

Performance of In₂TiO₅ Based Bio-sensor Treated by CF₄ Plasma

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Abstract- In this study, the electrolyte-insulator-semiconductor biosensor device using In₂TiO₅ sensing membrane and combined with CF₄ plasma treatment has been investigated. It can be found that the better sensing characteristics such as higher sensitivity (59.64 mV/pH) after plasma treatment at 60sec, including lower hysteresis voltage and higher linearity about 99.68%.

I. Introduction

Electrolyte-Insulator-Semiconductor (EIS) biosensors are one of the most promising candidates for bio-analyte sensing due to its excellent transducers for sensing biochemical reactions owing to their small size, fast response, and reliability. properties. Recently, researchers have focused on Ti-doped Indium oxide (Ti-doped In₂O₃) thin films owing significant attention as a sensing membrane because it has higher mobility (160 cm²/(V·s)) [1], higher melting point (1910 °C), larger band gap (3.5–3.7 eV) [2], According to the literature, In₂TiO₅ films can easily be reduced to create an oxygen deficiency, where insufficient oxygen atoms are contained in the crystal structure. As a potential solution, CF₄ plasma treatment of the In₂TiO₅ film can enhance the In-Ti-O bonding as well as increase the sensing performance. In this work, we used post-deposition CF₄ plasma treatment on In₂TiO₅ thin-films to improve their physical and biosensing characteristics.

II. Experiment

The EIS structures of a high-k In₂TiO₅ sensing membrane were fabricated on 4-inch n-type (100) Si wafers. A 50 nm thick SiO₂ substrate was thermally oxidized on the silicon wafer. A 50 nm In₂O₃ film was deposited on the Si substrate by using sputtering system. The indium target is used for deposition. An another Ti doped In₂O₃ sample is prepared using co-sputter technique. Subsequently, samples were plasma treatment in CF₄ ambient for 15 sec, 30 sec and 60 sec. After that, the back-side contact of the Si wafer was deposited by Al film with 300nm-thick. The sensing area of the deposited In₂TiO₅ films was defined by an automatic robot dispenser with an adhesive silicone gel. Finally, the samples were placed on the copper lines of printed circuit board (PCB) by pasting with silver gel. Epoxy package was used to separate the EIS structure and the copper line (Fig.1)

III. Results and Discussion

Fig. 2(a)-(d) shows the C-V curves In₂TiO₅ and sensing membrane by CF₄ plasma treatment (15 sec to 60sec) in buffer solution with various pH values. The sample of In₂TiO₅ and treated in the plasma of 60sec possessed the highest sensitivity. It showed that the capacitance value of the EIS device with CF₄ plasma treated at 30sec is smaller than that of treated at 60 sec indicating the formation of a thicker amorphous In-Ti silicate layer at the In₂TiO₅/Si interface. Therefore, the pH sensing membrane with a proper CF₄ plasma treatment at 60

sec can obtain an excellent linearity and high sensitivity. Fig. 3(a)-(c) shows hysteresis voltage and drift rate under different CF₄ plasma treatment time conditions, and illustrates the lowest hysteresis voltage of 2.72 mV and the lowest drift rate of 0.42 mV/hr for the 60 seconds plasma treated sample. Proper plasma treatment can repair bond connections and defects in the In₂TiO₅ film by fluorine incorporation.

Fig. 4 shows the XRD spectra of high-k In₂TiO₅ film before and after CF₄ plasma treatment samples. It can be seen that the CF₄ plasma treatment can induce crystallization in high-k sensing membrane, and the sample with CF₄ plasma treatment at 60 sec exhibited a stronger peak intensity in In₂TiO₅ (222). Besides, in order to investigate the composition and chemical behaviors of high-k In₂TiO₅ film, Fig. 5(a)-(b) shows the O 1s and In 3d XPS spectra of In₂TiO₅ sensing membrane. In addition, the In₂TiO₅ plasma treatment at 60 sec film can be seen the In 3d peak located at 452.26 eV and 444.65 eV. The O 1s spectra of In₂TiO₅ film exhibited a smaller In-Ti silicate peak (531.8 eV) when the sensing membrane is treated by plasma for 60 seconds. This result indicates that the In₂TiO₅ with CF₄ plasma at 60 seconds could reduce SiO₂ and silicate formation.

Fig. 6(a)-(b) shows AFM images of the In₂TiO₅ sensing membrane for the as-deposited and CF₄ plasma treatment at 60 sec samples. The root mean square (rms) values of the above samples were 0.62 nm, and 2.92 nm, respectively. Because of interior grain size and structure, which became stronger and larger as CF₄ plasma treatment time increased.

Fig. 7 shows the urea concentration controlled in a range between 5mM, and 40mM. The sensitivity values of the In₂TiO₅ film above as-deposited and CF₄ plasma treatment at 60 sec samples were 1.55 and 2.69 mV/mM, respectively. Therefore, the In₂TiO₅ sensing membrane CF₄ plasma treatment at 60 sec has better sensitivity and linearity for urea detection.

Fig. 8 shows the glucose sensing properties of the In₂TiO₅ sensing membrane on EIS structure. From these figures, In₂TiO₅ film shows the sensitivity is 2.85 and 6.63 mV/mM in the concentration range between 2mM to 7mM. The In₂TiO₅ sensing membrane with CF₄ plasma treatment at 60 seconds has better sensitivity and linearity than as-deposited sample for glucose detection.

IV. Conclusion

In this study, the sensing membrane with CF₄ plasma treatment at 60 sec is improved biosensor device performance, such as excellent C-V curve, better hysteresis voltage, and smaller drift rate about 0.42 mV/hr. Therefore, the In₂TiO₅ sensing membrane with proper CF₄ plasma treatment is very promising for EIS biosensor applications.

V. References

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- [2]D. Beena, K. J. Lethy, R. Vinodkumar, A.P. Detty, V.P. Mahadevanpillai, V. Ganesan, Optoelectron. Adv. Mat. 489 (2011) 215-223.

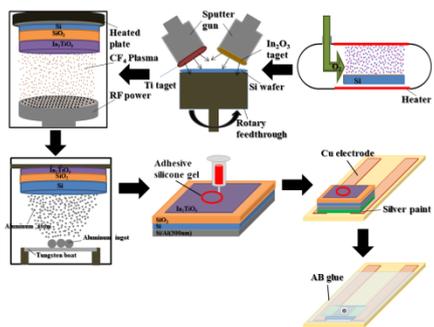


Fig. 1 The process flow of In_2TiO_5 sensing membrane with CF_4 plasma treatment.

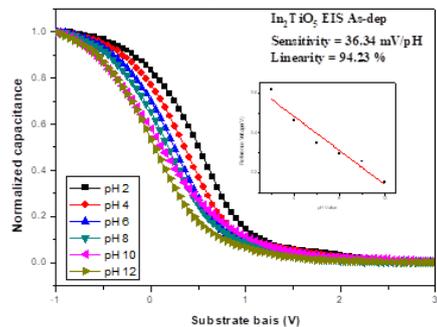


Fig. 2(a) In_2TiO_5 as-dep

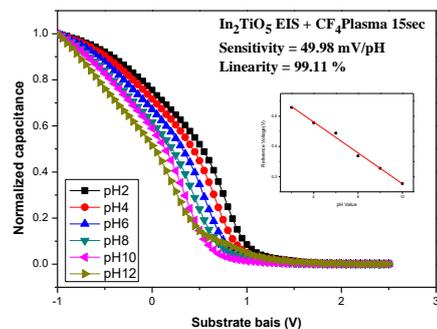


Fig. 2(b) In_2TiO_5 with CF_4 15sec

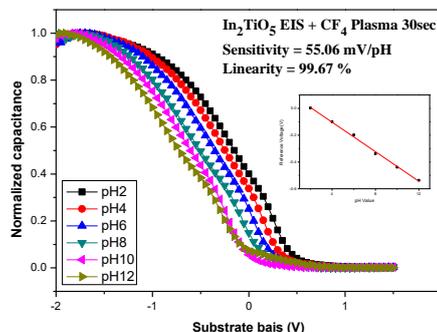


Fig. 2(c) In_2TiO_5 with CF_4 30sec

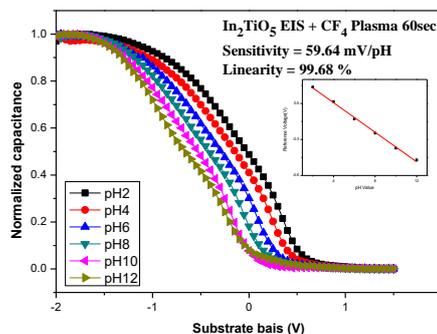


Fig. 2(d) In_2TiO_5 with CF_4 60sec

Fig. 2(a)-(d) The normalized C-V curve of the sample without and with CF_4 plasma treatment at various conditions, the inset figure represents the sensitivity and linearity.

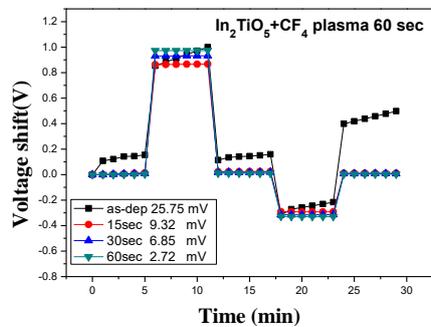


Fig. 3(a) The hysteresis of In_2TiO_5 sensing membrane with CF_4 plasma treatment at various conditions during the pH loop of 7→4→7→10→7 over a period of 30 minutes.

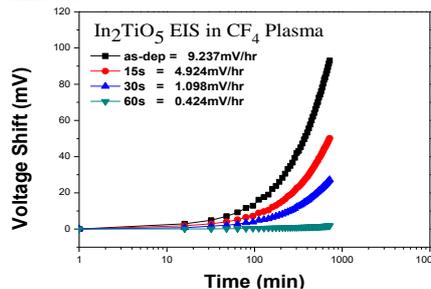


Fig. 3(b) The drift voltage of In_2TiO_5 sensing membrane annealed with CF_4 plasma treatment at various conditions, then dipped in pH 7 buffer solution for 12 hours.

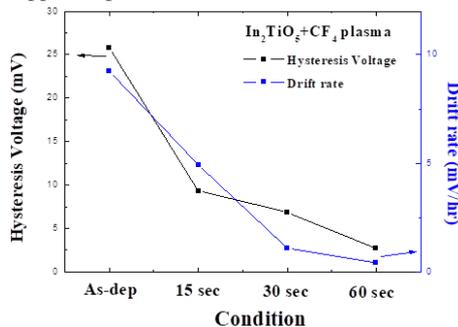


Fig. 3(c) Shows the samples with CF_4 plasma treatment at various conditions of hysteresis and drift rate.

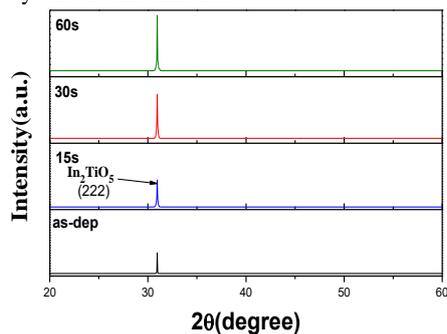


Fig. 4 XRD of the In_2TiO_5 film with CF_4 plasma treatment on single crystalline silicon

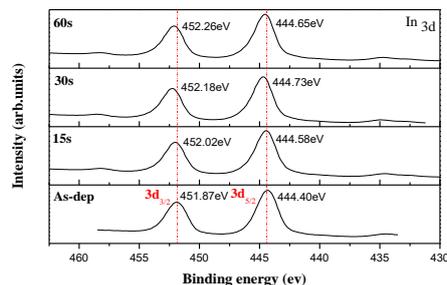


Fig. 5(a) XPS In 3d results of In_2TiO_5 film with CF_4 plasma treatment.

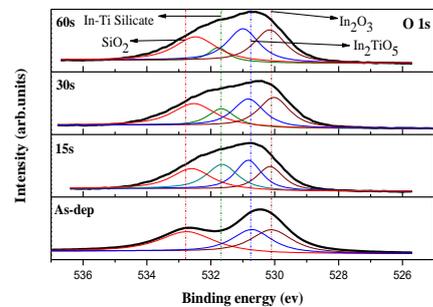


Fig. 5(b) XPS O 1s results of In_2TiO_5 film with CF_4 plasma treatment.

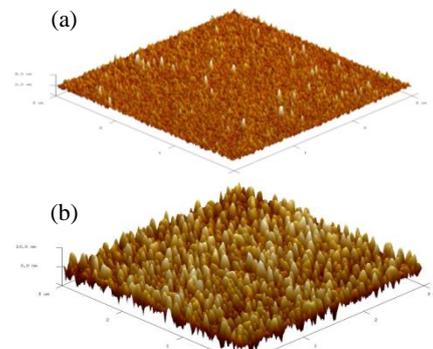


Fig. 6(a)-(b) (a) As-dep $R_{\text{rms}}=0.62(\text{nm})$, (b) CF_4 plasma at 60sec $R_{\text{rms}}=2.92(\text{nm})$

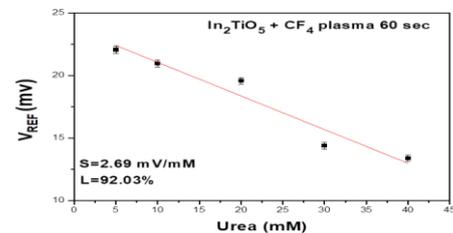
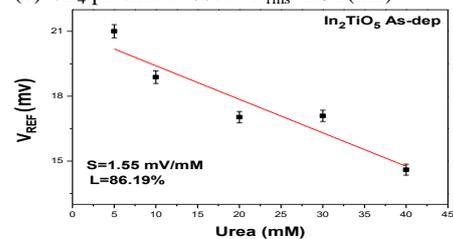


Fig. 7 pUrea-responses of enzyme-immobilized In_2TiO_5 as-dep and In_2TiO_5 with CF_4 plasma 60sec EIS structure by covalent bonding method.

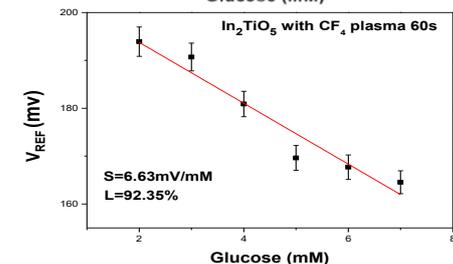
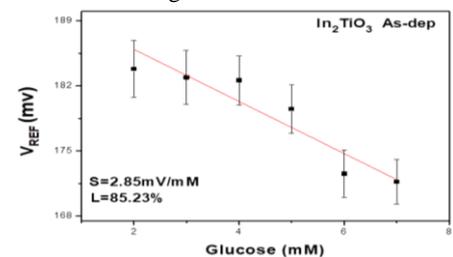


Fig. 8 pGlucose-responses of enzyme-immobilized In_2TiO_5 as-dep and In_2TiO_5 with CF_4 plasma 60sec EIS structure by covalent bonding method.