

Reactive Deposition Epitaxy of SrGe₂ Thin Films on Ge (111) and (001) Substrates

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Abstract

Semiconductor SrGe₂ is a new material having a large absorption coefficient at near-infrared light. SrGe₂ thin films are firstly fabricated by using reactive deposition epitaxy on Ge substrates.

1. Introduction

Alkaline-earth silicides and germanides have useful functions for many technological applications such as solar-cells, thermoelectric, and optoelectronics. However, the study of germanides has not been active compared to that of silicides.

SrGe₂ is one of the Alkaline-earth germanides. The theoretical and experimental study of bulk SrGe₂ revealed the following properties [1–3]: (i) the BaSi₂-type structure (orthorhombic, space group: $D_{2h}^{16} - Pnma$, no. 62, $Z = 8$), (ii) indirect transition semiconductor with a band gap of approximately 0.82 eV, (iii) absorption coefficient of $7.8 \times 10^5 \text{ cm}^{-1}$ at 1.5 eV photon, which is higher than that of Ge ($4.5 \times 10^5 \text{ cm}^{-1}$ at 1.5 eV photon). These properties mean that SrGe₂ is an ideal material for use in the bottom cell of high-efficiency tandem solar cells. Therefore, the fabrication of a SrGe₂ thin film on arbitrary substrates realizes thin film tandem solar cells simultaneously achieving high conversion efficiency and low process cost. In this study, we investigate the reactive deposition epitaxy (RDE) of SrGe₂ on Ge substrates to explore the possibility of the thin film formation of SrGe₂.

2. Experimental Procedures

We fabricated high quality BaSi₂ thin films on Si(111) and Si(001) substrates using a two-step way: BaSi₂ template layer is formed by RDE, which is the Ba deposition with heating Si substrates, followed by molecular beam epitaxy (MBE) [4]. This method will be applicable for SrGe₂. Thus, we have tried to form SrGe₂ template layers on Ge(111) and Ge(001) substrates using RDE, as shown in Fig. 1. The deposition rate and time of Sr were 1.3 nm/min and 60 min for Ge(111), and 0.7 nm/min and 120 min for Ge(001). The substrate temperature was set at 300–700 °C during the Sr deposition. After that, 5-nm-thick amorphous Si was deposited at room temperature to protect the RDE layer from oxidation because Sr-Ge compounds are easily oxidized by air.

The crystallinity of the sample was evaluated by using reflection high energy electron diffraction (RHEED) and X-ray diffraction (XRD) with Cu K α radiation. Besides, the surface morphology was observed by scanning electron microscopy (SEM).

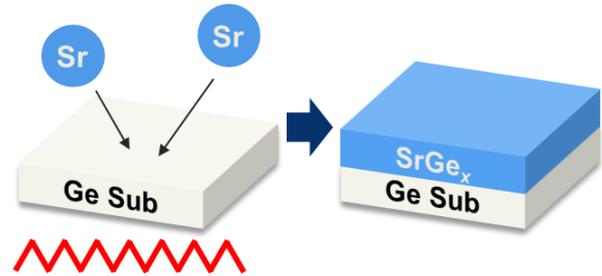


Fig. 1. RDE of a SrGe_x thin film on a Ge substrate.

3. Results and Discussion

For all the samples, streaky RHEED patterns were observed after the Sr deposition, indicating the epitaxial growth of Sr-Ge compounds. For the samples with a Ge(111) substrate, Fig. 2(a) shows that the $\theta-2\theta$ XRD pattern varies with the substrate temperature. The peaks corresponding to SrGe₂ appear for the samples with 500 and 600 °C though the other samples exhibit the peaks corresponding to Sr₂Ge. This behavior is probably related to the balance of the supply rate of Ge atoms from the substrate and the evaporation rates of Sr and Ge atoms from the sample surface. We note that a comparison with the XRD pattern of powder SrGe₂ determines that the 600 °C sample is preferentially [111] oriented.

Fig. 2(b) shows the $\theta-2\theta$ XRD patterns of the samples with a Ge(001) substrate. The 300 °C sample exhibits a peak corresponding to SrGe₂(200) in addition to peaks corresponding to Sr₅Ge₃. This result indicates that the SrGe₂ thin film is preferentially [100] oriented on a Ge(001) substrate. Because the SrGe₂ peak is weak, there is a possibility that the 300 °C sample contains SrGe₂ crystals other than [200] orientation. When the substrate temperature is higher than 300 °C, the SrGe₂(200) peak is not observed. The optimum temperature for forming SrGe₂ is much different between Ge(111) and Ge(001) substrates. This means that the formation energy of SrGe₂ significantly varies with the crystal orientation of the substrate.

Fig. 3 shows the SEM images of the sample surfaces. For all the samples, island patterns are observed. When the substrate temperatures are 400–600 °C, we observe the unique patterns reflecting the crystal orientation of the substrates, that is, three-fold symmetry for Ge(111) and two-fold symmetry for Ge(001). These patterns ensure the epitaxial growth of Sr-Ge compounds and account for the streaky RHEED patterns. The 700 °C samples exhibit dot patterns, suggesting that the Sr atoms migrated rapidly or Ge atoms evaporated.

4. Conclusions

We firstly fabricated SrGe₂ thin films using RDE on Ge substrates. For a Ge(111) substrate, preferentially [111] oriented SrGe₂ thin film was formed at 600 °C. On the other

hand, for a Ge(001) substrate, preferentially [100] oriented SrGe₂ thin film was formed at 300 °C together with other Sr-Ge compounds. The detailed crystal quality and surface morphology are discussed in the presentation.

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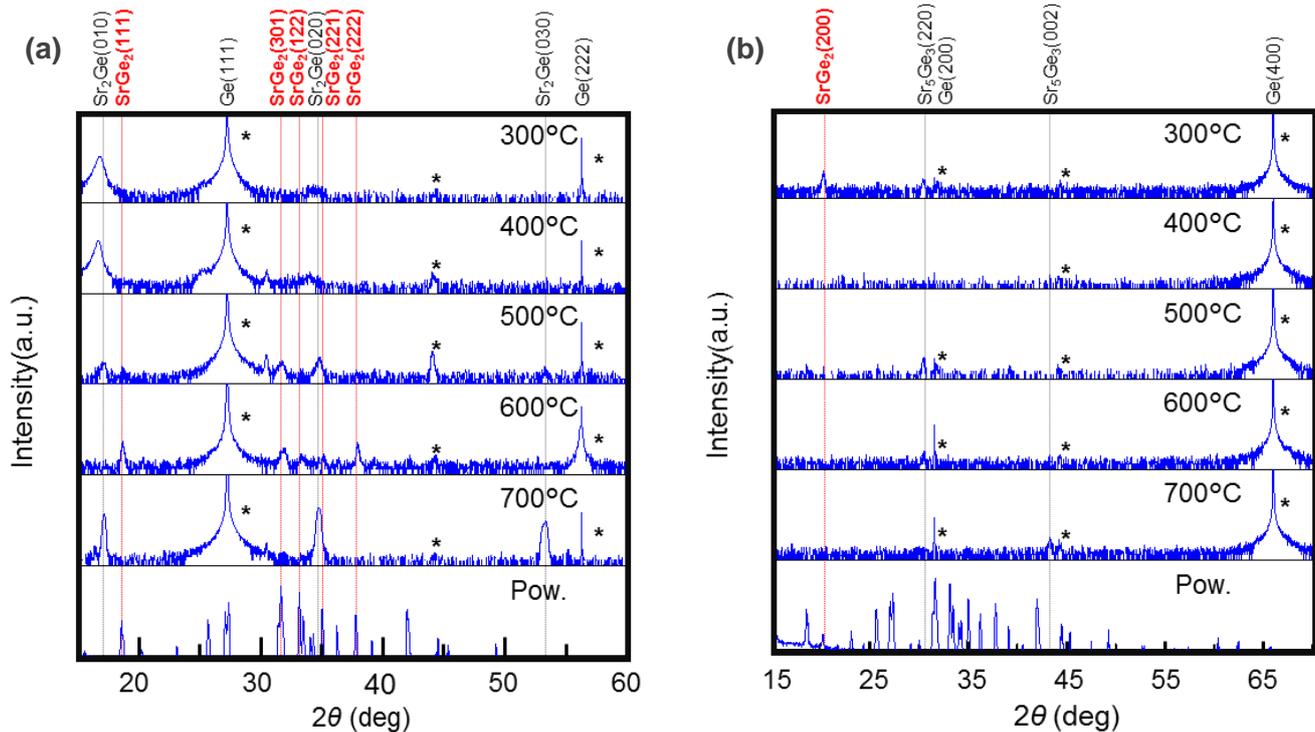


Fig. 2. θ - 2θ XRD patterns of the samples with (a) Ge(111) and (b) Ge(100) substrates where the substrate temperatures are 300-700 °C. The powder diffraction of SrGe₂ is shown for comparison.

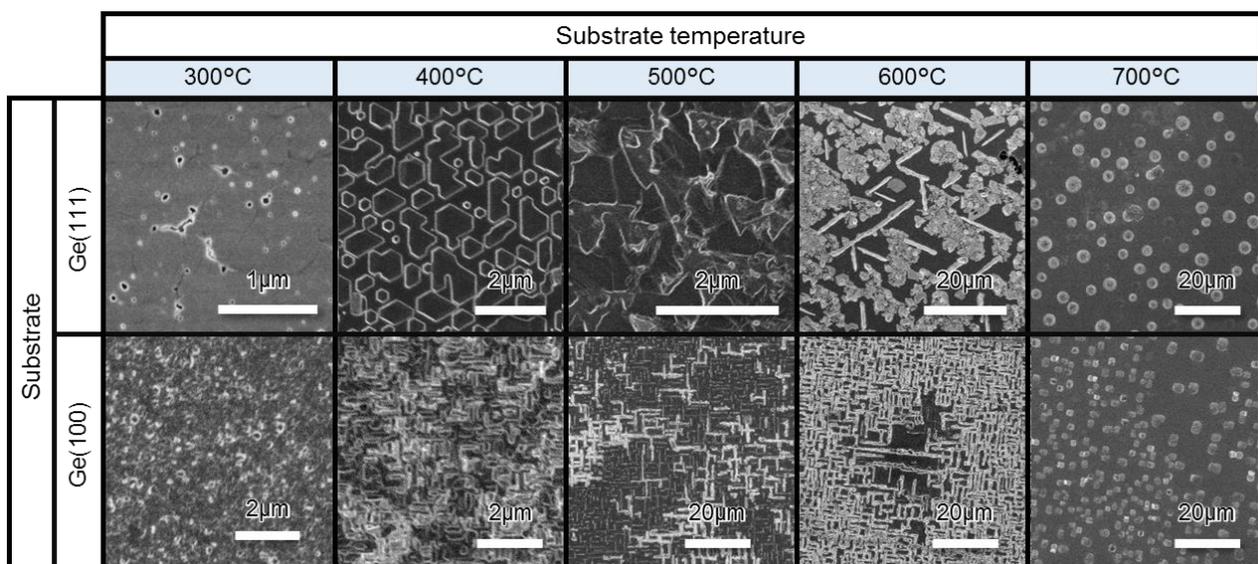


Fig. 3. SEM images of the surfaces of the samples.