Synthesis and Photophysical Properties of Yb(III) Coordination Polymer with Phosphine Oxides Having Porphyrin Moiety

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Keywords: Ytterbium; Porphyrin; Coordination Polymer; Near-infrared Luminescence; 4*f*-4*f* transition

Near-infrared (NIR) luminophores have attracted attention for bioimaging applications due to their high biopermeability. Yb(III) luminophores exhibit characteristic NIR emission based on 4*f*-4*f* transition. Yb(III) complexes with porphyrin photosensitizers have been reported as a NIR luminescent materials.¹ The strong NIR emission of Yb(III) complexes and coordination polymers with hexafluoroacetylacetonato (hfa) and phosphine oxide ligands have also been reported.^{2,3} Here, we report synthesis

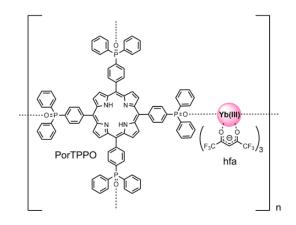


Figure 1. Molecular structure of Yb(III) coordination polymer [(Yb(hfa)₃)₂(PorTPPO)]_n.

and photophysical properties of novel Yb(III) coordination polymer composed of Yb(hfa)₃ and phosphine oxide linker ligands with porphyrin π -skeleton.

The porphyrin ligand (PorTPPO) was synthesized from 4-bromobenzaldehyde via five step reactions. The novel Yb(III) porphyrin coordination polymer $[(Yb(hfa)_3)_2(PorTPPO)]_n$ was synthesized by mixing PorTPPO ligand and Yb(hfa)_3·2H₂O in CH₂Cl₂ and MeOH solution (vol/vol = 1/1) (Figure 1). The resulting Yb(III) coordination polymer was characterized by ESI-MS, XRD and IR measurements. Prepared $[(Yb(hfa)_3)_2(PorTPPO)]_n$ showed absorption bands of hfa and PorTPPO Soret band at 312 and 415 nm, respectively. The NIR emission of $[(Yb(hfa)_3)_2(PorTPPO)]_n$ excited at 415 nm (Soret band) was observed at around 1000 nm. These results indicate that the porphyrin units show effective photosensitization for NIR emission in a Yb(III) coordination polymer.

V. Bulach, F. Sguerra, M.W Hosseini, *Coord. Chem. Rev.* 2012, 256, 1468. 2) S. Kishimoto, T. Nakagawa, T. Kawai, Y. Hasegawa, *Bull. Chem. Soc. Jpn.* 2011, 84, 148. 3) S. Omagari, T. Nakanishi, Y. Hirai, Y. Kitagawa, T. Seki, K. Fushimi, H. Ito, Y. Hasegawa, *Eur. J. Inorg. Chem.* 2018, 561.