

## ROS/RNS generation by various discharge plasma

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### 1. Introduction

In recent years, plasma treated water has gained increasing attention, because reactive oxygen species (ROS) and reactive nitrogen species (RNS) in the plasma treated water play important roles in various fields such as plasma medicine<sup>(1)</sup> and agriculture.<sup>(2)</sup> These ROS/RNS are produced by dissolving species in plasma. To use the plasma treated water effectively and efficiently, it is important to control the ROS/RNS concentration.

In this work, we generated an Ar plasma jet, a positive DC corona discharge, a positive pulsed discharge and a packed-bed dielectric barrier discharge (PB-DBD), produced ROS/RNS in deionised water using these discharges, and then investigated the concentration and generation efficiency of the ROS/RNS in the water.

### 2. Experimental apparatus and conditions

Experimental apparatuses are similar to those used in previous works.<sup>(3-6)</sup> The Ar plasma jet is generated using a dielectric barrier discharge, and N<sub>2</sub>-O<sub>2</sub> mixture gas is mixed into the plasma jet, and then deionised water is exposed to the plasma jet. The corona discharge and the pulsed discharge are generated above a water surface in N<sub>2</sub>-O<sub>2</sub> mixture gas as a background gas. The PB-DBD is generated in N<sub>2</sub>-O<sub>2</sub> mixture gas, and the off-gas of the PB-DBD reactor packed with soda-lime glass balls is sparged into deionised water. Samples of 1.2 mL from the water are taken.

The concentrations of H<sub>2</sub>O<sub>2</sub>, NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup> in the sampled water are measured using High Performance Liquid Chromatograph (Shimadzu, Prominence, column: IC NI-424) in combination with an auto sampler. An eluent is the aqueous solution of acetic acid (3 mmol/L) and potassium hydroxide (1.9 mmol/L), and the wavelength of an absorbance detector is fixed at 220 nm.

### 3. Results and discussion

H<sub>2</sub>O<sub>2</sub> is detected in the sampled water after the plasma exposure except in the case of PB-DBD, and these concentrations increase with input energies. H<sub>2</sub>O<sub>2</sub> is mainly formed by the combination reaction of two OH radicals,<sup>(7)</sup>

which are produced by the dissociation of H<sub>2</sub>O. Although O<sub>3</sub> is detected in the off-gas from the PB-DBD reactor, no H<sub>2</sub>O<sub>2</sub> is detected in the sampled water; therefore, O<sub>3</sub> does not contribute to H<sub>2</sub>O<sub>2</sub> production. Consequently, the dissociation of H<sub>2</sub>O by high-energy electron and excited molecules (atoms), such as metastable N<sub>2</sub> and Ar, may be dominant to OH production.

NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup> are detected in the sampled water except for NO<sub>2</sub><sup>-</sup> in the case of PB-DBD, so that it is necessary to generate plasma in the vicinity of water for NO<sub>2</sub><sup>-</sup> generation. NO<sub>2</sub><sup>-</sup> concentrations show tendencies to be constant or decrease with the increase of input energy, but NO<sub>3</sub><sup>-</sup> concentrations are roughly relative to the input energy.

Table 1 shows the generation efficiency of H<sub>2</sub>O<sub>2</sub>, NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup> in each discharge. The pulsed discharge has higher generation efficiency of H<sub>2</sub>O<sub>2</sub> and NO<sub>2</sub><sup>-</sup> than others, and PB-DBD off-gas sparging can only and efficiently generate NO<sub>3</sub><sup>-</sup>.

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Table 1. H<sub>2</sub>O<sub>2</sub>, NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup> generation efficiency

discharge	generation efficiency [ $\mu\text{g}/\text{Wh}$ ]		
	H <sub>2</sub> O <sub>2</sub>	NO <sub>2</sub> <sup>-</sup>	NO <sub>3</sub> <sup>-</sup>
plasma jet	10	20 ~ 50	60 ~ 90
corona	200	50 ~ 200	800 ~ 1400
pulsed	700 ~ 1500	< 350	650 ~ 1100
PB-DBD	—	—	800 ~ 2500