

Effect of annealing conditions on photocatalytic activity of TiO₂ nanorods prepared by hydrothermal method

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Owing to the excellent chemical and photochemical stability, TiO₂ has always been a promising material for numerous applications, such as water splitting. For better application, the morphology and crystal structure of TiO₂ need to be well controlled[1, 2]. The electron transport along the long axis of the 1D crystalline nanostructure such as nanorod is expected to be several orders of magnitude faster than that in the random network of nanoparticles as 1D nanostructure can provide direct transport pathways for charge carriers. Among several preparation methods for TiO₂ nanorods, the hydrothermal method is commonly used for synthesis as it offers low temperature synthesis, the flexibility to attain different particle sizes and morphologies. In the present work, a seed layer of TiO₂ thin film is prepared on the substrate to avoid big lattice mismatch between anatase TiO₂ and substrate and further TiO₂ nanorods were hydrothermally grown on the prepared anatase TiO₂ thin film. The goal of the present work is to understand the effect of post treatment such as annealing time, temperature as well as atmosphere on the photocatalytic activity prepared TiO₂ nanorods.

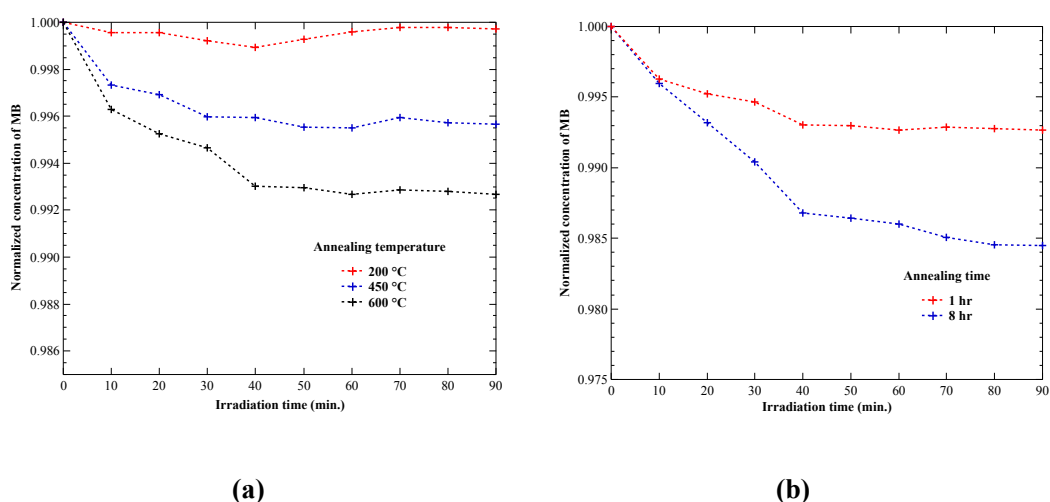


Fig. 1 : Photocatalytic degradation of methylene blue under UV light irradiation (a) different annealing temperatures for 1hr respectively (b) different annealing time at 600 °C

References:

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