

## Influence on Fluorescence Properties of Substituents on Benzene Ring of Diimidazo- and Dipyrrolo[1,2-*a*:2',1'-*c*]quinoxalines

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**Keywords:** Fluorescence; Diimidazoquinoxaline; Dipyrroloquinoxaline; Substituent Effect; CT Transition

We have reported the fluorescence of diimidazo- and dipyrrolo[1,2-*a*:2',1'-*c*]quinoxalines (**1a** and **2a**) in UV and blue region.<sup>1)</sup> To achieve longer fluorescence wavelength using CT transition, we examined the introduction of substituents on their benzene rings.

When fluorine substituent was introduced at 6 position of **1a**, longer fluorescence peaks of **1b** (385 and 396 nm) were obtained compared with **1a** (367 nm) in THF and CH<sub>3</sub>CN (Table 1) accompanied with hypsochromic shift of absorption peaks (Table 1). We also found solvent effect in the fluorescence of **1b** although there was no change in **1a**. It would be caused by the CT transition in the excited state of **1b**.

In the case of dipyrrolo[1,2-*a*:2',1'-*c*]quinoxalines, similar bathochromic shift of the fluorescence peaks was observed in **2b** against **2a**. Both compounds showed the solvent effect on fluorescence peaks. It would be reason for the electron-rich character of bipyrrole moiety. To compare with **2b**, we investigated methoxy substituted **2c**. Little change of fluorescence peaks was observed with no solvent effect, which means  $\pi$ - $\pi^*$  transition in the excited state of **2c**. Thus, we revealed that the longer fluorescence was achieved by introducing the electron-withdrawing substituent on benzene rings of **1a** and **2a**.

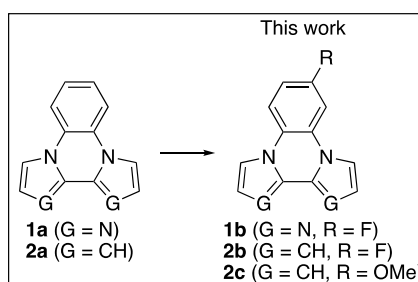


Table 1. Optical Properties of **1** and **2**

Compound	Solvent	$\lambda_{\text{abs}}$ (nm) [ $\epsilon$ (M <sup>-1</sup> cm <sup>-1</sup> )] <sup>a)</sup>	$\lambda_{\text{em}}$ (nm) <sup>b)</sup> [ $\Phi_F$ ] <sup>c)</sup>
<b>1a</b>	CH <sub>3</sub> CN	307.5 [12,200]	367 [0.70]
	THF	309 [9,400]	367 [0.72]
<b>1b</b>	CH <sub>3</sub> CN	247(sh) [11,900]	396 [0.33]
	THF	249(sh) [12,800]	385 [0.44]
<b>2a</b>	CH <sub>3</sub> CN	320 [10,700]	434 [0.17]
	THF	321 [10,600]	416 [0.43]
<b>2b</b>	CH <sub>3</sub> CN	322 [15,600]	453 [0.12]
	THF	322 [11,200]	434 [0.22]
<b>2c</b>	CH <sub>3</sub> CN	369 [4,900]	434 [0.02]
	THF	367 [4,800]	435 [0.01]

a) Concentration:  $3.0 \times 10^{-5}$  M. b) Concentration:  $3.0 \times 10^{-7}$  M.

c) Determined using *p*-terphenyl ( $\Phi_F=0.87$ , 265 nm) as a standard.

1) Matsumoto, S.; Sakamoto, K.; Akazome, M. *Heterocycles* **2015**, *91*, 795.