# Investigation of Thermal Treatment Effects of PbI<sub>2</sub> Film Yielded Two-step Type Perovskite Solar Cells

<u>Kohei Yamamoto</u><sup>1</sup>, Keitaro Hamada<sup>2</sup>, Md. Shahiduzzaman<sup>3</sup>, Kyosuke Yonezawa<sup>1</sup>, Makoto Karakawa<sup>1, 3</sup>, Takayuki Kuwabara<sup>1</sup>, Kohshin Takahashi<sup>1</sup>, and Tetsuya Taima<sup>1, 3</sup>

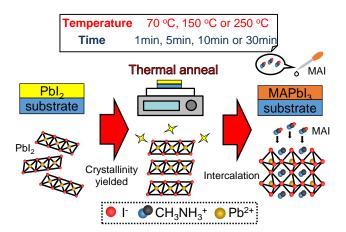
<sup>1</sup> Graduate School of Natural Science and Technology, Kanazawa University, Kakuma, Kanazawa 920-1192, Japan <sup>2</sup> Japan Advanced Institute of Science and Technology, Nomi, Ishikawa 923-1292, Japan <sup>3</sup>Institute for Frontier Science Initiative (InFiniti), Kanazawa University, Kakuma, Kanazawa 920-1192, Japan E-mail: k yama@stu.kanazawa-u.ac.jp; taima@se.kanazawa-u.ac.jp

# Abstract

A precise control of the morphology and crystallization of perovskite thin-films is well-correlated to higher perovskite solar cells performances. The methylamonium lead iodide (MAPbI<sub>3</sub>) film is fabricated by intercalating of MAI molecules into PbI<sub>2</sub> film. Hence, crystal growth control of PbI<sub>2</sub> is important to obtain highly efficient perovskite film. Herein, we attempt to control the crystal growth of PbI<sub>2</sub> through a simple spin-coating method via varying annealing temperature of 70, 150, and 250 °C for 1, 5, 10, and 30 min. We also investigate the effect of PbI<sub>2</sub> crystallinity on the performance of resulting perovskite solar cells.

### 1. Introduction

Organometallic halide perovskite (PSCs) have recently emerged as promising cost-effective and highly efficient nanostructured solar cells<sup>1, 2)</sup>. The first PSC with a power conversion efficiency (PCE) of 3.81% was reported in 2009 by Kojima *et al*<sup>3)</sup>. Currently, typical PCE values of perovskite solar cells are over  $20\%^{11}$  far higher than that of organic thin-film solar cells. In the formation of MAPbI<sub>3</sub> (methylamonium lead iodide, CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3</sub>) perovskite crystal, it is considered that MAI (methylammonium iodide, CH<sub>3</sub>NH<sub>3</sub>I) intercalates into PbI<sub>2</sub> film as shown in **Fig. 1**. However, it is not cleared that the relationship of the film quality and crystallinity of the PbI<sub>2</sub> film with influence of MAI intercalation by using two-step method.



**Fig. 1.** Schematic illustration of formation mechanism of PbI<sub>2</sub> and MAPbI<sub>3</sub> film affected by thermal annealing.

The  $PbI_2$  assumes different crystal phases depending on temperature; therefore it is considered that the crystallinity, morphology and the like of the  $PbI_2$  film change by thermal annealing.

In this study, we aimed to elucidate the control of PbI<sub>2</sub> crystal growth for precise intercalation to yield efficient, perovskite thin-film and solar cells.

#### 2. Experiment

Device structure of our PSCs is ITO / compact-TiO<sub>2</sub> / MAPbI<sub>3</sub> / spiro-OMeTAD / Ag. Bare ITO was treated with oxygen plasma for 20 min before use. The compact-TiO<sub>2</sub> layer was prepared by chemical bath deposition (CBD) method5). The perovskite layer was formed by sequential solution deposition and spin-soating of PbI<sub>2</sub> and MAI. MAPbI<sub>3</sub> was formed using two-step spin-coating procedure. PbI<sub>2</sub> solution was prepared by dissolving 230 mg PbI<sub>2</sub> in 1 ml N, N-dimethylformamide (DMF, 99.5%, Kanto Chemical) under stirring at room temperature (RT). PbI<sub>2</sub> solution (100 µl) was spin-coated on the compact-TiO<sub>2</sub> film at 2,000 rpm. for 30 s without loading time. After spinning, the film was dried at 70 °C, 150 or 250 °C for 1, 5, 10 or 30 min and after cooling to RT, MAI solution (0.063 M, 10mg/ml) in 2-propanol was loaded on the PbI2 coated substrate for 10 s as loading time, which was spun at 2,000 rpm. for 30 s and dried at 70°C for 5 min and 100 °C for 10 min, and 120 °C for 10 min. Finally, we obtained 200 nm-thick MAPbI<sub>3</sub> film. Then, hole transport layer (spiro-OMeTAD) and Ag electrode were deposited in glove box and evaporation chamber without air expose, respectively. X-ray diffraction (XRD) patterns of the prepared films were measured using an X-ray diffractometer (SmartLab, Rigaku, Japan) with an X-ray tube (Cu K $\alpha$ ,  $\lambda$ = 1.5406 Å). Ultraviolet-visible (UV-Vis) absorption spectra of perovskite films were measured using an absorption spectrophotometer.

# 3. Results

In terms of addressing the structural properties (e.g. full-width at half maximum, FWHM) of PbI<sub>2</sub> and MAPbI<sub>3</sub> films, we applied X-ray diffraction (XRD). The lower FWHM has a relationship with higher crystallinity of film. In case of annealing temperature of PbI<sub>2</sub> film for 1 min, the FWHM of PbI<sub>2</sub> yielded MAPbI<sub>3</sub> film was reduced by higher annealing temperature, as shown in **Fig. 2**.

The PbI<sub>2</sub> and MAPbI<sub>3</sub> films morphology were analyzed with scanning electron microscopy (SEM) as shown in **Fig.** 

**3.** The images in **Fig. 3(d)-3(f)** reveal that while increasing the annealing time and temperatures increases the crystallinity of PbI<sub>2</sub>, leading to the large grain size of resulting PbI<sub>2</sub> yielded MAPbI<sub>3</sub> thin-films. In contrast, lower annealing temperature conditions have remained extra MAI (red circle) as shown in **Fig. 3(d) and 3(e)** on the resulting MAPbI3 thin-film. These results suggest that high crystallinity PbI<sub>2</sub> allow the perfect intercalation of MAI molecules, compared to low crystallinity PbI<sub>2</sub> film.

As can be revealed that in terms of annealing time of  $PbI_2$  film at 250 °C, the crystallinity was remained the same. In contrast, the UV-vis spectrum of MAPbI<sub>3</sub> film were found to increase absorption depending on annealing time of  $PbI_2$  film at 250 °C as shown in **Fig. 4**.

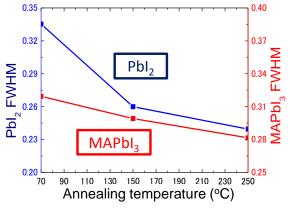


Fig. 2. The FWHM of  $PbI_2$  diffraction peak (001) and MAPbI<sub>3</sub> diffraction peak (110) calculated by XRD spectrum depending on annealing temperature of  $PbI_2$  film for 1 min.

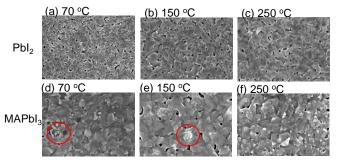
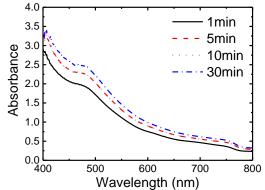
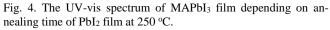


Fig. 3. The SEM images of PbI<sub>2</sub> film surface annealed by (a)70 °C, (b) 150 °C and (c) 250 °C. The MAPbI<sub>3</sub> surface using PbI<sub>2</sub> annealed by (d) 70 °C, (e) 150 °C and (f) 250 °C.





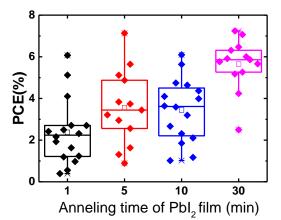


Fig. 5. Reproducibility results of perovskite solar cells in term of PbI<sub>2</sub> film annealing time at 250  $^{\circ}$ C.

Optimum annealing temperature was found at 250 °C for the resulting devices fabrication. Power conversion efficiencies (PCEs) were increased considerably in the average range of  $2.38\pm1.52\%$  to  $5.65\pm1.18\%$ , while improving the annealing time of 1 to 30 min as shown in **Fig. 5**. The best control and highest reproducibility was observed for the control PbI<sub>2</sub> crystal film annealed with 30 min, leading to an enhanced charge dissociation and transport efficiency and reduction in recombination events in the solar cells. This highlights that the crystallinity of PbI<sub>2</sub> significantly affected to the resulting of MAPbI<sub>3</sub> solar cell performance.

# 4. Conclusions

We investigated the effect of crystallinity of PbI<sub>2</sub> on the performance of resulting MAPbI<sub>3</sub> solar cells. The crystallinity of PbI<sub>2</sub> yielded MAPbI<sub>3</sub> film was increased by higher annealing temperature. PCEs were increased considerably in the average range of  $2.38\pm1.52\%$  to  $5.65\pm1.18\%$ , while improving the annealing time of 1 to 30 min. Our results suggested that the crystallinity of PbI<sub>2</sub> significantly affected to the resulting of MAPbI<sub>3</sub> solar cell performance.

### Acknowledgements

This research work was partially supported by research grant from Japan Power Academy, Research Grant of Tateishi Science and Technology Foundation, 2017 Research Grant of Iketani Science and Technology Foundation, 2016 Research Grant of Iwatani Naoji Fundation, 2015 Basic Research Grant of TEPCO Memorial Foundation, 35th Research Grand of Tonen General Sekiyu Foundation, and JSPS KAKENHI Grant 16K05882.

#### References

- [1] NREL Best Research-cell Efficiencies
  - [http://www.nrel.gov/ncpv/images/efficiency\_chart.jpg]
- [2] J-H Im et al., Nature Nanotech, 9, 927 (2014)
- [3] A. Kojima et al., J. Am. Chem. Soc. 131, 6050 (2009)
- [4] S. N. Habisreutinger et al., Nano Lett., 14, 5561 (2014)
- [5] T. Kuwabara et al., Org. Electron., 11, 1136 (2010)