# Highly Efficient Planar Perovskite Solar Cells Utilizing Electron Transport Bilayer

Md. Shahiduzzaman, <sup>1,2\*</sup> Hiroto Ashikawa, <sup>1</sup> Mizuki Kuniyoshi, <sup>1</sup> Sem Visal, <sup>1</sup> Tetsuya Kaneko, <sup>1</sup> Tetsuhiro Katsumata, <sup>1</sup> Tetsuya Taima, <sup>2</sup> Satoru Iwamori, <sup>1</sup> Masao Isomura, <sup>1\*</sup> and Koji Tomita <sup>1\*</sup>

<sup>1</sup>Tokai Univ., Hiratsuka, Kanagawa 259-1292, Japan <sup>2</sup>Kanazawa Univ., Kakuma, Kanazawa 920-1192, Japan

Phone: +81-463-58-1211(ext.3742) E-mail: shahiduzzaman@tokai.ac.jp; isomura@tokai-u.jp; tomita@keyaki.cc.u-to-kai.ac.jp

#### **Abstract**

Herein, we report a facile method for enhancing the performance of planar heterojunction perovskite solar cells by inserting a compact TiO<sub>2</sub>/Anatase (AT) TiO<sub>2</sub> NPs bilayer, respectively, by spray pyrolysis (SP) deposition and spin-coating (SC) technique. The improved photovoltaic performance of devices with an SP-TiO<sub>2</sub>/SC-AT TiO<sub>2</sub> NPs bilayer facilitated more efficient electron transport, charge extraction, and low interfacial recombination, and ultimately yields champion efficiencies up to 17.05% by a significant decrease of *J-V* hysteresis, presenting almost 12% enhancement compared to the TiO<sub>2</sub> single layer-based counterparts.

#### 1. Introduction

Kojima *et al.* first introduced perovskite solar cells (PSCs) in 2009 with a power conversion efficiency (PCE) of 3.8% [1]. This encouraging discovery led to further performance enhancements, resulting in a remarkable efficiency of 22.7% at the present date [2].

In this study, we report a novel, higher potential and environmental friendly approach for preparing single-crystal-line anatase (AT) TiO<sub>2</sub> NPs by a simple, scalable one-step hydrothermal route [3] using water-soluble titanium complex as a titanium source. The resultant AT TiO<sub>2</sub> NPs performs as electron transport layer (ETL) bilayer in planar heterojunction (PHJ) PSCs.

### 2. Materials and Method

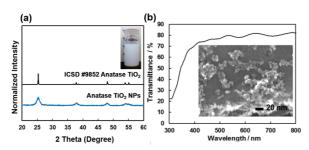
Titanium fine powder of 10 mmol was dissolved in the mixed solution of 30%-H<sub>2</sub>O<sub>2</sub> aq. and 28%-NH<sub>3</sub> aq., then malic acid of 10 mmol was added here. After the solution was dried at 50 °C in order to remove excess H<sub>2</sub>O<sub>2</sub> and NH<sub>3</sub>, 0.5 M Ti-malate complex solution was prepared by adding distilled water. The Ti complex solution was heat-treated in an autoclave at 200 °C for 5 h. Formed TiO<sub>2</sub> particles were separated by centrifuge and washed with distilled water twice. TiO<sub>2</sub> NPs were dispersed in distilled in water. Then, AT TiO<sub>2</sub> NPs layer was placed on the compact spray-pyrolysis (SP) deposited TiO<sub>2</sub> layer by spin-coating (SC) method and annealed at 100 °C for 5 min, followed by sintered at 450 °C for 30 min in a muffle furnace. Finally, a compact

SP-TiO<sub>2</sub>/SC- AT TiO<sub>2</sub> NPs bilayer was obtained by combining SP deposition and SC technique, respectively, and used as ETLs. The perovskite precursor solution (1M PbI<sub>2</sub> and 1M CH<sub>3</sub>NH<sub>3</sub>I in DMF and DMSO mixed solvent) was coated on the resulting substrates. The precursor coated substrates were then allowed to anneal at 100 °C on a hot plate for 1 h to crystallize perovskite. The optimized thickness of SP-TiO<sub>2</sub>, SC-AT TiO<sub>2</sub> (three cycle-coating), perovskite layer, Spiro-OMeTAD and Au layers were measured ~70, ~50, ~270, ~250, and ~60 nm, respectively.

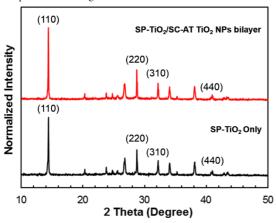
#### 3. Results and Discussion

The as-synthesized nocrystalline AT TiO2 NPs can formulate stable (more than a month) nonocolloidal solutions in aqueous media (inset of Fig. 1a; concentration = 10 mg/mL). The XRD pattern of AT-TiO<sub>2</sub> is depicted in Fig. 1a. Thus, confirms as-synthesized TiO2 NPs indeed comprise a significant amount of the AT domains. The resultant AT TiO2 NPs film exhibits well-behaved optical transparency, with transmittance greater than 80% over the entire visible region (Fig. 1b). To gain further insights into nanoscale morphology and sizes of AT TiO<sub>2</sub> NPs, scanning transmission electron microscopy (STEM) has been performed as shown in inset Figure 1b. The STEM image reveals that the AT TiO<sub>2</sub> NPs is spherical-like, with a particle size ranging from 6 to 10 nm in diameter. The XRD patterns of the perovskite films formed with, and without bilayer are depicted in Figure 2. The single SP-TiO<sub>2</sub> layer based-perovskite film yields a poly-crystalline morphology with a grain size of average around 351 nm, whereas the SP-TiO<sub>2</sub>/SC-AT TiO<sub>2</sub> NPs bilayer based-perovskite film reveals an average grain size of 765 nm, respectively, as estimated full-width half maximum (FWHM) from whole pattern fitting method using the Scherrer equation and resultant values are consistent with the topview morphologies as shown in the SEM images in Fig. 3c and 3d. Interfacial modification of compact SP-TiO2 by a novel single-crystalline SC-AT TiO<sub>2</sub> NPs layer with the 50nm-thickness is homogenously distributed, followed by uniform and denser scaffolds of SP-TiO2/SC-AT TiO2 NPs bilayer (Figure 3b) without visible cracks and thus offers more efficient the separation of charges and suppresses the

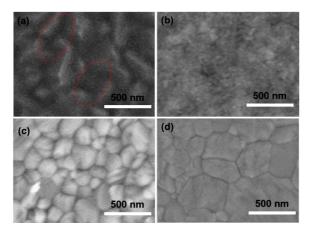
recombination rate.



**Fig. 1.** (a) XRD pattern of AT TiO<sub>2</sub> NPs. Inset shows AT TiO<sub>2</sub> nonocolloidal solutions vial in aqueous media; (b) Transmittance spectra of SC-AT TiO<sub>2</sub> NPs film. Inset showing a representative STEM image of SC-AT TiO<sub>2</sub> NPs deposited on FTO/glass substrate.



**Fig. 2.** The XRD patterns of perovskite films formed with single SP-TiO<sub>2</sub> layer and SP-TiO<sub>2</sub>/SC-AT TiO<sub>2</sub> NPs bilayer.



**Fig. 3.** Top view of SEM images of (a) SP-TiO $_2$  and (b) SP-TiO $_2$ /SC-AT TiO $_2$  NPs bilayer; perovskite films grown on (c) SP-TiO $_2$  and (d) SP-TiO $_3$ /SC-AT TiO $_2$  NPs bilayer layer.

The photoluminance (PL) measurement reveals that the PL of perovskite is quenched significantly in the presence of SP-TiO<sub>2</sub>/SC-AT TiO<sub>2</sub> NPs bilayer as shown in **Fig. 4.** The PL for SP-TiO<sub>2</sub>/SC- AT TiO<sub>2</sub> NPs/perovskite film shows a stronger quenching, revealing efficient charge transfer at the interface between the photo-excited perovskite and SP-TiO<sub>2</sub>/SC-AT TiO<sub>2</sub> NPs bilayer.

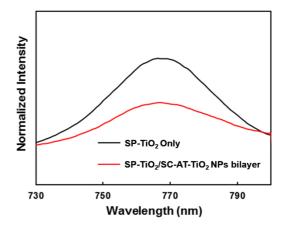
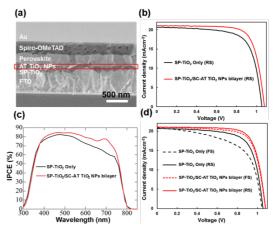


Fig. 4. PL spectra of perovskite films deposited on with or without bilayers.



**Fig. 5**. (a) A cross-sectional SEM image of the PSCs processed with  $TiO_2$  NPs-based bilayer. (b) Reverse-scan (RS) J-V curves obtained for solar cells based on with and without bilayer. (c) IPCE spectra of representative devices. (d) Forward-scan and RS J-V characteristics of solar cells.

The SP-TiO<sub>2</sub> singled layer-based champion (RS) PSCs exhibited considerably low a  $J_{\rm sc}$  of 20.68 mAcm<sup>-2</sup>,  $V_{\rm oc}$  of 1.05 V, FF of 0.70, and PCE of 15.33% as shown in **Fig. 5b**. In contrast, the FF was enhanced from 0.70 to 0.75 upon the inclusion of SP-TiO<sub>2</sub>/SC-AT TiO<sub>2</sub> NPs bilayer. We assumed that the performance enhancement of a device utilizing bilayer ETL is caused by the high content of the nanocrystalline AT phase within the compact layer and the highly homogeneous morphology that favored together with electron extraction by reducing the rate of carrier recombination, compared to that of SP-TiO<sub>2</sub> single layer.

## Conclusions

The optimized three cycle-coating layers of SC-AT TiO<sub>2</sub> NPs on SP-TiO<sub>2</sub> based devices yield the maximum PCE of 17.05% was obtained for the resulting PSCs.

#### Acknowledgements

This study was supported in part by Research and Study Project of Tokai University General Research Organization.

#### References

- [1] A. Kojima, K. Teshima, Y. Shirai and T. Miyasaka, J. Am. Chem. Soc. **131** (2009) 6050.
- [2] https://www.nrel.gov/pv/assets/images/efficiency-chart.png vol. (accessed 2018).
- [3] K. Tomita, V. Petrykin, M. Kobayashi, M. Shiro, M. Yoshimura, M. Kakihana, Angew. Chem. Int. Ed 45 (2006) 2378.