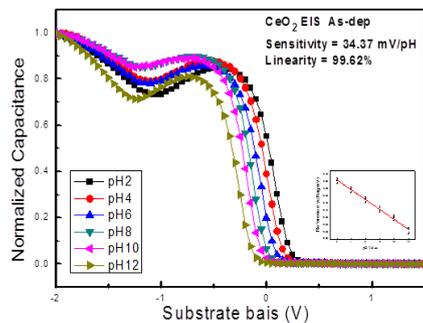


Fig. 2(a) CeO₂ as-dep

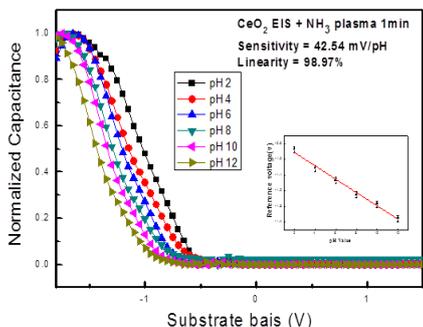


Fig. 2(b) CeO₂ with NH₃ 1min

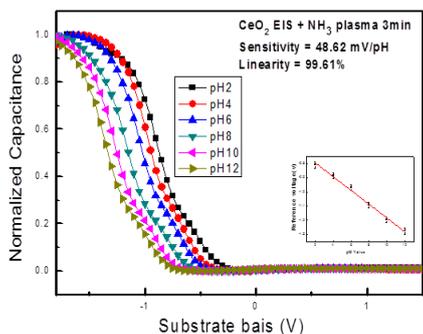


Fig. 2(c) CeO₂ with NH₃ 3min

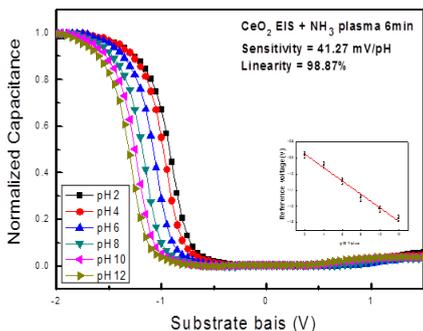


Fig. 2(d) CeO₂ with NH₃ 6min

Fig. 2(a)-(d) The normalized C-V curve of the sample without and with NH₃ plasma treatment at various conditions, the inset figure represents the sensitivity and linearity.

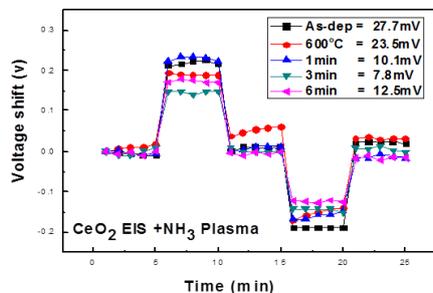


Fig. 3(a) The hysteresis of CeO₂ sensing membrane with NH₃ plasma treatment at various conditions during the pH loop of 7→4→7→10→7 over a period of 25 minutes.

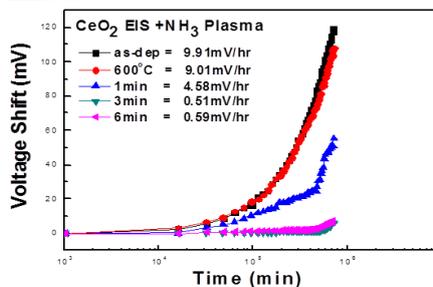


Fig. 3(b) The drift voltage of CeO₂ sensing membrane annealed with NH₃ plasma treatment at various conditions, then dipped in pH 7 buffer solution for 12 hours.

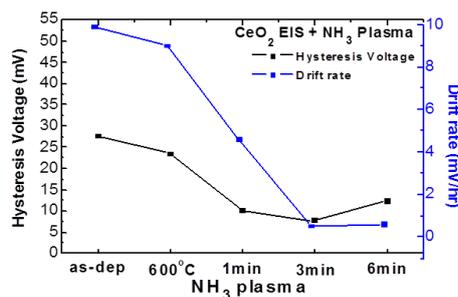


Fig. 3(c) Shows the samples with NH₃ plasma treatment at various conditions of hysteresis and drift rate.

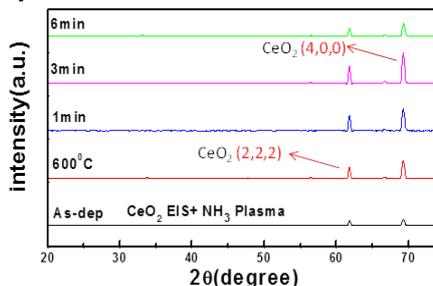


Fig. 4 XRD of the CeO₂ film with NH₃ plasma treatment on single crystalline silicon

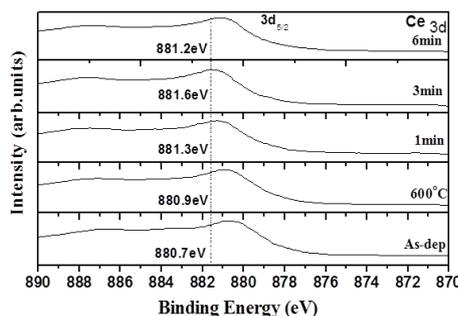


Fig. 5(a) XPS Ce 3d results of CeO₂ film with NH₃ plasma treatment.

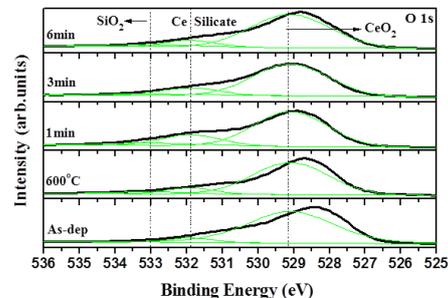


Fig. 5(b) XPS O 1s results of CeO₂ film with NH₃ plasma treatment.

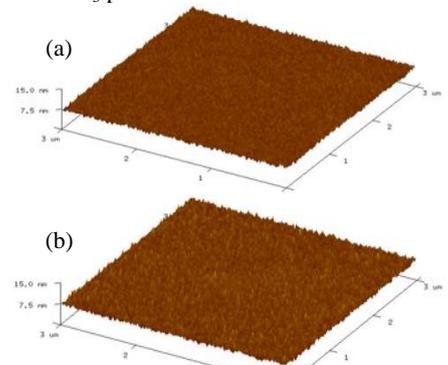


Fig. 6(a)-(b) (a) As-dep R_{rms}=0.42(nm), (b) NH₃ plasma at 3min R_{rms}=0.56(nm)

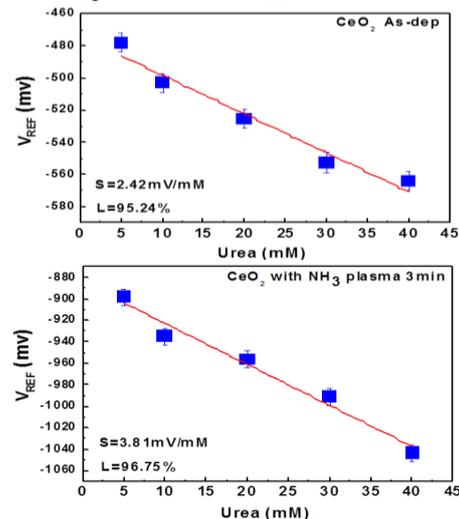


Fig. 7 pUrea-responses of enzyme-immobilized CeO₂ as-dep and CeO₂ with NH₃ plasma 3min EIS structure by covalent bonding method.

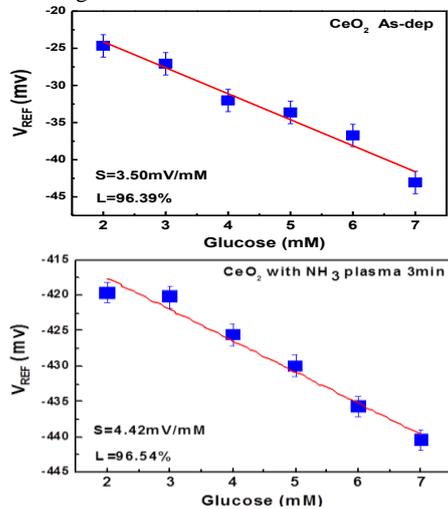


Fig. 8 pGlucose-responses of enzyme-immobilized CeO₂ as-dep and CeO₂ with NH₃ plasma 3min EIS structure by covalent bonding method.