## Control of Aggregation Pathway and Self-Assembled Morphology by Partial Disabling of Hydrogen-bonding Group

(<sup>1</sup>Graduate School of Science and Engineering, Chiba University, <sup>2</sup>Institute for Global Prominent Research, Chiba University,) OTakumi Aizawa,<sup>1</sup> Shiki Yagai<sup>2</sup> Keywords: Hydrogen-bonding; Supramolecular Polymer; Self-assembly; Thiocarbonyl Group; Barbituric Acid

Hydrogen bond is one of the reliable noncovalent interactions to achieve controlled supramolecular assembly. However, for even those systems with highly directional multiple hydrogen bonding, polymorphism often results.<sup>1a</sup> A particular example is the crystalline polymorphism of barbiturate compounds showing characteristic hydrogen-bonded structures.<sup>1b</sup> In contrast, we have developed unique barbiturate supramolecular assemblies from barbituric acid-functionalized (barbiturated)  $\pi$ -conjugated molecules such as 1 (Figure 1, left) through hydrogen-bonded cyclic hexamers (rosettes).<sup>2</sup> The rosettes stack through  $\pi$ - $\pi$  interactions to provide supramolecular polymers with intrinsic curvature. The formation of the classical tape-like hydrogen-bonded motifs might be involved in the supramolecular polymerization process, but it is elusive due to strong aggregation trend of rosettes.<sup>3</sup> If the selective formation of tape-like hydrogen bonds is achieved by a structural modification of barbiturate units, we can make distinct materials from the same  $\pi$ - conjugated molecules. As the formation of rosette requires the hydrogen-bonding on C = O group at position 2 of barbiturate unit,<sup>4</sup> in this study we synthesized compound 2 as a 2-thiobarbituric acid analogue of 1 with the expectation that the weaker hydrogen-bonding ability of the C=S group at position 2 disable rosette formation.

Based on XRD and microscopic techniques, we show that 2 form crystalline fibers through the formation of the tapelike motif. (Figure. 2, right). We also applied computational approaches to explain selective formation of tapelike aggregates at the level of hydrogen-bonding.

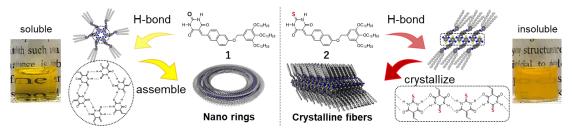


Figure 1. Schematic illustrations of this study.

[1] (a) M. Wehner, F. Würthner et al., J. Am. Chem. Soc. 2019, 141, 6092–6107; (b) J. Bernstein, Acta Crystallogr. Sect. B 1991, 47, 1004 – 1010; [2] S. Yagai et al., Acc. Chem., Res. 2019, 52, 1325–1335; [3] A. Isobe, T. Aizawa, S. Yagai et al., Chem. Eur. J. 2020, 26, 8997–9004; [4] J. C. McDonald, G. M. Whitesides, Chem. Rev. 1994, 94, 2383–2420.